

ISOLATION OF SECONDARY METABOLITES FROM *Teclea nobilis* AND *Rapanea melanophloeos* ACTIVE AGAINST *Schistosoma mansoni*

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award of Master of Science Degree in Chemistry of Egerton University**

EGERTON UNIVERSITY

JUNE, 2014

DECLARATION AND RECOMMENDATION

DECLARATION

I, Njogu Mark Kimani, declare that this research thesis is my original work and has not been submitted wholly or in part for any award in any institution.

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SM11/3130/11

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RECOMMENDATION

We wish to confirm that this research thesis has been prepared under our supervision and is presented for examination as per the Egerton University regulations with our approval.

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DEDICATION

To

My dear dad; Isaiah Wang'endo, who always supports and believes in me. You are my inspiration and source of strength.

And

My lovely mom; Dorcas Njambi, whose love and prayers are immeasurable.

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ABSTRACT

Helminthic infections have severe consequences on the health of millions of people worldwide and cause serious economic losses. Synthetic drugs have been effective wormicides, however, they are expensive, show side effects and develop resistance. This has generated interest in the use of plant based anti-helmintics, which seem to offer a reliable, cheap and cost effective method. In this research, the anti-helmintic activity of secondary metabolites from *Teclea nobilis* and *Rapanea melanophloeos*, which are used ethnomedically as dewormers, against *Schistosoma mansoni* are reported. The *Teclea nobilis* essential oil was extracted using hydro-distillation in a modified Clevenger apparatus. The essential oil was analyzed using gas chromatography-mass spectrometry and determined to be majorly constituted of monoterpenes and sesquiterpenes which had concentrations of 42.21 and 33.09%, respectively. The major monoterpenes were β -Ocimene (10.15%) and γ -Terpinene (6.11%) while major sesquiterpenes were β -Cadinene (4.98%) and 1,6-Germacradien-5-ol (4.38%). The essential oil showed lethal effects against *Schistosoma mansoni* miracidia with LC₅₀ and LC₉₀ values of 196.29 and 367.24 ppm, respectively. The non-volatiles of both *Teclea nobilis* and *Rapanea melanophloeos* were also active against *Schistosoma mansoni* miracidia. Fractionation of *Teclea nobilis* and *Rapanea melanophloeos* leaf ethyl acetate extracts over silica gel column chromatography yielded six compounds which were successfully identified through analysis of their 1D and 2D nuclear magnetic resonance spectroscopy and mass spectrometry data as well as comparison with literature data. Out of the six compounds, *Teclea nobilis* had four furoquinoline alkaloids; Tecleoxine **10**, Methylnkolbisine **13**, Kokusagine **21** and Nkolbisine **22** while the *Rapanea melanophloeos* had two benzoic acid derivatives; Myrsinoic acid B **23** and Myrsinoic acid C **24**. All these compounds were active against *S. mansoni* miracidia. Compound **23** which recorded LC₅₀ and LC₉₀ mortality values of 98.06 and 236.51 ppm, respectively was the most potent followed by compound **24** which registered LC₅₀ and LC₉₀ values of 139.89 and 314.23 ppm, respectively. The compounds **10**, **13** and **21**, which due to their small amounts could not be separated thus tested as a mixture, registered LC₅₀ and LC₉₀ values of 270.18 and 690.93 ppm, respectively and compound **22** which recorded LC₅₀ and LC₉₀ values of 287.97 and 631.73 ppm, respectively was the least potent. These findings show that these compounds can be used as lead compounds in the development of new, biodegradable, environmentally friendly and more potent miracidicides or anti-helmintics.

TABLE OF CONTENTS

DECLARATION AND RECOMMENDATION	ii
DEDICATION	iii
COPYRIGHT	iv
ACKNOWLEDGEMENT	v
ABSTRACT	vi
LIST OF FIGURES	x
LIST OF TABLES	xi
LIST OF APPENDICES	xii
LIST OF ABBREVIATIONS AND ACRONYMS	xiv
CHAPTER ONE	1
INTRODUCTION	1
1.1 Background information	1
1.2 Statement of the problem	2
1.3 Objectives.....	2
1.4 General objective.....	2
1.5 Specific objectives.....	2
1.6 Hypotheses	2
1.7 Justification	3
CHAPTER TWO	4
LITERATURE REVIEW	4
2.1 Helminthosis	4
2.2 <i>Schistosoma mansoni</i>	5
2.3 Anti-helminthics.....	5
2.4 <i>Teclea nobilis</i>	7

2.5	<i>Rapanea melanophloeos</i>	9
CHAPTER THREE		11
MATERIALS AND METHODS		11
3.1	Collection of plant material.....	11
3.2	Extraction and analysis of essential oil	11
3.2.1	Distillation of essential oil.....	11
3.2.2	Essential oil chemical composition determination and spectroscopic analysis.....	11
3.3	Extraction, isolation and structure elucidation of non-volatiles.....	13
3.3.1	Extraction of phytochemicals	14
3.3.2	Solvent partitioning of crude methanol extract	14
3.3.3	Thin layer chromatography (TLC)	14
3.3.4	Column chromatography	15
3.3.5	Purification of compounds.....	15
3.3.6	Nuclear magnetic resonance (NMR) spectroscopy	16
3.3.7	Two dimensional NMR spectroscopy	16
3.3.8	Mass spectrometry.....	17
3.4	<i>In vitro</i> anti-helminthic activity test.....	17
3.4.1	Collection and preparation of <i>S. mansoni</i> miracidia.....	17
CHAPTER FOUR		19
RESULTS AND DISCUSSION		19
4.1	Structure elucidation of isolated compounds.....	19
4.1.1	Structure elucidation of compounds 10, 13 and 21.....	19
4.1.2	Structure elucidation of compound 13.....	22
4.1.3	Structure elucidation of compound 21	25
4.1.4	Structure elucidation of compound 22.....	26

4.1.5	Structure elucidation of compound 23	29
4.1.6	Structure elucidation of compound 24	34
4.2	Miracidicidal assay of the methanol crude extract of <i>T. nobilis</i>	38
4.3	Miracidicidal assay of <i>T. nobilis</i> ethyl acetate and hexane extracts	39
4.4	Miracidicidal assay of fractions from ethyl acetate crude extract	41
4.5	Bio-assay of purified compounds isolated from <i>T. nobilis</i>	45
4.6	Miracidicidal assay of <i>R. melanophloeos</i> methanol, ethyl acetate and hexane crude extracts	48
4.7	Bio-assay of fractions from ethyl acetate extract.....	52
4.8	Bio-assay of purified compounds isolated from <i>R. melanophloeos</i>	55
4.9	Anti-miracidia assay of <i>T. nobilis</i> essential oil.....	58
4.10	Determination of the chemical composition of the oil extract.....	59
CHAPTER FIVE		67
CONCLUSIONS AND RECOMMENDATIONS		67
5.1	Conclusion	67
5.2	Recommendations.....	68
REFERENCES		69
APPENDICES		79

LIST OF FIGURES

Figure 1: Phytochemicals with anti-helmintic activity	7
Figure 2: Alkaloids and sesquiterpene ketones isolated from <i>T. nobilis</i>	8
Figure 3: Photographs of <i>T. nobilis</i>	8
Figure 4: Photographs of <i>R. melanophloeos</i>	9
Figure 5: Some of the compounds isolated from <i>R. melanophloeos</i>	10
Figure 6: A flow chart showing the summary of isolation and analysis of essential oil	12
Figure 7: A summary of isolation and structure elucidation of non-volatiles	13
Figure 8: Structure of compound 10 showing COSY and HMBC	21
Figure 9: Mass spectrum of compound 10.....	22
Figure 10: Structure of compound 13 showing COSY and HMBC	24
Figure 11: Mass spectrum of compound 13.....	24
Figure 12: Structure of compound 21 showing COSY and HMBC	26
Figure 13: Mass spectrum of compound 21.....	26
Figure 14: Structure of compound 22 showing COSY and HMBC	28
Figure 15: Mass spectrum of Compound 22.....	29
Figure 16: Structure of compound 23 showing COSY and HMBC	32
Figure 17: M+H mass spectrum of compound 23	33
Figure 18: M+Na mass spectrum of compound 23.....	33
Figure 19: M+K mass spectrum of compound 23	34
Figure 20: Structure of compound 24 showing COSY and HMBC	36
Figure 21: M+K mass spectrum of compound 24	37
Figure 22: M+H mass spectrum of compound 23	37
Figure 23: Structures of praziquantel and triclabendazole	49
Figure 24: Mass spectra of compound 25	61
Figure 25: Mass spectra of compound 28	62
Figure 26: Mass spectra of compound 29	63
Figure 27: Mass spectra of compound 30	64
Figure 28: Scheme showing fragmentation pattern of compound 30	64
Figure 29: Mass spectra of compound 32	65
Figure 30: Some major compounds of <i>T. nobilis</i> essential oil.....	66

LIST OF TABLES

Table 1: NMR data of compound 10	20
Table 2: NMR data of compound 13	23
Table 3: NMR data of compound 21	25
Table 4: NMR data of compound 22	27
Table 5: NMR data of compound 23	31
Table 6: NMR data of compound 24	35
Table 7: Miracidicidal activity of <i>T. nobilis</i> methanol crude extract.....	39
Table 8: Miracidicidal activity of <i>T. nobilis</i> ethyl acetate crude extract	40
Table 9: Miracidicidal activity of <i>T. nobilis</i> hexane crude extract	41
Table 10: Miracidicidal activity of <i>T. nobilis</i> ethyl acetate F ₁	42
Table 11: Miracidicidal activity of <i>T. nobilis</i> ethyl acetate F ₃	43
Table 12: Miracidicidal activity of <i>T. nobilis</i> ethyl acetate F ₄	44
Table 13: Miracidicidal activity of <i>T. nobilis</i> ethyl acetate methanol fraction.....	45
Table 14: Miracidicidal activity of compounds 10, 13 and 21 mixtures	46
Table 15: Miracidicidal activity of compound 22.....	47
Table 16: Miracidicidal activity of <i>R. melanophloeos</i> methanol crude extract	49
Table 17: Miracidicidal activity of <i>R. melanophloeos</i> ethyl acetate crude extract.....	50
Table 18: Miracidicidal activity of <i>R. melanophloeos</i> hexane crude extract.....	51
Table 19: Miracidicidal activity of <i>R. melanophloeos</i> ethyl acetate F ₁	52
Table 20: Miracidicidal activity of <i>R. melanophloeos</i> ethyl acetate F ₃	53
Table 21: Miracidicidal activity of <i>R. melanophloeos</i> ethyl acetate F ₄	54
Table 22: Miracidicidal activity of <i>R. melanophloeos</i> ethyl acetate methanol fraction	55
Table 23: Miracidicidal activity of compound 23.....	56
Table 24: Miracidicidal activity of compound 24.....	57
Table 25: Miracidicidal activity of <i>T. nobilis</i> essential oil	59
Table 26: Major constituents of <i>T. nobilis</i> essential oil	60

LIST OF APPENDICES

Appendix 1: Generated LC values for <i>Teclea nobilis</i> methanol crude extract	79
Appendix 2: Generated LC values for <i>Teclea nobilis</i> hexane extract	80
Appendix 3: Generated LC values for <i>Teclea nobilis</i> ethyl acetate extract.....	81
Appendix 4: Generated LC values for <i>Teclea nobilis</i> fraction F ₁	82
Appendix 5: Generated LC values for <i>Teclea nobilis</i> fraction F ₃	83
Appendix 6: Generated LC values for <i>Teclea nobilis</i> fraction F ₄	84
Appendix 7: Generated LC values for <i>Teclea nobilis</i> methanol fraction	85
Appendix 8: Generated LC values for <i>Teclea nobilis</i> compounds 10, 13 and 21	86
Appendix 9: Generated LC values for <i>Teclea nobilis</i> compound 22.....	87
Appendix 10: Generated LC values for <i>Teclea nobilis</i> Essential oil	88
Appendix 11: Generated LC values for <i>Rapanea melanophloeos</i> methanol crude extract	89
Appendix 12: Generated LC values for <i>Rapanea melanophloeos</i> Ethyl acetate extract	90
Appendix 13: Generated LC values for <i>Rapanea melanophloeos</i> hexane extract.....	91
Appendix 14: Generated LC values for <i>Rapanea melanophloeos</i> fraction F ₁	92
Appendix 15: Generated LC values for <i>Rapanea melanophloeos</i> fraction F ₃	93
Appendix 16: Generated LC values for <i>Rapanea melanophloeos</i> fraction F ₄	94
Appendix 17: Generated LC values for <i>Rapanea melanophloeos</i> methanol fraction	95
Appendix 18: Generated LC values for <i>Rapanea melanophloeos</i> compound 23	96
Appendix 19: Generated LC values for <i>Rapanea melanophloeos</i> compound 24.....	97
Appendix 20: GC/MS spectrum of <i>Teclea nobilis</i> essential oil.....	98
Appendix 21: Constituents of <i>Teclea nobilis</i> Essential Oil	99
Appendix 22: ¹ H NMR spectrum of compound 23	101
Appendix 23: APT spectrum of compound 23	102
Appendix 24: HSQC spectrum for the compound 23	103
Appendix 25: ¹ H- ¹ H COSY spectrum for the compound 23.....	104
Appendix 26: HMBC spectrum for the compound 23.....	105
Appendix 27: ¹ H NMR spectrum of compound 24	106
Appendix 28: APT NMR spectrum of compound 24	107
Appendix 29: HSQC spectrum of compound 24	108
Appendix 30: ¹ H- ¹ H COSY NMR spectrum of compound 24	109

Appendix 31: HMBC NMR spectrum of compound 24.....	110
Appendix 32: ^1H NMR spectrum of compounds 10, 13 and 21	111
Appendix 33: APT NMR spectrum of compounds 10, 13 and 21	112
Appendix 34: HSQC NMR spectrum of compounds 10, 13 and 21.....	113
Appendix 35: ^1H - ^1H COSY NMR spectrum of compounds 10, 13 and 21	114
Appendix 36: HMBC NMR spectrum of compounds 10, 13 and 21.....	115
Appendix 37: ^1H NMR spectrum of compound 22	116
Appendix 38: APT NMR spectrum of compound 22	117
Appendix 39: HSQC NMR spectrum of compound 22	118
Appendix 40: ^1H - ^1H COSY NMR spectrum of compound 22	119
Appendix 41: HMBC NMR spectrum of compound 22.....	120

LIST OF ABBREVIATIONS AND ACRONYMS

APT	Attached proton test
COSY	Correlation spectroscopy
DALYs	Disability-adjusted life years
DMSO	Dimethyl sulfoxide
GC/MS	Gas chromatography/ Mass spectrometry
HMBC	Heteronuclear multiple bond correlation
HPLC	High performance liquid chromatography
HSQC	Heteronuclear single quantum correlation
KEMRI	Kenya medical research institute
LC	Lethal concentration
LC ₅₀	Lethal concentration killing 50% of parasite
LC ₉₀	Lethal concentration killing 90% of parasite
NMR	Nuclear magnetic resonance
PTLC	Preparative thin layer chromatography
R _f	Retention factor
SPSS	Statistical package for social science
STH	Soil transmitted helminth
TB	Tuberculosis
TEEAL	The essential electronic agricultural library
TLC	Thin layer chromatography

CHAPTER ONE

INTRODUCTION

1.1 Background information

Helminths belong to two major groups: platyhelminthes and nematodes (Ijagbone and Olagunju, 2006). They include roundworms, hook worms and whip worms which cause high morbidity and mortality in humans. These parasites are common in sub-Saharan Africa due to poor sanitation, poverty, malnutrition, ignorance, lack of protective clothing, lack of clean water as well as limited access to preventive measures and health care (Zvi *et al.*, 2010). These factors also make helminths co-infections in human such as with *Plasmodium falciparum*, HIV/AIDS and tuberculosis prevalent. Helminthic morbidity is also common with obesity, cardiovascular diseases, allergy and diabetes (Perry and Randolph, 1999; Zvi *et al.*, 2010).

Gastrointestinal parasites have also been reported to be a major menace facing animals and birds. These parasites are known to reduce the animals' survival, growth rate and reproductive performance thus affecting their overall production. A study conducted in Kenya, showed that helminthiasis accounted for the second most important disease accounting for 21% of the deaths by parasitic conditions, between 1989 and 1998 in chicken (Permin *et al.*, 1999). It has been estimated that more than 750 million chickens, guinea fowls and ducklings in Africa die each year as a result of various infections, helminthiasis being a major contributor (Matur *et al.*, 2010). Therefore, the economic and social impact of helminths cannot be underestimated. In humans and livestock their effects range from stunted growth, reduced weight gain, diarrhoea, anaemia, respiratory problems, reduced productivity and death especially in the tropics and developing countries (Jozef *et al.*, 2011; Velkers *et al.*, 2011).

The use of synthetic drugs has been effective in controlling the helminths. However, development of anti-helminthic resistance, the problem of drug residues, deficient livestock extension services and the high cost of conventional anti-helminthics, has led to the evaluation of medicinal plants as an alternative source of anti-helminthics for treatment of human and livestock against helminthes (Jozef *et al.*, 2011).

1.2 Statement of the problem

Helminth infections cause adverse effects on human and livestock such as haematological and biochemical disturbances, loss of body weight, death leading to economic losses. The use of synthetic anti-helminthics has for a long time been considered the only effective way of controlling these parasitic infections in humans and livestock. However, these drugs are very expensive, unaffordable and at times unavailable to poor people in rural areas particularly in sub-Saharan Africa. In addition, synthetic drugs have developed some serious problems such as drug resistance, ecological pollution and residues in food. Additionally, there is a general stagnation in the development of new safe, efficacious and reliable conventional drugs, which has led to an increased need for the treatment and control of helminth infections. Thus people have resorted to the continued use of plant based de-wormers. The use of ethno-medicine or ethno-botanical medicine and ethno-veterinary medicine seem as feasible alternative anti-helminthics and is therefore gaining a lot of importance.

1.3 Objectives

1.4 General objective

The general objective of this study was to isolate and characterize secondary metabolites from *T. nobilis* and *R. melanophloeos* and determine their anti-helminthic activity, against *Schistosoma mansoni* *in vitro*.

1.5 Specific objectives

- i. To isolate and determine the chemical composition of essential oil from *T. nobilis*.
- ii. To evaluate anti-helminthic activity of *T. nobilis* essential oil against *S. mansoni*.
- iii. To determine anti-helminthic activity of non-volatiles from *T. nobilis* and *R. melanophloeos* against *S. mansoni*.
- iv. To isolate and elucidate the structure(s) of *T. nobilis* and *R. melanophloeos* anti-helminthic compounds.

1.6 Hypotheses

- i. The chemical composition of essential oil from *T. nobilis* cannot be determined.
- ii. Essential oil of *T. nobilis* has no anti-helminthic activity.

- iii. Non-volatile extracts of *T. nobilis* and *R. melanophloeos* have no anti-helminthic activity.
- iv. The structure of the active anti-helminthic compounds cannot be elucidated.

1.7 Justification

Plant anti-helminthics for a long time have lacked scientific evaluation and medical awareness. Studies on such plants need to go beyond mere anthropological curiosity. This is because they are readily applicable elements of ethno-medicine in human and livestock development. Thus studying these herbal medicines can serve to validate and enhance existing local uses and can give clues to remedies with further potential. Plant metabolites also offer cheaper, more sustainable, available, efficacious and reliable alternatives to synthetic drugs. They also offer treatment methods that are more environmentally friendly, since they are less toxic, produce fewer side effects, are more biodegradable, accumulate no drug residues in meat or faeces and do not trigger anti-helminthic chemo-resistance. *T. nobilis* and *R. melanophloeos* have been ethno-medicinally used as anti-helminthics, as analgesics and for respiratory problems among other uses. In this study, anti-helminthic properties of their secondary metabolites against *S. mansoni* were evaluated.

CHAPTER TWO

LITERATURE REVIEW

2.1 Helminthosis

Helminthosis is of both economic and public health importance as well as being one of the world's most important neglected tropical diseases (Maduiké *et al.*, 2012). According to recent reports by WHO, helminthosis burden is equivalent to 50% of that of malaria and 25% of that of HIV/AIDS with approximately 2.9 billion people being infected with nematode (WHO, 2010). It is estimated that, about 90% of the 200 million people with Schistosomiasis live in Africa. Similarly, Africa bears 198 million people with Hookworm, 173 million with *Ascaris lumbricoides* and 162 million with *Trichuris trichiura* (Zvi *et al.*, 2010, WHO; 2010). Hookworm morbidity associated with anaemia, likely contributing to maternal mortality, among pregnant women currently stands at 44 million. Also 12% of the total disease burden among children aged 5-14 years is contributed by intestinal worms (Jozef *et al.*, 2011).

Soil transmitted helminth (STH) infections have been estimated to cause 135,000 deaths annually, mainly due to anaemia, caused by hookworm and whipworm infections, intestinal, or biliary obstruction and chronic dysentery caused by roundworms and whipworms (Jozef *et al.*, 2011). STH infections are as important as malaria and TB in terms of disability-adjusted life years (DALYs) lost, that is the number of healthy years lost to premature death or disability, (Stephenson *et al.*, 2000). It is estimated that 22.1 million DALYs are lost due to hook worm, 10.5 million due to *Ascaris lumbricoides*, 6.4 million due *Trichuris trichiura*, and 37 million for the three combined while that of malaria is approximately 35.7 million globally (Jozef *et al.*, 2011).

Helminthic infections, HIV, TB and malaria have also been reported to have extensive overlap in sub-Saharan Africa (Jozef *et al.*, 2011) with both helminths and HIV-1 estimated to be over 22 million (Judd *et al.*, 2010). De-worming individuals with HIV-1 has been suggested to delay HIV-1 disease progression by up to 25% and delay AIDS development by upto 3.5 years (Judd *et al.*, 2010). According to Midzi *et al.* (2011) 45 million (25%) school-aged children are also at risk of Hookworm and malaria infection.

2.2 *Schistosoma mansoni*

S. mansoni is a blood dwelling digenetic trematode which causes intestinal Schistosomiasis or bilharziasis. It is ranked as the second most important parasitic infection after malaria in terms of morbidity and mortality. In 2011, at least 243 million people were reported to require treatment. In sub-Saharan Africa, an estimated mortality of 280,000 people is reported annually (Boissier *et al.*, 2009).

S. mansoni has a complex life cycle. Briefly, parasitic eggs excreted in faeces from infected persons on reaching fresh water hatch into larvae called miracidiae which then seek out *Biomphalaria alexandriana* snails. Upon infection of snail, the miracidiae are transformed into mother sporocytes which then develop into daughter sporocytes. These in turn develop into mature cercariae. The cercariae are released into the water, where they infect human by penetration of the skin. The head of the cercariae transform, in the skin, into an endoparasitic larva called schistosomulae which enters the vascular system. The schistosomulae migrates to the lung, develop further and move to the liver. In the liver, they feed on the red blood cells and mature worms pair up. The worm pairs travel to the mesenteric veins, where they produce eggs. The eggs pass through the walls of the blood vessels through the intestinal wall to be passed out in faeces (Mohamed *et al.*, 2005; Frelick, 2012).

2.3 Anti-helminthics

An anti-helminthic or de-wormer is a substance that expels or destroys gastrointestinal worms. Anti-helminthics eliminates worms by either paralyzing or starving them to death. Worms must eat almost continuously to meet their metabolic needs since they do not have means to store energy. Thus any disruption in the feeding for 24 hours or less process results in energy depletion and eventual death. Worm paralysis and temporally lose of the ability to maintain posture in the gut also leads to death (Shaziya and Goyal, 2012). Chemotherapy is mostly the method of choices in helminthes control. However, the world wide long-term application of the drugs coupled with the recent development of anti-helminthic tolerant strains (Shaziya and Goyal, 2012; Boissier *et al.*, 2009), in parasites of high economic significance generates concern over the development of new drugs.

Modern synthetic anti-helminthics are majory grouped into four classes:

- I. Benzimidazoles such as albendazole and mebendazole. They are broad spectrum drugs that bind to free β -tubulin, inhibiting its polymerization and so interfering with microtubule-dependent glucose uptake by the parasite.
- II. Imidazothiazoles and tetrahydropyrimidines such as levamisole and pyrantel which stimulate the nicotinic acetylcholine receptors, resulting in over-stimulation, blockage of the neuromuscular junctions, and rigid paralysis of the worms. The parasites are then unable to move in the intestinal tract and are swept out by the peristaltic action in the intestine.
- III. Macrocyclic lactones such as Ivermectin which open glutamate-gated chloride channels, increasing chloride ion conductance, leading to defects in neurotransmission and flaccid paralysis.
- IV. Heterocyclic ethyleneamines, of which the best known member, piperazine, is only used against *A. lumbricoides* and *E. vermicularis*. This drug acts by reversibly inhibiting neuromuscular transmission by stimulating gamma-aminobutyric acid (GABA) receptors in nematode muscle, causing flaccid paralysis of the worms, which are then removed by normal intestinal peristalsis (Stepek *et al.*, 2004; Stepek *et al.*, 2006).

In the developing countries, ethno-veterinary medicines have been used for a long time as anti-helmintic (Singh and Shalini, 2000; Lalchandama, 2011; Nalule *et al.*, 2011; Velkers *et al.*, 2011; Shaziya and Goyal, 2012). Plants containing proteolytic enzymes of the cysteine catalytic class such as papaya, which contain enzymes such as papain, chymopapain and lysozymes in the latex as well as in leaves, have mostly been used (Dakpogan, 2005). Moreover, various plants have been studied and reported to have anti-helmintic activity. Some of these plants include *Einocostemma littorale* whose extracts have been shown to cause paralysis of *Pheretima pothera* worms (Vidyadhar *et al.*, 2010). The plants *Annona senegalensis*, *Anogeissus leiocarpus*, *Lippia rogersii*, *Strerospermum kontihianum*, *Vernonia tonoreana* (Mali and Mehta 2008) and *Corriandum sativum* (Egualle *et al.*, 2006) have been shown to have effects against *Heamonchus contortus*. Furthermore, *Ocimum sanctum*, *Piliostigma thoningii*, *Moghania vestite*, *Mimusops elergi*, *Punica granatum*, *Alotropis procera*, *Capparis deciduas*, *Butea monosperma*, *Aloe vera*, *Neolamarckia cadamba*, *Xylopiya aethiopica*, *Cynadropsis gynandra*, *Evolvus alsinoides*, *Carica papaya*, *Piper longum*, *Nigella sativa*, *Ficus insipida*, *Nicotiana tabacum*, *Cannabis sativa*,

Cleome icosandra, *Trifolium repers*, *Strobilanthes discolor* among others have been shown to have anti-helminthic activity (Munglo *et al.*, 2006). Several phytochemicals such as nefuridin, linalool, chavicol, kaempferol among others (**1-9**) have been isolated and also reported to have anti-helminthic properties. Some of these compounds are shown in figure 1 below (Chitwood, 2002; Lasisi and Kareem, 2011).

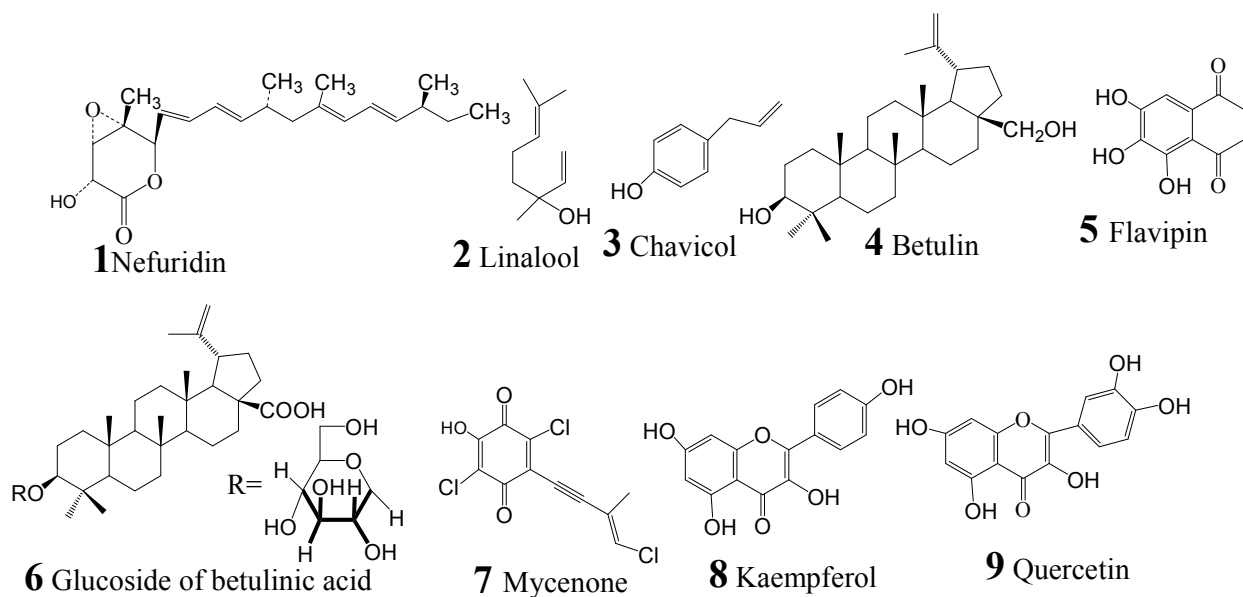


Figure 1: Phytochemicals with anti-helminthic activity

2.4 *Teclea nobilis*

T. nobilis is an ever green shrub or tree of the Rutaceae family with a smooth grey bark native to Ethiopia, Kenya, Tanzania, and Uganda rain forests. It has simple leaves on glabrous branchlets. A photograph of the plant is shown in figure 3 below. It is reported to have been used ethno-medicinally in many African countries and in Saudi Arabia (Adnan *et al.*, 2001a).

The leaves and stem bark has been used as a remedy for gonorrhoea. It has been reported to have anti-inflammatory, analgesic and antipyretic effect (Adnan *et al.*, 2001a). Quinoline and furoquinoline alkaloids, limonoids, triterpenes, lignin and flavonoid have been reported in Rutaceae family in which this plant belongs. In a previous study the presence of two isomeric axane and oppositane sesquiterpene ketones (**15-16**), teclenone A [1 α -(1-Oxo-2-methylpropyl)-3 $\alpha\alpha$ -methyl-7-methyleneoctahydroinden-4 α -ol] and teclenone B [1 α -(1-Oxo-2-methylpropyl)-

3 β -methyl-7-methyleneoctahydroinden-4 β -ol] and alkaloids (**10-14**) from aerial parts (Figure 4) of *T. nobilis* were reported (Adnan *et al.*, 2002; Adnan *et al.*, 2003).

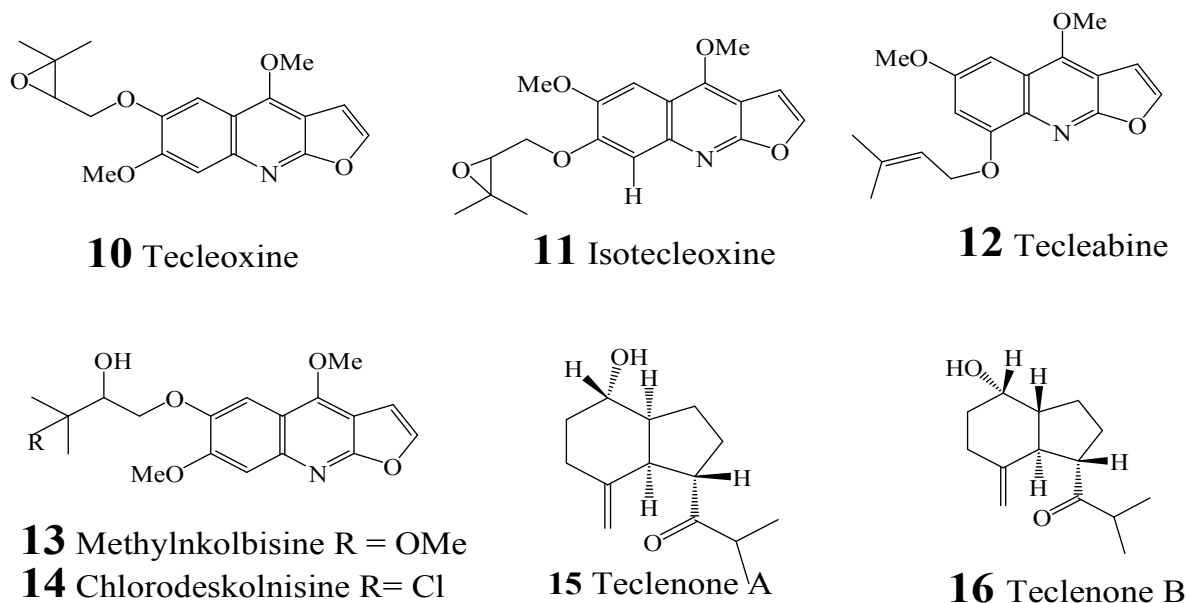


Figure 2: Alkaloids and sesquiterpene ketones isolated from *T. nobilis*.

The essential oil from *T. nobilis* are also said to contain Germacrene-d, Ocimene isomer, Gualol, Elemol and Bulnesol as the main components. The oil is also reported to show antipyretic activity and sedative effects (Adnan, 2001b).



Figure 3: Photographs of *T. nobilis*

2.5 *Rapanea melanophloeos*

R. melanophloeos is an evergreen tree of the Myrsinaceae family widespread in Kenyan upland forests (Mosa *et al.*, 2010). When mature, leaves are simple oblong-lanceolate, dark green, paler below, leathery and dull with reddish leaf stalks and are clustered mostly at the last part of the branches. When young, leaves are pale green and maroon. Small, whitish or creamy yellow clusters of flowers appear on the branchlets in June to December. The fruits are thinly fleshed and round in shape, green when young and purple when mature (Githiori, 2004).



Figure 4: Photographs of *R. melanophloeos*

The grey bark and roots are used ethno-medicinally for respiratory problems, stomach, muscular and heart complaints. The bark, roots and fruits are reported to have anti-helmintic properties. The fruits are particularly used as anti-helmintic in livestock and humans, primarily against cestodes by either being chewed or eaten in porridge (Midiwo *et al.*, 2002). According to Githiori *et al.* (2002) the plant has some *in vivo* anti-helmintic activity against the nematode parasite, *Haemonchus contortus*.

Triterpenoid saponins (**19**) isolated from *Rapanea melanophloeos* have been scientifically proven to have molluscicidal and antifungal activities (Kazuhiro *et al.*, 2002). The compound 3 β -Hydroxylanosta-9,24-dien-21-oic acid (**17**) isolated from the bark, has been reported to exhibit anti-platelet aggregation activity hence can be used to treat clotting related ailments (Gwala, 2011). This property is attributable to its antioxidant activity. Phytochemical analysis shows that the plant contains saponin, terpenoids, tannins, alkaloids, flavanoids, cardiac glycosides and phlobatanins. The plant also contains benzoquinones particularly Embelin (**18**), 2,5-dihydroxy-3-undecyl-1,4-benzoquinone, which has larvicidal effects against *Aedes aegypti*, antibacterial activity, antitumor, anti-inflammatory, anti-helminthic, antioxidant and analgesic activity (Marston *et al.*, 2000; Joy and Lakshimi, 2010).

The family Myrsinaceae is reported to contain benzoquinones as the major compounds. The family also accumulates flavonols in their glycoside form as well as alkaloids but none have been reported in *R. melanophloeos*. Calcium oxalate crystals have also been found in the tissues of plants in this family. The family exhibits a range of biological activity including acaricidal, antimicrobial, insecticidal, nematocidal and phototoxic activity (Lukhele, 2010).

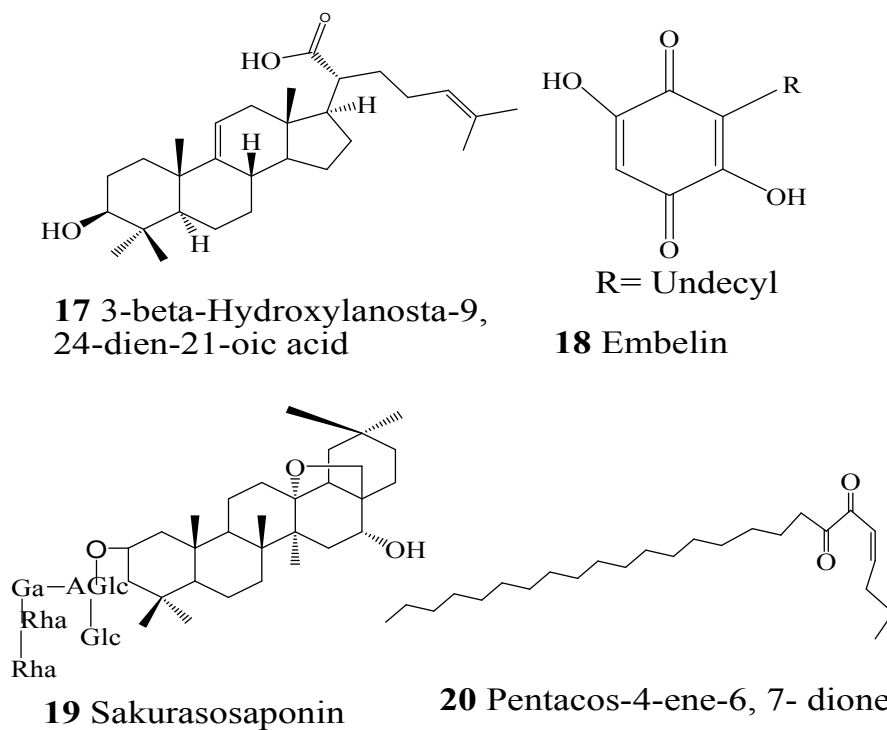


Figure 5: Some of the compounds isolated from *R. melanophloeos* (Joy and Lakshimi 2010; Gwala, 2011).

CHAPTER THREE

MATERIALS AND METHODS

3.1 Collection of plant material

Plant materials of *T. nobilis* and *R. melanophloeos* were collected from Kakamega rain forest (0° 10' – 0° 21' N 34° 44' – 34° 58' E) and Sururu Forest, Mau Narok, Nakuru County (0°66' 55.2"S 36°1' 47.3"E) respectively. They were identified by a taxonomist at the Department of Biological Sciences, Egerton University, where a voucher specimen was deposited. The *T. nobilis* materials were divided into two. One batch of the *T. nobilis* and *R. melanophloeos* materials were taken to the centre for Herbal research in Egerton University where they were dried under shade for three weeks. The other batch of *T. nobilis* was taken to Biotechnology laboratory for extraction of essential oil.

3.2 Extraction and analysis of essential oil

3.2.1 Distillation of essential oil

The fresh leaves of *T. nobilis* were subjected to hydro distillation in a modified Clevenger apparatus to extract essential oils. These leaves were cut into pieces of about 2 X 2 cm and 100g boiled with 500 ml of distilled water in a modified Clevenger apparatus until oil distillation ceased after 4-6 h. The essential oil in the distillate was dried over anhydrous Na₂SO₄ and refrigerated at 4°C.

3.2.2 Essential oil chemical composition determination and spectroscopic analysis

The essential oil was analysed by use of an Agilent GC-MSD apparatus equipped with an Rtx-5SIL MS ('Restek') (30 m x 0.25 mm, 0.25 µm film thickness) fused-silica capillary column. Briefly, the following protocol was applied. The essential oil was diluted in methyl-t-butyl ether (MTBE) (1:100). The carrier gas used was Helium (at 0.8 mL/min). Sample was injected in the split mode at a ratio of 1:10 – 1:100. The injector was kept at 250°C and the transfer line at 280°C. The column was maintained at 50°C for 2 min and then programmed to 260°C at 5°C/min and held for 10 min at 260°C.

The MS was operated in the electron impact ionization (EI) mode at 70 eV, in m/z range 42-350. The identification of the compounds was performed by comparing their retention indices and mass spectra with those found in literature and supplemented by Wiley 7N.1, HPCH 1607.L and FLAVORS.L GC-MS libraries. The relative proportions of the essential oil constituents are expressed as percentages obtained by peak area normalization, all relative response factors being taken as one. A summary of the extraction is shown as per the flow diagram in figure 6.

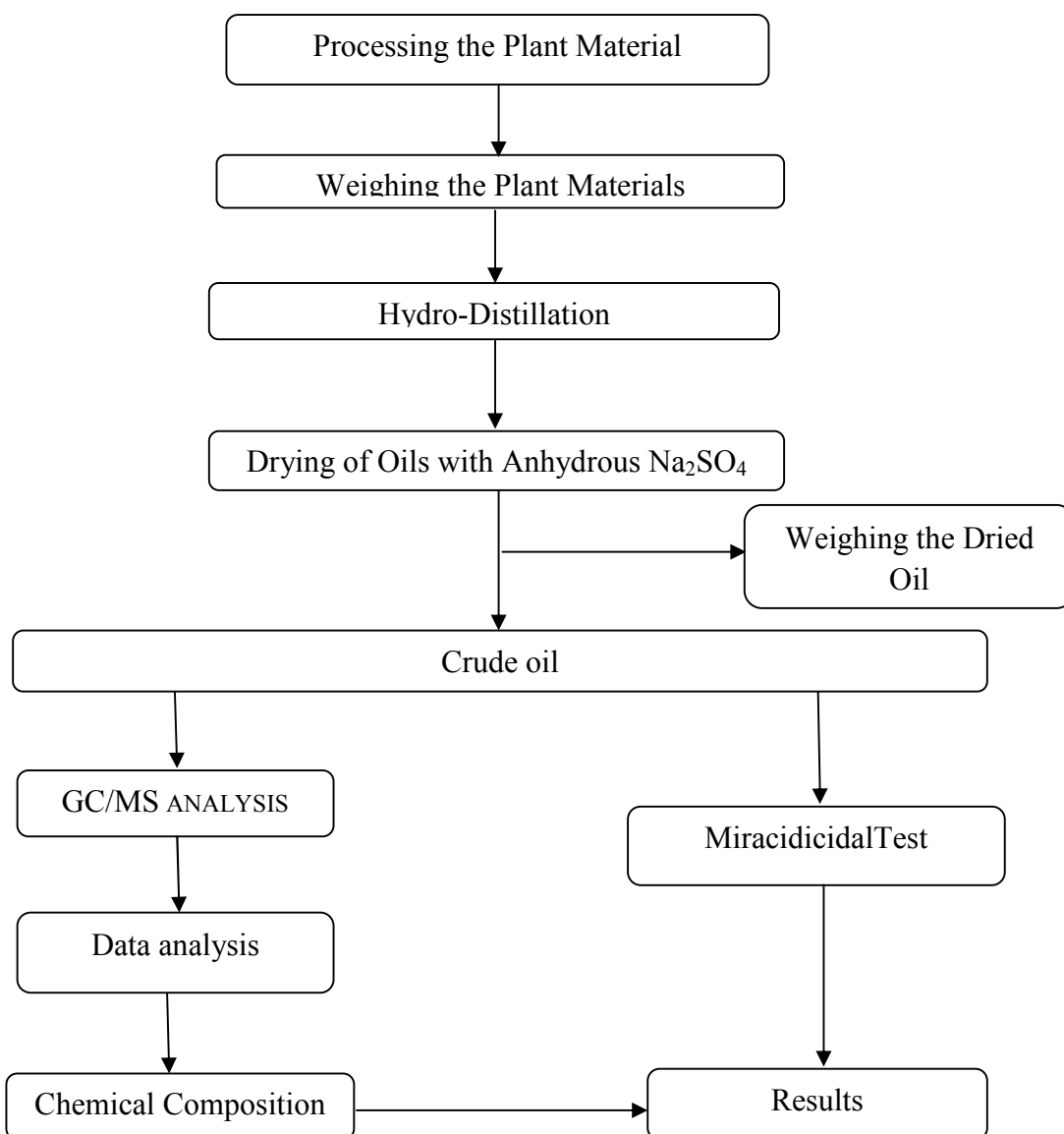


Figure 6: A flow chart showing the summary of isolation and analysis of essential oil

3.3 Extraction, isolation and structure elucidation of non-volatiles

Extraction was done using solvents of different polarities acquired from Indo laboratories. The solvents were of GPR grade and were distilled before use. These solvents included methanol, ethyl acetate and hexane. The procedure of the extraction, isolation and structure elucidation of the pure compounds is briefly described below and summarized in figure 7.

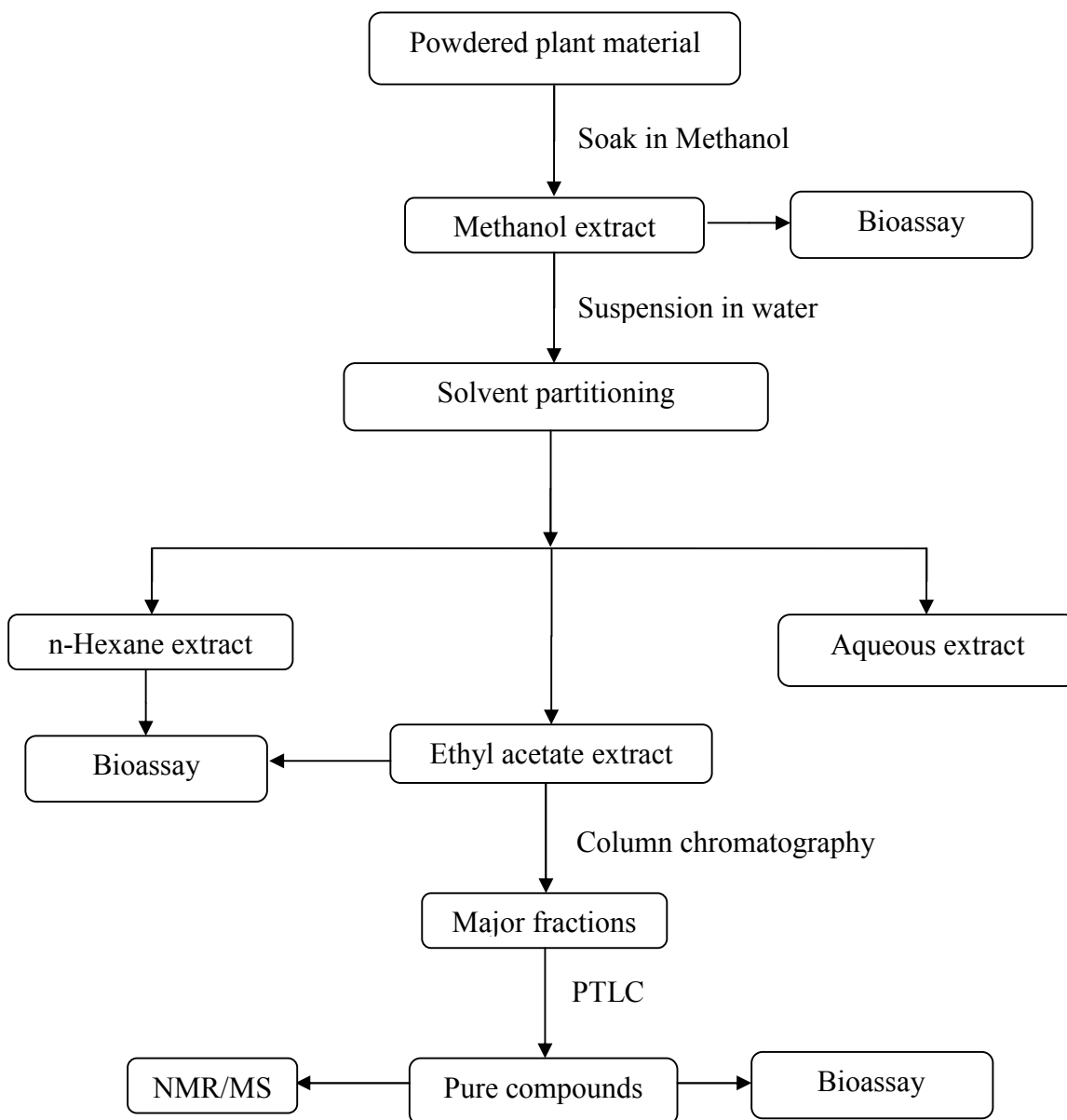


Figure 7: A summary of isolation and structure elucidation of non-volatiles

3.3.1 Extraction of phytochemicals

Leaves of *T. nobilis* and *R. melanophloeos* were dried under shade and at ambient temperatures to avoid loss of labile compounds and also avoid formation of other artefacts. To avoid the formation of mould during drying, the plant materials were periodically turned over. The materials were separately ground to a fine powder using a Thomas-Wiley mill model 4. Four and five kilograms of the *T. nobilis* and *R. melanophloeos* ground materials were respectively soaked in methanol at room temperature for 24 hours with periodical shaking. The contents were filtered through Whatman no. 1 filter paper and the filtrate evaporated to dryness in vacuum at 40°C using Buchi Rotavapor R-205 rotary evaporator. The methanol crude extracts were dried in a fume hood.

3.3.2 Solvent partitioning of crude methanol extract

The crude methanol extracts of *T. nobilis* and *R. melanophloeos* were subjected to liquid-liquid fractionation by suspending them in water and sequentially extracting with hexane and ethyl acetate to afford three extracts namely; hexane, ethyl acetate and aqueous extracts. From the hexane and ethyl acetate extracts, the solvents were removed from the hexane and ethyl acetate extracts in a rotary evaporator at 40°C and residues dried completely in a fume hood. Ethyl acetate extracts were then subjected to extensive TLC analysis and column chromatography as outlined below.

3.3.3 Thin layer chromatography (TLC)

Analytical TLC was performed on silica gel GF 254nm, (Merck, Germany) 0.25mm thickness. Briefly, the dry extracts were reconstituted in ethyl acetate to make up a final concentration of about 10mg/cm³. To ensure homogeneity all extracts were thoroughly mixed. Preliminary analysis was performed to identify optimum solvent systems for use as mobile phases. Ethyl acetate and hexane solvent mixtures were tried and modified accordingly to give optimum separation for ethyl acetate extracts. The solvent mixtures that were giving optimum separations were, 5:5 ethyl acetate-hexane mixtures for *R. melanophloeos* and 6:4 ethyl acetate-hexane mixtures for *T. nobilis*. Visualization was done by illumination under UV lamp (Uvitec-LF-204.LS) at 254 nm and 365nm.

Samples were spotted on 4×5cm or 2×5cm aluminium backed TLC plates. The spotting was done with care being taken not to over load the plate. For each of the different plants, samples were spotted on separate plates. In each plate samples were about half a centimeter from the base and developed to a distance of 4cm in an aluminium foil covered 100ml glass beaker pre-saturated with the corresponding mobile phase. The developed chromatograms were then visualized under UV light at 254 nm and 365 nm to detect UV visible compounds.

3.3.4 Column chromatography

After intensive TLC analysis of *T. nobilis* and *R. melanophloeos* ethyl acetate extracts to identify the optimum solvent systems that were giving good separation of compounds, the extracts were then prepared for column chromatography. The dry extracts were re-dissolved in minimum amount of ethyl acetate and separately loaded on evenly packed silica gel columns, by dripping on the column walls, cautioning the disturbance of the silica gel layer. Silica gel 60 0.06-0.2mm (70-230mesh ASTM) supplied by Scharlau Lab supplies Limited was used for the column chromatography. The columns were eluted gradually with the appropriate mobile phase. The lengths of the columns were 50 cm with a diameter of 20mm and the flow rates were maintained at approximately 20ml/5min.

Fractions of equal volumes were collected and the TLC of each fraction done. Fractions with nearly similar TLC patterns were pooled. The *T. nobilis* ethyl acetate; extract yielded five major fractions while *R. melanophloeos* yielded four major fractions. Fraction three of *R. melanophloeos* was added in activated charcoal, filtered, concentrated in vacuo and eluted on a column chromatography with a pre-determined mobile phase of 4:6 Ethyl acetate-Hexane mixtures. From this, three sub-fractions were collected then subjected to PTLC. This yielded two purified fractions RMF₃C₂(**23**) and RMF₃C₃(**24**). *T. nobilis* fractions three and four were also subjected to PTLC. This yielded two purified fractions TNF₃C₁(**10**, **13** and **21**) and TNF₄C₁(**22**).

3.3.5 Purification of compounds

The preparative TLC (PTLC) were prepared using silica gel 'G' (for TLC, containing 13% calcium sulphate) supplied by Laborama limited on 20 cm × 20 cm glass plates. 90 g of silica gel was added to 20 ml of distilled water to make slurry. The slurry was then evenly spread on the glass plates to ensure uniform thickness. Care was taken to avoid very thick

plates. Fractions that were completely dry were re-dissolved in ethyl acetate and spotted on a preparative TLC plate with care not to overload using pasture pipette. The plates were then developed to a distance of 18 cm in a closed TLC tank pre-saturated with the corresponding pre-determined solvent mixtures. The plates were then visualized under 254 nm UV radiation and the bands containing the target compounds scraped off. The compounds were then extracted with ethyl acetate, filtered and then concentrated in vacuum at 40⁰C. The purity of the compounds was determined by TLC. Where the compounds did not give clear single spots they were re-spotted and developed on PTLC. They were then dried under a current of air in the fume hood. The dry samples were weighed and the w/w percentage yield calculated. The compounds were taken for further analysis using 1 and 2D high field NMR spectroscopy and mass spectrometry.

3.3.6 Nuclear magnetic resonance (NMR) spectroscopy

The ¹H, APT, HSQC, COSY and HMBC NMR spectra were recorded on the Bruker Advance 500 MHz NMR spectrometer at the Technical University of Berlin, Germany. All the readings were done in Deuterated chloroform and chemical shifts assigned by comparison with the residue proton and carbon resonance of the solvent. Tetramethylsilane (TMS) was used as an internal standard and chemical shifts were given as δ (ppm). The structures were simulated using ACD NMR manager program to obtain the chemical shifts of proton.

3.3.7 Two dimensional NMR spectroscopy

The off- diagonal elements were used to identify the spin – spin coupling interactions in the ¹H –¹H COSY (Correlation spectroscopy). The proton-carbon connectivity, up to three bonds away, was identified using ¹H–¹³C HMBC (Heteronuclear Multiple Bond Correlation) spectrum. The ¹H–¹³C HSQC spectrum (Heteronuclear Single Quantum Coherence) was used to determine the connectivity of hydrogen to their respective carbon atoms. The APT (Attached proton test) spectrum was used to identify the resonances of Quaternary, methines, methylene and methyl carbon atoms.

3.3.8 Mass spectrometry

The mass spectra of the compounds were recorded on Finnigan Triple Stage Quadrupole Spectrometer (TSQ-70) with electro spray ionization (ESI) Method. The Thermo Xcalibur Qual computer software was used in analysis of the mass chromatograms.

3.4 *In vitro* anti-helminthic activity test

In vitro anti-helminthic activity was assessed through miracidia motility assays for *S. mansoni*.

3.4.1 Collection and preparation of *S. mansoni* miracidia

Stool samples were collected from volunteers who work as car washers and sand harvesters in Usoma village Kisumu County. Diagnosis was established by the help of field teams of KEMRI Schisto branch by use of the Kato-Katz faecal thick smear.

The cellophane fecal thick smears were prepared according to the method described by Lofty (2009) with some modifications. A small mound of faecal material was placed on a scrap paper and a piece of 80 mesh nylon screen (30-35mm) pressed on top so that some of the faeces sieved through the screen and accumulated on top. A flat-sided spatula was scraped across the upper surface of the screen to collect the sieved faeces. A plastic Kato-Katz template, with a hole of 6mm on a 1.5mm thick template delivering 41.7mg of faeces, was then placed on the centre of the microscope slide and faeces from the spatula added so that the hole was completely filled. The template was then passed over using the side of the spatula to remove excess faeces from the edge of the hole. The template was carefully removed so that the cylinder of faeces was left on the slide. The faecal material was covered with hydrophilic cellophane, 34µm thick, pre-soaked in glycerol and 1ml of 3% aqueous malachite green. The microscope slide was inverted and the faecal sample firmly pressed against the hydrophilic strip on another microscope slide. The faecal material was evenly spread between the microscope slide and the cellophane strip. Carefully the slide was removed by gently sliding it sideways to avoid separating the cellophane strip. The slide was then placed on the bench with the cellophane upward to allow water to evaporate while glycerol cleared the faeces. The slide was kept at room temperature for one hour to clear the faecal material, prior to examination under the microscope. The number of eggs

counted from the two slides was multiplied by 24 to obtain the number of eggs per gram stool. The samples found positive by the Kato-Katz fecal thick smears were prepared for hatching.

S. mansoni eggs were recovered from positive stools as described below. Fresh fecal sample was taken and emulsified in about 250ml of de-chlorinated water. The homogenized fecal sample was placed on a bank of arranged nested sieves in descending order of pore size from the top (750, 410, 212 and 45 μm) and passed through with aid of a cap of an inverted 50ml Tube while splashing with water using a wash bottle. The top most sieve was removed after the filtrate passed through. The same procedure was repeated with the other sieves. The schistosome ova were collected at the 45 μm sieve. The last sieve was splashed with water while agitating until clean water was seen to pass through. The filtrate containing eggs was Collected in a 50ml centrifuge tube and stored at 4⁰C. To confirm the presence of schistosome ova, a direct smear was prepared using the supernatant in the tube and observed under a dissecting Microscope.

When hatching schistosome ova, all the filtrate contained in the 50ml tube was poured into a 1litre flat bottomed flask and filled with water to the brim. The flask was exposed to the open light (next to the window) for 1 hour. The flask was covered with a dark cloth for 10 minutes. The top part of the water was poured into a Petri-dish and observed for miracidia under 10X objective under a dissecting microscope. The miracidia were picked using a micropipette and transferred into 24 well multiwell plates.

3.4.2 Effect of the extracts on miracidia

24 wells multiwell plates were used as test chambers to observe the viability and death of the miracidia under a dissecting microscope. Twenty miracidia were picked using 100 μl pipette and placed in each well. Serial dilutions of the crude extracts and major fractions were added to each well as follows; 4000, 2000, 1000, 500, 450, 400, 350, 300, 250, 200, 150, 100 and 50 ppm. Two negative controls were also set up using de-chlorinated water and 1% DMSO in de-chlorinated water, respectively. Praziquantel was used as a positive control. Three replicates were prepared for each test and mortality observed after 30 minutes. Percentage mortality was calculated and LC₅₀ for each sample determined by use of IBM SPSS 15.0.

CHAPTER FOUR

RESULTS AND DISCUSSION

4.1 Structure elucidation of isolated compounds

Six compounds were isolated and their structures elucidated. Of the six, four were from *T. nobilis* that is, Tecleoxine **10**, Methylnkolbisine **13**, Kokusagine **21** and Nkolbisine **22**. From *R. melanoploeos*, two benzoic acid derivatives; Myrsinoic acid B **23** and Myrsinoic acid C **24** were obtained. The structure elucidation of these compounds is explained in the following sections.

4.1.1 Structure elucidation of compounds **10**, **13** and **21**

These compounds were obtained as an oily dark mixture with a mass of 213.1 mg and an extended purple spot was observed under 254 nm UV irradiation on TLC. They were not resolved individually into purified forms due to their minimal amounts and close R_f values. The analysis of their NMR spectra indicated that they were three closely related alkaloids, that is, **10**, **13** and **21**. The difference between these three compounds was determined to be the substituent at C-6 (see figures 8, 10 and 12). The 1D and 2D NMR spectral data of **10** are summarized in table 1.

Compound **10** had eighteen carbon atoms, nineteen hydrogen atoms and hence the molecular formula $C_{18}H_{19}NO_5$ with ten hydrogen deficiencies. The ^{13}C APT NMR spectrum showed the presence of four aromatic CH signals at δ 142.5, 104.7, δ 102.5 and 106.4. A total of seven aromatic quaternary carbons at δ 102.2, 155.9, 112.7, 146.7, 152.9, 142.3, 163.9 and a non-aromatic one at δ 58.6 were observed. Two methoxy carbons were also observed at δ 55.9 and 58.9.

The HSQC spectrum showed correlations of the carbon atoms with the protons directly attached to them. From the spectrum there were correlations between protons at δ 7.47, 6.96, 7.43, 7.24, 4.12, 3.17, 1.30, 1.28, 4.32, 3.90 and C-2, C-3, C-5, C-8, C-1', C-2', C-4', C-5', 4-OMe, 7-OMe respectively. The 4-Omethyl group at δ 4.32 is characteristic for furoquinoline alkaloids. However, the coupling constants and the proton multiplicities were not determined as the H spectrum had distorted baseline with presence of broad peaks (Appendix 36).

The proton-carbon HMBC spectrum showed proton correlations with carbon atoms which are two bonds or three bonds away. This helps in identification of carbon atoms which are next to each other and those which are two bonds away from each other.

Table 1: NMR data of compound 10

Carbon	¹³ C(δ)	¹ H(δ)	APT	COSY	HMBC	¹³ C(δ) Literature*
2	142.5	7.47	CH	3	3, 3a, 9a	142.7
3	104.7	6.96	CH	2	2, 3a, 9a	104.9
3a	102.2	-	C	-	-	102.2
4	155.9	-	C	-	-	155.9
4a	112.7	-	C	-	-	113.1
5	102.5	7.43	CH	-	4, 4a, 6, 7, 8a	102.7
6	146.7	-	C	-	-	147.0
7	152.9	-	C	-	-	153.1
8	106.4	7.24	CH	-	4a, 6, 7, 8a	107.2
8a	142.3	-	C	-	-	143.1
9a	162.9	-	C	-	-	163.5
1'	67.8	4.12	CH ₂	2'	6, 2', 3'	68.1
2'	61.2	3.17	CH	1'	1', 3', 4', 5'	61.6
3'	58.6	-	C	-	-	58.4
4'	24.5	1.30	CH ₃	-	2', 3', 5'	24.9
5'	19.0	1.28	CH ₃	-	2', 3', 4'	19.4
4-OMe	58.9	4.32	CH ₃	-	4	59.2
7-OMe	55.9	3.90	CH ₃	-	7	56.3

*Adnan *et al.*, 2003

From the HMBC spectrum the proton H-2 absorbing at δ 7.47 showed correlation with aromatic carbons C-3 and C-3a which are one bond away and a quaternary carbon, C-9a, which is two bonds away. Similarly, H-3 absorbing at δ6.96 correlated with carbons at position C-2, C-3a, and C-9a. The aromatic proton H-5 absorbing at δ7.43 showed correlations with C-4, C-4a,

C-6, C-7 and C-8a while H-8 absorbing at δ 7.24 correlated with carbons at position C-4a, C-6, C-7 and C-8a. The protons H-1' (δ 1.30) and H-2' (δ 1.28) correlates with C-6, C-2', C-3' and C-1', C-3', C-4', C-5', respectively. This helped in confirmation of the substituent at C-6. In the same note, the methoxy protons at δ 4.32 and δ 3.90 also helped in confirmation of these methoxy groups to their respective positions as they correlated with C-4 and C-7, respectively.

The proton-proton COSY correlation for the compound **10** was also determined. COSY spectrum gave information on protons which are attached to adjacent carbons. Aromatic protons H-2 (δ 7.47) and H-3 (δ 6.96) correlated with each other. Protons H-1' (δ 4.12) and H-2' (δ 3.17) also showed correlations. The structure is shown in figure 8.

The high resolution positive electron impact mass spectrometry (HREIMS) of this compound at 9.55 minutes retention time showed a molecular ion peak at m/z 330.13 ($[M+H]^+$) (calculated for $[C_{18}H_{19}NO_5 + H]$; m/z 330.14). The mass spectrum of the compound is shown in figures 9.

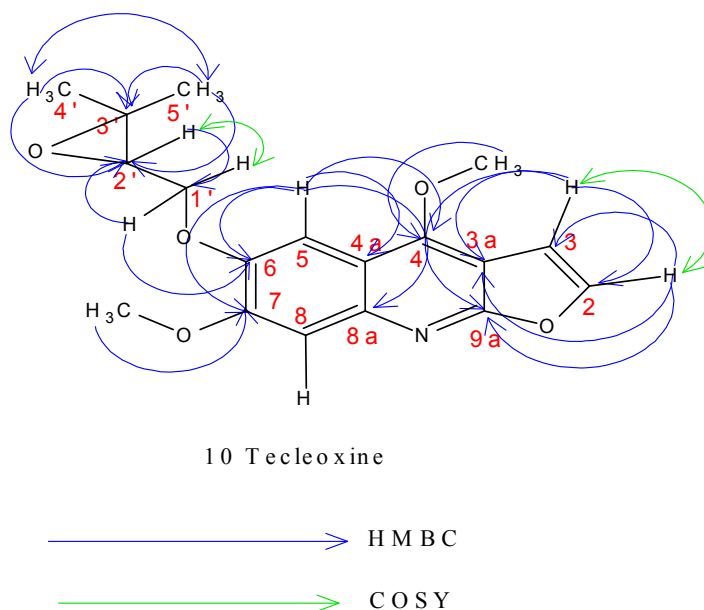


Figure 8: Structure of compound 10 showing COSY and HMBC

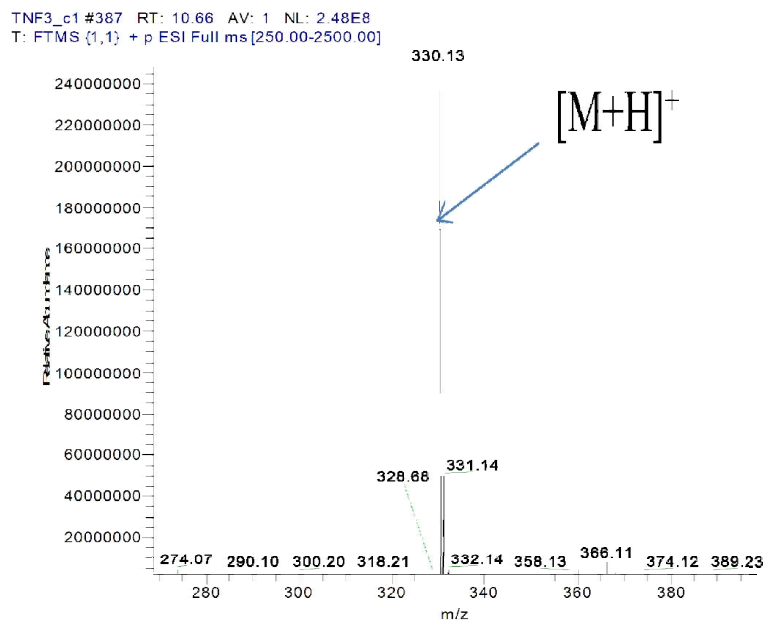


Figure 9: Mass spectrum of compound 10

4.1.2 Structure elucidation of compound 13

The NMR spectra of this compound were similar to that of compound **10** with the difference only being in the substituent at C-6. From the ^{13}C APT NMR spectrum, the substituent had six carbons, one CH (C-1'- δ 68.3), one CH_2 (C-2'- δ 71.0), one Cq (C-3'- δ 71.8), two CH_3 (C-4'- δ 26.4 and C-5'- δ 26.8) and one methoxy carbon (3'-OMe- δ 52.5). The 1D and 2D NMR spectral data are summarized in table 2.

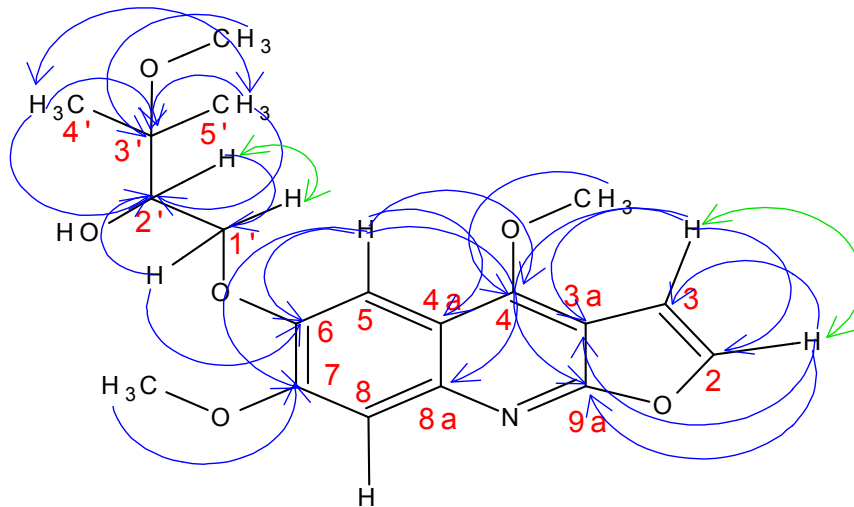
HSQC spectrum showed that the protons absorbing at δ 3.42/3.37, δ 3.63, δ 1.34 and δ 3.62 correlated with carbon atoms C-1', C-2', C-4'/C-5' and C-3'-OMe respectively. The proton-proton COSY showed correlations between protons H-1' and H-2' absorbing at δ 3.42 and δ 3.63 respectively. The HMBC spectrum showed proton H-1' absorbing at δ 3.42/3.37 correlated with carbons C-6, C-2' and C-3'. Proton H-2', absorbing at δ 3.63 correlated with carbons C-1', C-3', C-4', C-5' and protons H-4' and H-5' absorbing at δ 1.34 correlated with carbons C-2' and C-3'. The methoxy protons absorbing at δ 3.62 also showed correlation with carbon C-3'. This helped in confirmation of the position of the methoxy group in the substituent.

Table 2: NMR data of compound 13

Carbon	¹³ C(δ)	¹ H(δ)	APT	COSY	HMBC	¹³ C(δ) Literature*
2	142.5	7.47	CH	3	3, 3a, 9a	142.2
3	104.7	6.96	CH	2	2, 3a, 9a	104.3
3a	102.2	-	C	-	-	101.9
4	155.9	-	C	-	-	155.4
4a	112.7	-	C	-	-	112.7
5	102.5	7.43	CH	-	4, 4a, 6, 7, 8a	102.3
6	146.7	-	C	-	-	146.8
7	152.9	-	C	-	-	152.7
8	106.4	7.24	CH	-	4a, 6, 7, 8a	106.2
8a	142.3	-	C	-	-	142.6
9a	162.9	-	C	-	-	163.0
1'	71.0	3.42, 3.37	CH ₂	2'	6, 2', 3'	70.1
2'	68.3	3.63	CH	1'	1', 3', 4', 5'	74.6
3'	71.8	-	C	-	-	75.9
4'	26.4	1.34	CH ₃	-	2', 3', 5'	21.1
5'	26.8	1.34	CH ₃	-	2', 3', 4'	20.4
3'-OMe	52.5	3.62	CH ₃	-	3'	49.2
4-OMe	58.9	4.32	CH ₃	-	4	58.6
7-OMe	55.0	3.90	CH ₃	-	7	55.7

*Adnan *et al.*, 2003

The compound was identified as with molecular ion peak m/z 362.2 ($[M+H]^+$) (calculated for $[C_{19}H_{23}NO_6+ H]$ (m/z 362.15) using high resolution positive electron impact mass spectrometry (HREIMS) at 10.62 minutes retention time. The mass spectrum of the compound is shown in figure 11.



13 Methylkolbisine

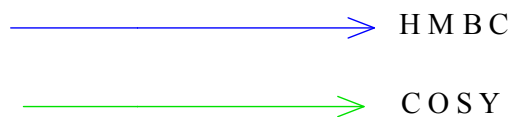


Figure 10: Structure of compound 13 showing COSY and HMBC

TNF3_c1 #342 RT: 9.64 AV: 1 NL: 7.80E3
 T: FTMS (1,2) - p ESI Full ms[250.00-2500.00]

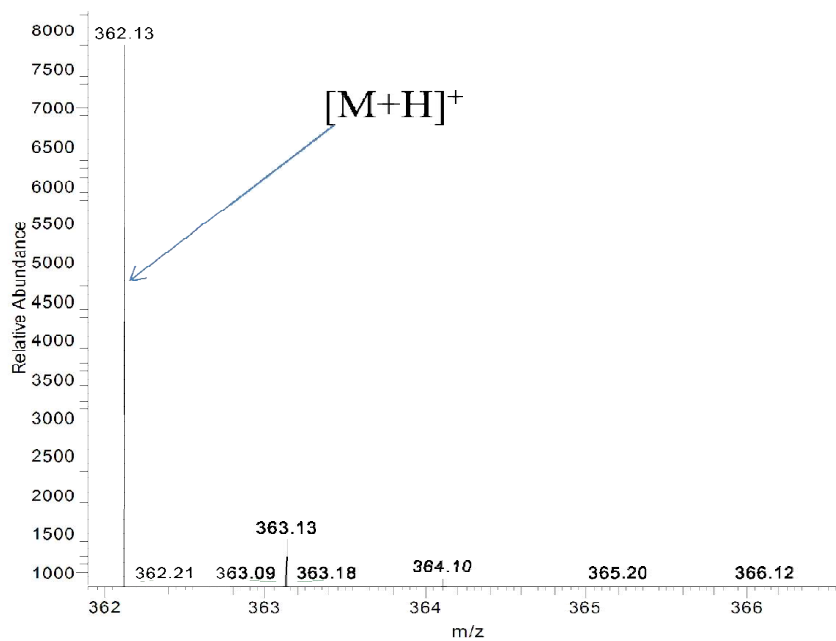


Figure 11: Mass spectrum of compound 13

4.1.3 Structure elucidation of compound 21

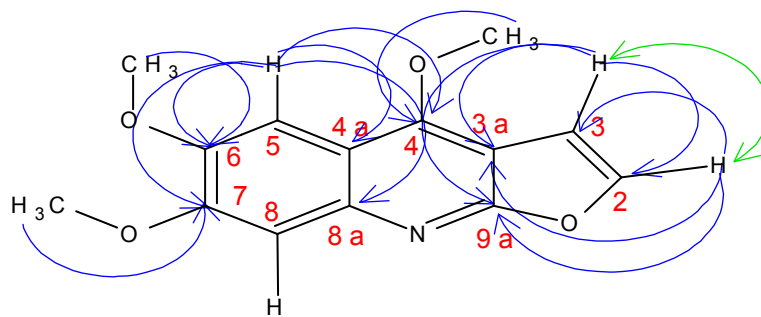
The NMR spectra of this compound were similar to that of compound **10** with the difference only being in the substituent at C-6. From the ^{13}C APT NMR spectrum, the substituent had one methoxy carbon (6-OMe) absorbing at δ 52.5. The position of the methoxy group was confirmed using HMBC in which the proton 6-OMe absorbing at δ 3.73 showed correlation with carbon C-6. The structure of the compound is shown in figure 12 below. The 1D and 2D spectral data are summarized in table 3.

The compound was identified as with molecular ion peak m/z 260.1 ($[\text{M}+\text{H}]^+$) (calculated for $[\text{C}_{14}\text{H}_{13}\text{NO}_4 + \text{H}]$ (m/z 260.3) using high resolution positive electron impact mass spectrometry (HREIMS) at 10.62 minutes retention time. The mass spectrum of the compound is shown in figure 13.

Table 3: NMR data of compound 21

Carbon	$^{13}\text{C}(\delta)$	$^1\text{H}(\delta)$	APT	COSY	HMBC	$^{13}\text{C}(\delta)$ Literature*
2	142.5	7.47	CH	3	3, 3a, 9a	142.7
3	104.7	6.96	CH	2	2, 3a, 9a	104.8
3a	102.2	-	C	-	-	102.4
4	155.9	-	C	-	-	-
4a	112.7	-	C	-	-	112.9
5	102.5	7.43	CH	-	4, 4a, 6, 7, 8a	100.2
6	146.7	-	C	-	-	-
7	152.9	-	C	-	-	-
8	106.4	7.24	CH	-	4a, 6, 7, 8a	105.8
8a	142.3	-	C	-	-	142.7
9a	162.9	-	C	-	-	-
4-OMe	58.9	4.32	CH ₃	-	4	59.1
6-OMe	56.3	3.73	CH ₃	-	6	56.2
7-OMe	55.0	3.90	CH ₃	-	7	56.0

*Adnan *et al.*, 2003



21 Kokusagine

→ H M B C

→ C O S Y

Figure 12: Structure of compound 21 showing COSY and HMBC

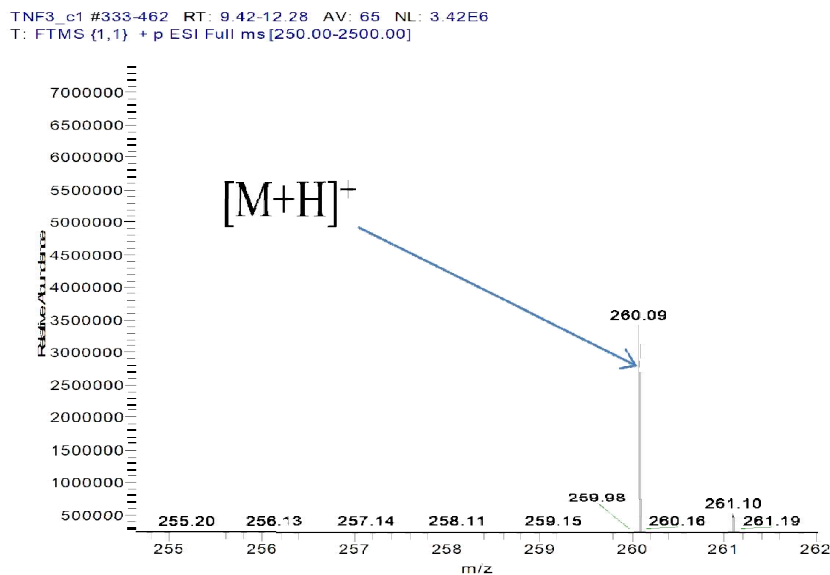


Figure 13: Mass spectrum of compound 21

4.1.4 Structure elucidation of compound 22

The compound **22** was obtained as an oily dark substance with a mass of 192.6 mg. The ^{13}C APT and ^1H NMR spectra of the compound, like the previously identified compounds, showed great similarity with the compound **10** with the major difference being on the substituent at C-6. The 1D and 2D spectral data are summarized in table 4.

The ^{13}C APT NMR spectrum showed the presence of five carbon atoms attributed to the substituent at C-6. These included a CH signal at $\delta 75.2$, a CH_2 at $\delta 70.9$, a Cq at $\delta 72.1$ and two CH_3 at $\delta 26.4$ and $\delta 25.9$. From HSQC spectrum, there was clear correlations between, C-1' and the proton absorbing at $\delta 4.31/4.14$, C-2' and the proton absorbing at $\delta 3.83$, C-4' and the proton absorbing at $\delta 1.29$ and C-5' and the proton absorbing at $\delta 1.27$. However, the coupling constants and the proton multiplicities were not determined as the H spectrum had distorted baseline with presence of broad peaks (Appendix 37).

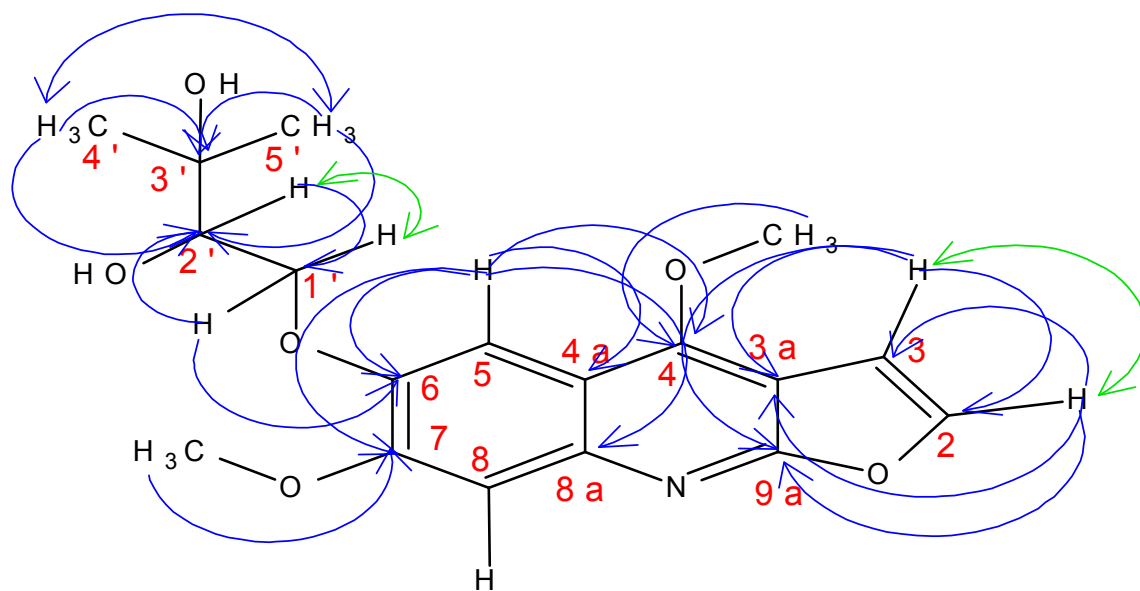
Table 4: NMR data of compound 22

Carbon	$^{13}\text{C}(\delta)$	$^1\text{H}(\delta)$	APT	COSY	HMBC	$^{13}\text{C}(\delta)$ # Literature(DMSO)
2	142.5	7.52	CH	3	3, 3a, 9a	143.6
3	104.8	7.01	CH	2	2, 3a, 9a	105.9
3a	102.3	-	C	-	-	102.5
4	155.9	-	C	-	-	155.7
4a	112.9	-	C	-	-	112.8
5	102.1	7.42	CH	-	4, 4a, 6, 7, 8a	101.4
6	146.6	-	C	-	-	147.7
7	152.6	-	C	-	-	153.1
8	106.4	7.28	CH	-	4a, 6, 7, 8a	107.0
8a	142.3	-	C	-	-	142.3
9a	163.0	-	C	-	-	163.1
1'	70.9	4.31, 4.14	CH_2	2'	6, 2', 3'	71.6
2'	75.2	3.83	CH	1'	1', 3', 4', 5'	76.3
3'	72.1	-	C	-	-	71.0
4'	26.4	1.29	CH_3	-	2', 3', 5'	24.8
5'	25.9	1.27	CH_3	-	2', 3', 4'	27.9
4-OMe	58.9	4.38	CH_3	-	4	59.8
7-Ome	56.0	3.93	CH_3	-	7	56.2

Cao *et al.*, 2009

The proton-carbon HMBC spectrum showed that the proton H-1' absorbing at δ 4.31/4.14 correlated with carbons C-6, C-2' and C-3'. Similarly, the proton absorbing at δ 3.83 correlated with carbons C-1', C-3', C-4' and C-5'. There was also correlations between proton absorbing at δ 1.29 and carbons C-2', C-3' and C-5'. The carbons C-2', C-3' and C-4' also had strong correlation with proton absorbing at δ 1.27. The proton-proton COSY correlation showed correlations between H-1' and H-2' which absorbed at δ 4.31/4.14 and δ 3.83 respectively. The structure of the compound is shown in figure 14.

The compound was identified as with molecular ion peak m/z 348.1 ($[M+H]^+$) (calculated for $[C_{18}H_{22}NO_6 + H]$ (m/z 348.4) using high resolution positive electron impact mass spectrometry (HREIMS) at 9.55 minutes retention time. The mass spectrum of the compound is shown in figure 15.



22 N k o l b i s i n e

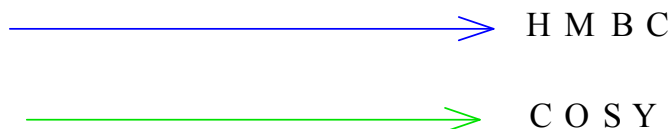


Figure 14: Structure of compound 22 showing COSY and HMBC

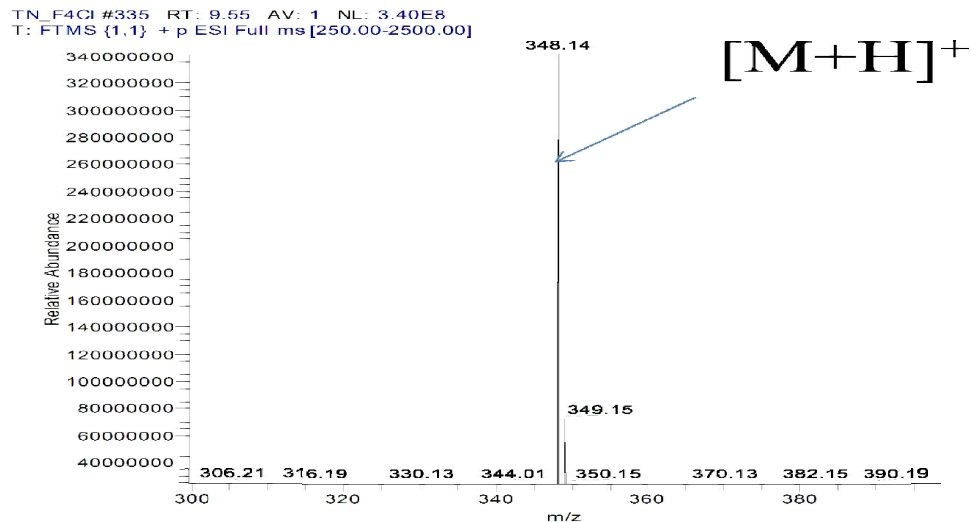


Figure 15: Mass spectrum of Compound 22

The compounds **10**, **13** and **21** were confirmed to be alkaloids Tecleoxine, Methylnkolbisine and Kokusagine earlier isolated from *T. nobilis* (Adnan *et al.*, 2003). The compounds were confirmed to be the actual compounds by comparison of their spectroscopic data and their mass spectrometric data with that from the literature. Kokusagine has also been isolated from the plants *Melicope bonickii*, *Haplophylum species* and *Esenbeckia leiocarpa* (Komala *et al.*, 2006 Elaine *et al.*, 2010; Ulubelen and Oztork, 2008). These compounds have also been isolated from other plants of the family Rutaceae including the genus *Teclea* which is well endowed with furoquinoline alkaloids (Tarus, 2005; Rios *et al.*, 2002; Ayafor and Okogun, 1982; Riyanto *et al.*, 2001; Terezen *et al.*, 2010; Wansi *et al.*, 2008; Kiplimo, 2012).

Compound **22** was identified as Nkolbisine. This was determined by comparison of its NMR and mass spectral data with those reported in literature (Cao *et al.*, 2009). This compound has been isolated in other plants (Ulubelen and Oztork, 2008; Kiplimo, 2012) but it is the first time it is being reported in *T. nobilis*.

4.1.5 Structure elucidation of compound 23

The compound **23** was obtained as an oily yellow substance with a mass of 450.2mg. The analysis of APT NMR spectra of this compound revealed the presence of 22 carbon atoms with ten of them being olefinic inclusive those of the benzene ring. Of these there were two aromatic

methine signals at δ 124.2 and δ 131.3. Two other olefinic methine signals at δ 121.5 and δ 124.2 and a hydroxymethine at δ 89.3 were also observed. There were also a total of eight quaternary carbon atoms absorbing at δ 73.9, 121.8, 122.9, 127.1, 31.9, 132.9, 162.4 and 171.5. Moreover, a total of four methylene carbon atoms absorbing at δ 21.9, 28.3, 29.9 and 37.6 were observed. Five methyl carbon atoms absorbing at δ 17.6, 17.8, 22.2, 25.7 and 25.7 were also observed.

The HSQC spectrum showed correlations of the protons to the carbon atoms in which they were directly bonded. From the spectrum the protons absorbing at δ 4.77, 3.13/3.25, 7.72, 7.73, 1.55/1.63, 2.09/2.20, 5.12, 1.67, 1.29, 1.72, 3.26/3.31, 5.29, 1.73 and 1.63 and the carbons absorbing at δ 89.3, 29.9, 124.9, 131.3, 37.6, 21.9, 124.2, 25.7, 22.2, 17.8, 28.3, 121.5, 25.7 and 17.6, respectively showed correlations.

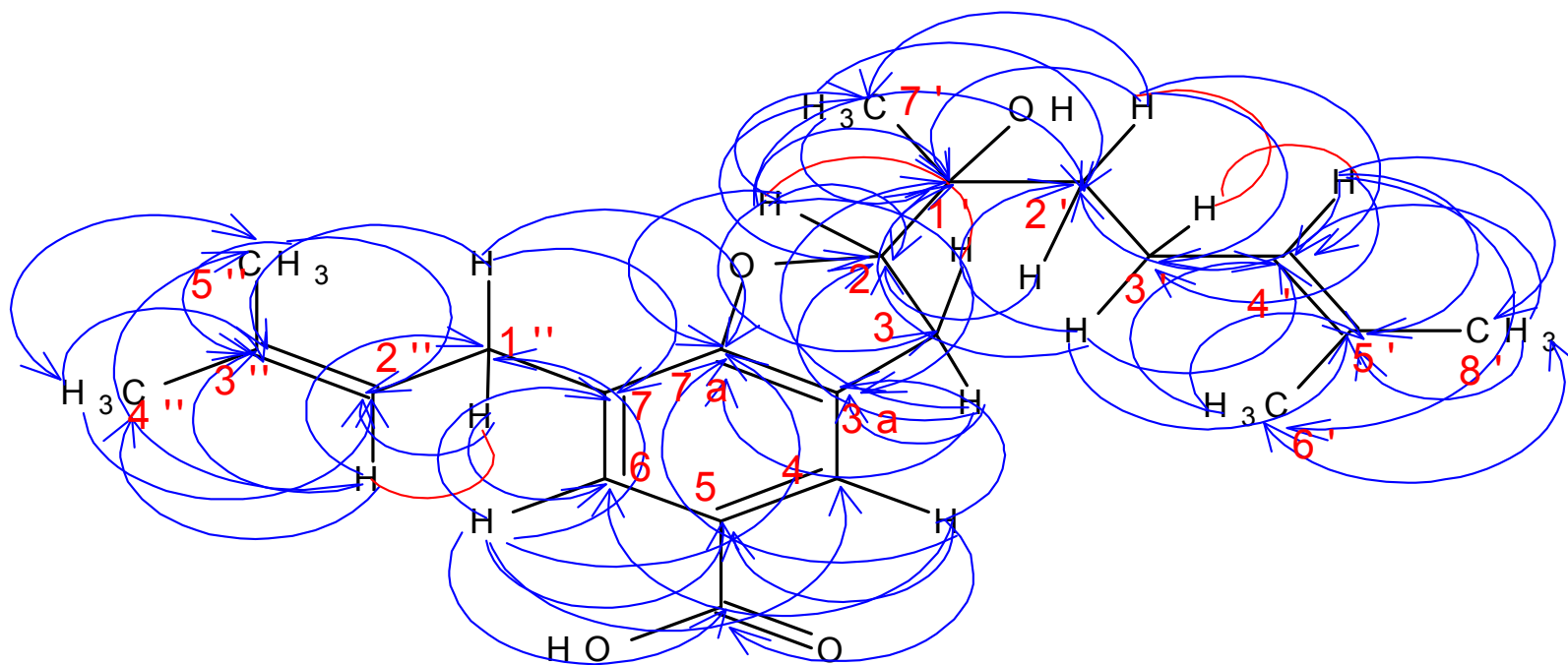
The HMBC spectrum helps in identification of carbon atoms which are next to each other and those two bonds away. These correlations are shown in table 5 and figure 16 below. The COSY spectrum showed correlations of H-2 and H-3, H-3' and H-4' as well as H-1'' and H-2''. The protons that showed correlations indicated that they were bonded to adjacent carbon atoms. A summary of the NMR spectral data is shown in table 5.

The compound was determined to be with eight hydrogen deficiencies thus the molecular formula $C_{22}H_{30}O_4$. It was identified as with molecular ion peaks m/z 359.22 ($[M+H]^+$) (calculated for $[C_{22}H_{30}O_4 + H]$ (m/z 359.47), m/z 381.20 ($[M+Na]^+$) (calculated for $[C_{22}H_{30}O_4 + Na]$ (m/z 381.47) and m/z 397.20 ($[M+K]^+$) (calculated for $[C_{22}H_{30}O_4 + K]$ (m/z 397.47), using high resolution positive electron impact mass spectrometry (HREIMS) at 10-12 minutes retention time. The mass spectra of the compound are shown in figures 17, 18 and 19. The structure of the compound is shown in figure 16.

Table 5: NMR data of compound 23

No.	¹³ C(δ)	¹ H(δ)	APT	COSY	HMBC	¹³ C(δ)#
2	89.3	4.74 t, J= 8.99Hz	CH	3	3,3a,,7a,1',7',2'	89.6
3	29.9	3.13 dd, J= 9.76, 9.50 Hz 3.25 dd, J= 8.22, 7.70 Hz	CH ₂	2	2,3a,7a,4,1'	29.9
3a	121.8	-	C	-	-	123.1
4	124.9	7.72 s	CH	-	3,3a,5.6,7a,COOH	131.4
5	122.9	-	C	-	-	121.8
6	131.3	7.73 s	CH	-	4,5,7,7a,1'', COOH	125.0
7	127.1	-	C	-	-	127.2
7a	162.4	-	C	-	-	162.4
1'	73.9	-	C	-	-	73.8
2'	37.6	1.55, 1.63 m	CH ₂	3'	2,1',7',3'4'	37.1
3'	21.9	2.09,2.20 m	CH ₂	2',4'	1',2',4'5'	22.0
4'	124.2	5.12 t, J= 6.94, 7.19Hz	CH	3'	2',3',5'6', 8'	124.1
5'	132.9	-	C	-	-	132.2
6'	25.7	1.67 s	CH ₃	-	4',5',8'	25.7
7'	22.2	1.29 s	CH ₃	-	2, 1',2'	22.6
8'	17.8	1.72 s	CH ₃	-	4',5',6'	17.7
1''	28.3	3.26,3.31 m	CH ₂	2''	6,7,7a,2'', 3''	28.3
2''	121.5	5.29 t, J=7.45, 7.19 Hz	CH	1''	1'',3'',4'',5''	121.4
3''	131.9	-	C	-	-	133.2
4''	25.7	1.73 s	CH ₃	-	2'',3''5''	25.8
5''	17.6	1.62 s	CH ₃	-	2'',3'',4''	17.9
COOH	171.5	-	C	-	-	172.0

Hirota *et al.*, (2002)



23 MYRSINOIC ACID B

—————> HMBC

————— COSY

Figure 16: Structure of compound 23 showing COSY and HMBC

RMF3_c2 #347-437 RI: 10.06-12.00 AV: 46 NL: 1.79E6
T: FTMS (1,1) + p ESI Full ms [250.00-2500.00]

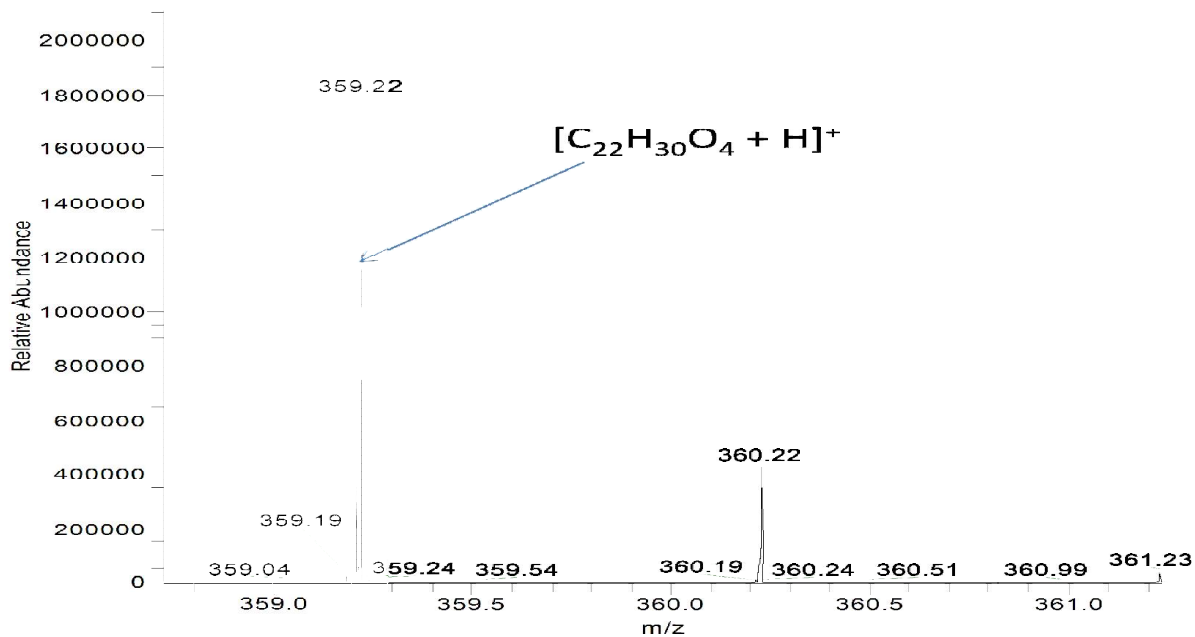


Figure 17: M+H mass spectrum of compound 23

RMF3_c2 #347-437 RT: 10.06-12.00 AV: 46 NL: 3.56E5
T: FTMS (1,1) + p ESI Full ms [250.00-2500.00]

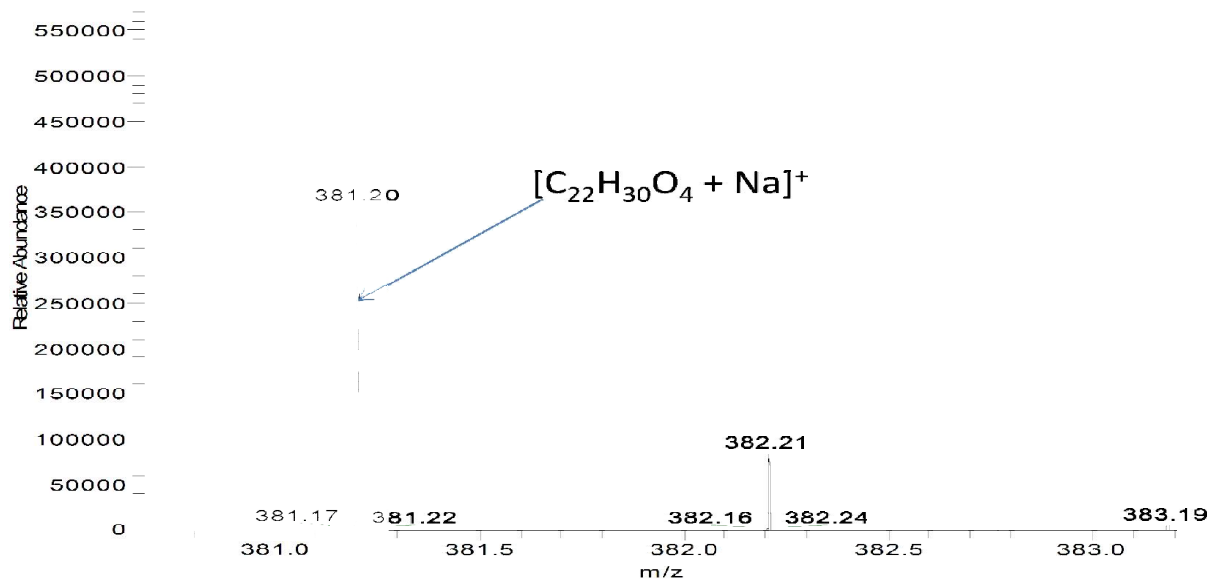


Figure 18: M+Na mass spectrum of compound 23

RMF3_c2 #347-437 RT: 10.06-12.00 AV: 46 NL: 1.29E6
T: FTMS (1,1) + p ESI Full ms [250.00-2500.00]

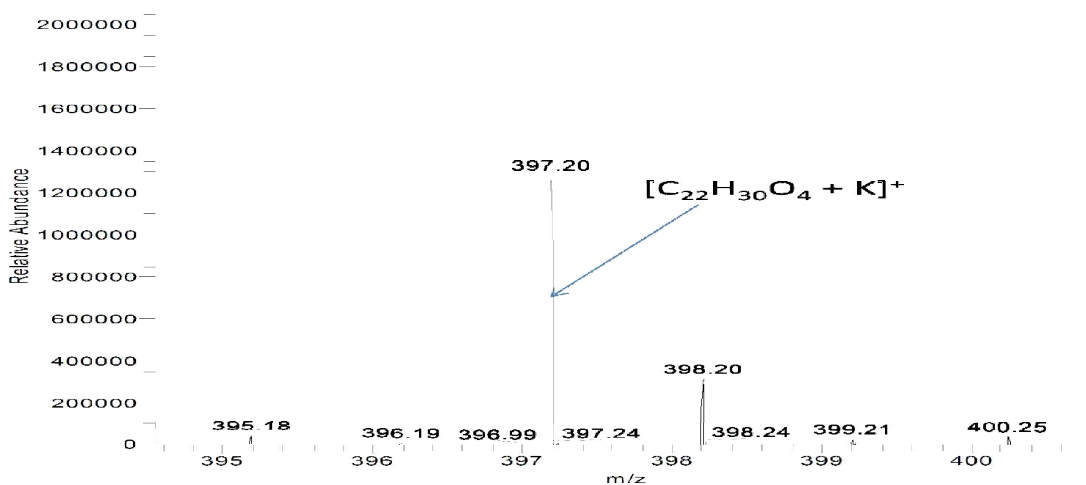


Figure 19: M+K mass spectrum of compound 23

4.1.6 Structure elucidation of compound 24

The compound **24** was obtained as an oily yellow substance with a mass of 345.3mg. The analysis of NMR spectra of this compound revealed that the compound was closely related to the compound **23**. The 1D and 2D NMR spectral data of **24** are summarized in table 6.

The compound **24** was identified to have twenty two carbon atoms and thirty two protons and hence the molecular formula $C_{22}H_{30}O_4$. The APT indicated two aromatic methine signals at δ 131.3 and 124.9. Three other methane signals were observed at δ 121.2, 124.1 and 67.8. A total of eight quaternary carbons at δ 76.9, 127.1, 123.0, 133.0, 162.3, 132.0, and δ 123.2 were observed. Four methylene signals were observed at δ 21.6, 37.4, 29.5 and 29.9 and five methyl signals absorbing at δ 17.8, 25.8, 22.3, 17.9 and 25.7.

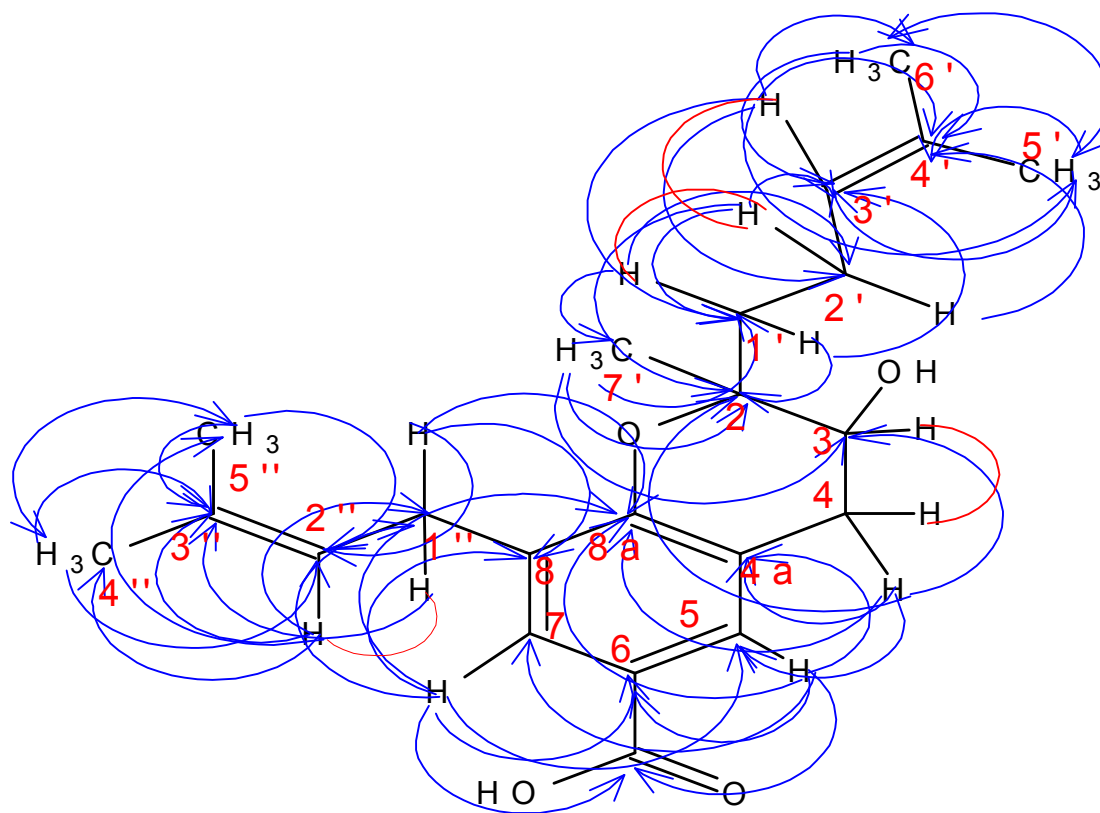
The HSQC spectrum showed correlations between protons resonating at δ 3.91, 1.32/3.13, δ 7.73, 1.53/1.63, 2.06/2.17, 5.11, 1.71, δ 1.73, 1.20/1.30, 3.32, 5.30, 1.61 and 1.68 with carbon atoms absorbing at δ 67.8, 29.5, 124.9/131.3, 37.4, 21.7, 124.1, 17.9, 25.7, 22.3, 29.9, 121.1, 17.8 and 25.8, respectively. This implied that these protons were attached to their respective carbons in which they showed correlations. However, the coupling constants and the proton multiplicities were not determined as the H spectrum had distorted baseline with presence of broad peaks (Appendix 27).

The HMBC and COSY correlations observed are shown in the figure 20 and also in table 6. The HMBC correlations indicated the carbons that are adjacent to each other and those two bonds away while COSY confirmed the protons bonded to two adjacent carbon atoms (see figure 20).

Table 6: NMR data of compound 24

Carbon	¹³ C(δ)	¹ H(δ)	APT	COSY	HMBC	¹³ C(δ) Literature#
2	76.9	-	C	-	-	79.8
3	67.8	3.91	CH	3	-	67.9
4	29.5	1.32,3.13	CH ₂	4	2,3,4a,8a,5	31.1
4a	123.2	-	C	-	-	120.9
5	124.9	7.73	CH	-	4a,8a,7,6, COOH	129.8
6	123.0	-	C	-	-	118.6
7	131.3	7.73	CH	-	5,6,8,8a,1'', COOH	130.6
8	133.0	-	C	-	-	130.2
8a	162.3	-	C	-	-	155.5
1'	37.4	1.53,1.62	CH ₂	2'	2,2',3',7'	37.6
2'	21.7	2.06,2.17	CH ₂	1',3'	2,1',3',4'	21.7
3'	124.1	5.11	CH	2'	1',2',3',4',5',6'	123.8
4'	132.0	-	C	-	-	132.3
5'	17.9	1.71	CH ₃	-	3',4',6'	17.6
6'	25.7	1.73	CH ₃	-	3',4',5'	25.6
7'	22.3	1.20,1.30	CH ₃	-	2,3,1'2'	19.0
1''	29.9	3.32	CH ₂	2''	7,8,8a,2''	28.5
2''	121.2	5.30	CH	1''	1'',3'',5'',4''	121.9
3''	127.1	-	C	-	-	132.8
4''	17.8	1.61	CH ₃	-	2'',3'',5''	17.9
5''	25.8	1.68	CH ₃	-	2'',3'',4''	25.8
COOH	171.4	-	C	-	-	171.7

Hirota *et al.*, (2002)



24 MYRSINOIC ACID C

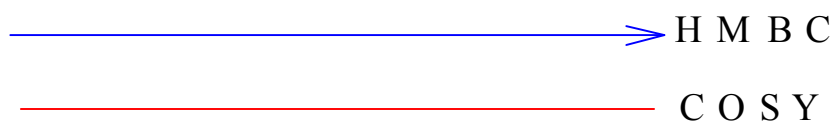


Figure 20: Structure of compound 24 showing COSY and HMBC

The compound was identified as with molecular ion peaks m/z 359.21 ($[M+H]^+$) (calculated for $[C_{22}H_{30}O_4 + H]$ (m/z 359.47) and m/z 397.20 ($[M+K]^+$) (calculated for $[C_{22}H_{30}O_4 + K]$ (m/z 397.47), using high resolution positive electron impact mass spectrometry (HREIMS) at 10-11 minutes retention time. The mass spectra of the compound are shown in figures 21 and 22. The structure of the compound is shown in figure 20.

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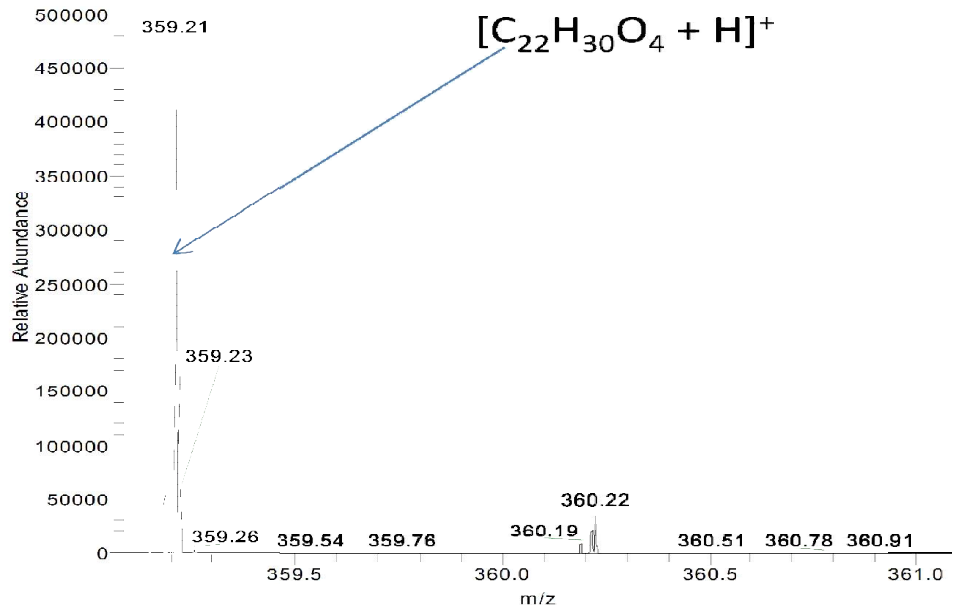


Figure 21: M+K mass spectrum of compound 24

RMF3_c3 #352-430 RT: 10.13-11.63 AV: 39 NL: 4.94E5
T: FTMS {1,1} + p ESI Full ms [250.00-2500.00]

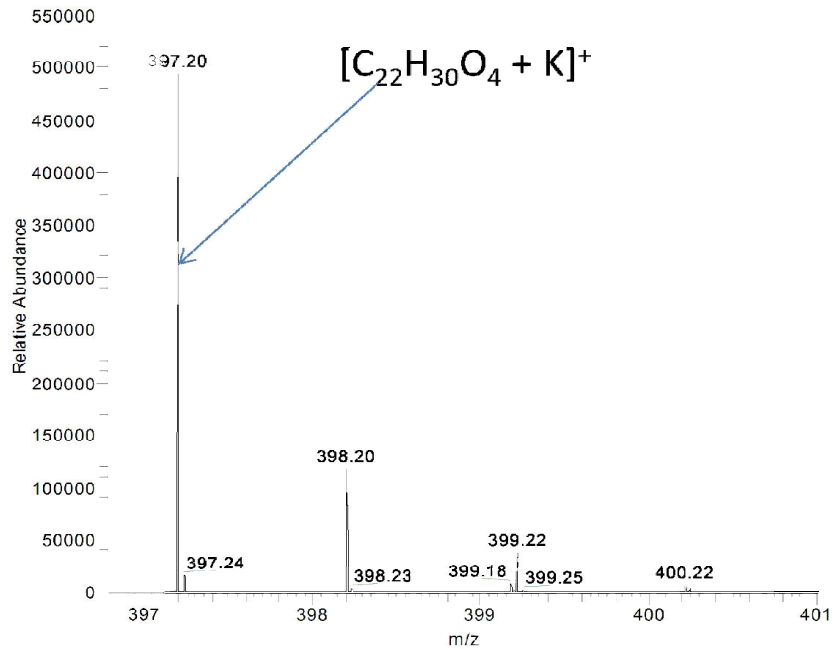


Figure 22: M+H mass spectrum of compound 23

These compounds **23** and **24** were confirmed, by comparison of their spectral data and that in literature, to be Myrsinoic acid B and Myrsinoic acid C, respectively. These compounds had previously been isolated from other plants of the family Myrsinaceae including the genus *Rapanea* and proven to have a range of biological activity. These plants include *Rapanea Umbellate*, *Myrsine seguinii*, *Rapanea ferruginea*, *Myrsine coriacea*, *Myrsine flocculosa* and *Gaballera ferruginea* (Januario *et al.*, 1991; Hirota *et al.*, 2002; Ito *et al.*, 2008; Hess *et al.*, 2010; Cruz *et al.*, 2013). However, these compounds are reported here for the first time from the plant *R. melanophloeos* and shown to have anti-helminthic activity against *S. mansoni*.

4.2 Miracidicidal assay of the methanol crude extract of *T. nobilis*

The *S. mansoni* miracidia were subjected to methanol crude extract in triplicate. The test solutions were prepared as described in section 3.4.2. The percentage mortality of the miracidia after 30 minutes exposure to the various serial dilutions is listed in table 1. IBM SPSS 15.0 software was used to calculate LC₉₀ and LC₅₀ using Log probit regression analysis at 95% confidence level. The negative controls used were 1% DMSO and dechlorinated water which showed no activity against the miracidia after 30 minutes. The positive control was praziquantel that gave a mortality of 100% at 1000 ppm. The results on table 7 shows that there is a positive correlation between the percentage mortality values and the crude extract concentrations. That is, the percentage mortality depends on the concentration of the extract with 1000 ppm having 100% mortality and 50ppm with the lowest mortality of 1.7%. Through log probit regression analysis at 95% confidence level, the crude extract had LC₅₀ and LC₉₀ values as 261.69 and 575.74 ppm, respectively. Thus, it's evident that there are miracidicidal compounds in the methanol crude extract.

These results have correlations with other extracts isolated from other plants. The plants *Zanthoxylum naranjillo* (Braguine *et al.*, 2009) and *Haplophyllum tuberculatum*, both from Rutaceae family have been reported to have reduced egg production by *S. mansoni* worm and reduced shedding of cercariae by *Biomphalaria alexandrina* snails, respectively (Rizk *et al.*, 2012). Mohamed *at al.*, (2005) and Abozeid *et al.*, (2012) also indicated that sativa seeds and *Punica granatum* extracts, respectively, can be used to control *S. mansoni* miracidia. Similarly other plants from different families have also been reported to have schistosomicidal activity at various stages of *S. mansoni* life cycle (Muchika, 2010; Parrakh, 2010; Ismail *et al.*, 2007).

Table 7: Miracidicidal activity of *T. nobilis* methanol crude extract

Conc. (ppm)	Mean mortality (%)	LC ₅₀ (ppm)	LC ₉₀ (ppm)
4000	100 ±0		
2000	100 ±0		
1000	100 ±0		
500	96.7±2.4		
450	81.7±2.4		
400	70±0		
350	61.7±2.4		
300	55±2.4	261.69 (228.79-	575.74 (474.71-
250	43.3±4.7	296.47)	760.53)
200	30±4.1		
150	20±4.1		
100	8.3±2.4		
1% DMSO ^a	0.0±0.0		
Praziquantel	100±0.0		

a- Negative control, b – Positive control

4.3 Miracidicidal assay of *T. nobilis* ethyl acetate and hexane extracts

The *T. nobilis* methanol crude extract was partitioned among water, ethyl acetate and hexane. The ethyl acetate and hexane extracts were subjected to miracidicidal assays and the results tabulated in tables 8 and 9. The LC₅₀ and LC₉₀ were obtained through log probit regression analysis at 95 confidence level and reported in tables 8 and 9.

The ethyl acetate and hexane extracts were active against miracidia but to different extents. From table 8 and 9 it is evident that, the percentage mortality at 500 ppm was 88.3% for ethyl acetate extract and 83.3% for hexane extract. At 50 ppm the ethyl acetate had a mortality of 1.7% while hexane did not show any activity at the same concentration. Similarly, the ethyl acetate had an LC₅₀ of 320.48 while that observed for hexane extract was 334.70. This indicates that ethyl acetate extracts was slightly more active than hexane extract. However, it was evident

that the two extracts do contain compounds that were quite active against miracidia and thus can be used for the control of the same.

Table 8: Miracidicidal activity of *T. nobilis* ethyl acetate crude extract

Conc. (ppm)	Mean % mortality	LC ₅₀ (ppm)	LC ₉₀ (ppm)
4000	100 ± 0		
2000	100 ± 0		
1000	98.3 ± 2.4		
500	88.3 ± 2.4		
450	78.3 ± 4.7		
400	68.3 ± 2.3		
350	55 ± 0		
300	48.3 ± 6.2		
250	33.3 ± 2.4	289.28(252.00-	672.79(547.23-
200	25 ± 0	329.65)	940.74)
150	20 ± 4.1		
100	11.7 ± 2.4		
1% DMSO ^a	0.0 ± 0.0		
Praziquantel (1,000 ppm) ^b	100 ± 0.0		

a- Negative control, b – Positive control

Table 9: Miracidicidal activity of *T. nobilis* hexane crude extract

Conc. (ppm)	Mean % mortality	LC ₅₀ (ppm)	LC ₉₀ (ppm)
4000	100 ±0		
2000	100 ±0		
1000	100 ±0		
500	83.3±6.2		
450	75±4.1		
400	63.3±2.4	307.92(273.05-	638.73(531.23-
350	56.7±2.4	347.23)	854.03)
300	41.7±2.4		
250	33.3±2.		
200	25±4.1		
150	16.7±2.4		
100	1.7±2.4		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

4.4 Miracidicidal assay of fractions from ethyl acetate crude extract

The *T. nobilis* ethyl acetate crude extract was fractionated into four fractions namely; F₁, F₃, F₄ and MeOH fraction through column chromatography on silica gel. These fractions were then subjected to anti-miracidia assay and the results obtained tabulated in tables 10, 11, 12 and 13 respectively. In these tables, also are listed LC₅₀ and LC₉₀ obtained through log probit regression analysis.

All the fractions were potent against the miracidia to different extents. The fraction F₃ was the most active with an LC₅₀ and LC₉₀ of 254.01 and 467.67 ppm, respectively. This was followed by F₄ which had LC₅₀ and LC₉₀ of 324.69 and 491.86 ppm, respectively. The fraction F₁ was less active with LC₅₀ and LC₉₀ of 439.36 and 787.20 ppm, respectively. The MeOH fraction was the least active with an observed LC₅₀ and LC₉₀ of 535.86 and 727.54 ppm, respectively. These four fractions generally do have anti-helminthic compounds active against *S.*

mansoni miracidia as it is clearly seen from tables 10, 11, 12 and 13. However, fractions F₃ and F₄, which recorded the highest mortality of the four fractions, were the ones which were purified further to yield pure compounds **10**, **13**, **21** and **22**, respectively.

Table 10: Miracidicidal activity of *T. nobilis* ethyl acetate F₁

Conc.(ppm)	Mean % mortality	LC ₅₀ (ppm)	LC ₉₀ (ppm)
4000	100 ±0		
2000	100 ±0		
1000	100 ±0		
800	96.7±2.4		
750	88.3±2.4		
700	80±4.1		
650	75±0		
600	71.7±2.4	439.36 (402.89-	787.20 (703.50-
550	68.3±2.4	475.28)	922.10)
500	61.7±2.4		
450	51.7±2.4		
400	45±4.1		
350	28.3±2.4		
300	21.7±2.4		
250	11.7±2.4		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

Table 11: Miracidicidal activity of *T. nobilis* ethyl acetate F₃

Conc.(ppm)	Mean % mortality	LC₅₀	LC₉₀
4000	100 ±0		
2000	100 ±0		
1000	100 ±0		
500	91.7±2.4		
450	85±0		
400	81.7±2.4	227.28 (195.64-	519.78 (434.15-
350	71.7±2.4	258.76)	680.45)
300	65±4.1		
250	51.7±2.4		
200	38.3±2.4		
150	26.7±2.4		
100	13.3±2.4		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

Table 12: Miracidicidal activity of *T. nobilis* ethyl acetate F₄

Conc.(ppm)	Mean % mortality	LC₅₀	LC₉₀
4000	100 ±0		
2000	100 ±0		
1000	100 ±0		
500	91.7±2.4		
450	78.3±2.4		
400	70±0	301.30(270.19-	564.62(482.97-
350	58.3±2.4	335.21)	719.07)
300	48.3±2.4		
250	33.3±2.4		
200	13.3±2.4		
150	8.3±2.4		
100	5±4.1		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

Table 13: Miracidicidal activity of *T. nobilis* ethyl acetate methanol fraction

Conc. (ppm)	Mean % mortality	LC ₅₀ (ppm)	LC ₉₀ (ppm)
4000	100 ±0		
2000	100 ±0		
1000	100 ±0		
750	96.7±2.4		
700	86.7±2.4		
650	73.3±2.4	520.05(489.09-	756.59(693.72-
600	61.7±2.4	551.99)	860.285)
550	50±0		
500	43.3±4.7		
450	33.3±2.4		
400	21.7±2.4		
350	8.3±2.4		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

4.5 Bio-assay of purified compounds isolated from *T. nobilis*

The fractions F₃ and F₄ were purified as described in section 3.3.5. Fraction F₃ yielded a mixture of compounds **10**, **13** and **21** while fraction F₄ yielded compound **22**. These compounds were then subjected to miracidicidal tests in three replicate for each serial dilution. Two negative controls of de-chlorinated water and 1% DMSO were also set up which showed no activity. The results observed are tabulated in tables 14 and 15.

Table 14: Miracidicidal activity of compounds 10, 13 and 21 mixtures

Conc. (ppm)	Mean % mortality	LC ₅₀ (ppm)	LC ₉₀ (ppm)
1000	98.3±2.4		
500	86.7±2.4		
450	78.3±2.4		
400	70.0±0.0	270.18(232.55-	690.93(552.63-
350	61.7±2.4	312.12)	979.49)
300	48.3±2.4		
250	41.7±2.4		
200	30.0±0.0		
150	20.0±4.1		
100	8.3±4.1		
50	3.3±2.4		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

The purified compounds from *T. nobilis* leaves showed activity against miracidia after 30 minutes of exposure. Compound **22**, which had LC₅₀ and LC₉₀ of 287.97 and 631.73 ppm, respectively, had lower activity than the compounds **10**, **13** and **21** mixtures which had LC₅₀ and LC₉₀ of 270.18 and 690.93 ppm, respectively. It was also clear that fraction F₃ had a higher activity than the purified compounds **10**, **13** and **21** mixtures. This may be due to synergistic effects with other compounds within the crude extract.

Table 15: Miracidicidal activity of compound 22

Conc. (ppm)	Mean % mortality	LC ₅₀ (ppm)	LC ₉₀ (ppm)
1000	100±0		
500	88.3±2.4		
450	76.7±2.4		
400	66.7±4.7		
350	56.7±2.4	287.97 (253.01-	631.73 (521.44-
300	46.7±4.7	326.61)	853.51)
250	41.7±2.4		
200	28.3±2.4		
150	16.7±6.2		
100	3.3±2.4		
50	1.7±2.4		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

To the best of my knowledge there is no reported study on the evaluation of miracidicidal activity of isolated alkaloids. However, the four alkaloids isolated, that is, compounds **10**, **13**, **21** and **22** had schistosomicidal activity and comparisons of their LC values showed excellent toxicities against *S. mansoni* miracidia. The miracidia mortality of the compounds had dose dependent effect, that is, the activity was proportional to the concentration. There are few reported studies of miracidicidal activity of isolated pure compounds on *S. mansoni* miracidia. However, alkaloids both synthetic and from natural sources have been shown to be excellent anti-helminthics. For instance, Glycoalkaloids from *Solanum* spp have been reported to have *in vitro* anti-schistosomal effects. Similarly, epiisopiloturine and Imidazole alkaloids from *Pilocarpus microphyllus* (Rutaceae) leaves have been reported to have *in vitro* effect on the survival time of *S.mansoni* Schistosomulae and adult worm stages (Rai and Kon , 2013). Moreover, the drugs Praziquantel and Triclabendazole (figure 23) are alkaloids which are known to cause tegumental damage and paralysis of the trematodes (How, 2007). According to Satou *et al.* (2002a), the isoquinoline alkaloids; Allocryptopine, Dehydrocaryodoline and Papaverine have

been shown to cause larval mobility inhibition against the helminth *Toxocara canis*. The alkaloids Protopine, d-Corydoline and l-Stylophine have also been reported to have anthelmintic activity against *Strongyloides ratti* and *Strongyloides venezuelensis* (Satou *et al.*, 2002b). Satou *et al.* (2005) also reported that β -Carboline alkaloids isolated from the plants *Picrasma quassoides* and *Ailanthus altissima* exhibit larval toxicity against *Toxocara canis*.

4.6 Miracidicidal assay of *R. melanoploeos* methanol, ethyl acetate and hexane crude extracts

The methanol crude extract was subjected to solvent-solvent partitioning after suspension in water to yield ethyl acetate, hexane and methanol/water fractions. Serial dilutions of methanol, ethyl acetate and hexane extracts were then subjected to *S. mansoni* bio-assay in three replicates for each concentration as described in section 3.4. The percentage mortality was recorded after 30 minutes of exposure as tabulated in tables 16, 17 and 18. The LC₅₀ and LC₉₀ were then calculated through log probit regression at 95% confidence level using IBM SPSS 11.5 software. Dechlorinated water and 1% DMSO which were used as negative controls showed 0% mortality. Praziquantel, the positive control, showed 100% mortality at 1000ppm.

From tables 16, 17 and 18, it is clearly evident that the methanol, ethyl acetate and hexane extracts were all active against *S. mansoni* miracidia to different extents. At concentrations of 300 ppm, 500 ppm and 400 ppm the methanol, ethyl acetate and hexane extracts had 100% mortality respectively. This is a clear indication that the methanol extract was the most active followed by hexane and lastly ethyl acetate extract. This was further confirmed by the LC₅₀ and LC₉₀. The methanol, ethyl acetate and hexane extracts had LC₅₀ of 96.57, 242.05 and 150.73 ppm, respectively. Similarly, the LC₉₀ were calculated to be 257.98, 558.43 and 327.73 for methanol, ethyl acetate and hexane respectively.

Table 16: Miracidicidal activity of *R. melanophloeos* methanol crude extract

Conc. (ppm)	Mean % mortality	LC ₅₀ (ppm)	LC ₉₀ (ppm)
1000	100 ±0		
500	100 ±0		
400	100 ±0		
350	100 ±0		
300	100 ±0		
250	98.3±2.9		
200	81.7±2.9	96.57(79.73-	257.98(197.49-
150	66.7±7.6	116.83)	394.76)
100	48.3±7.6		
75	33.3±7.6		
50	15±5		
25	8.3±2.9		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

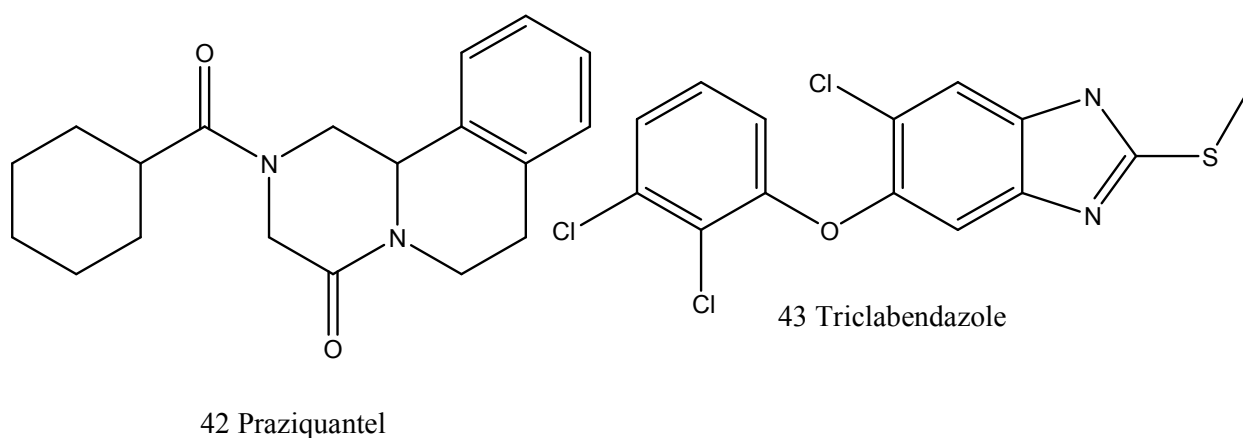


Figure 23: Structures of praziquantel and triclabendazole

Table 17: Miracidicidal activity of *R. melanophloeos* ethyl acetate crude extract

Conc. (ppm)	Mean % mortality	LC₅₀ (ppm)	LC₉₀ (ppm)
4000	100 ±0		
2000	100 ±0		
1000	100 ±0		
500	100 ±0		
450	88.3±2.9		
400	75±5		
350	65±5	242.05(209.68-	558.43(462.89-
300	51.7±5.9	275.89)	740.30)
250	43.3±5.8		
200	35±5		
150	20±5		
100	15±5		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

Table 18: Miracidicidal activity of *R.melanophloeos* hexane crude extract

Conc. (ppm)	Mean % mortality	LC ₅₀ (ppm)	LC ₉₀ (ppm)
4000	100 ±0		
2000	100 ±0		
1000	100 ±0		
500	100 ±0		
450	100 ±0		
400	100 ±0		
350	95±5	150.73 (126.77-	327.73 (279.03-
300	88.3±7.6	173.61)	408.96)
250	70±5		
200	53.3±2.9		
150	43.3±7.6		
100	23.3±2.9		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

It is therefore evident that the extracts do contain anti-helminthic compounds with anti-schistosomal activity against *S. mansoni* miracidia. However, the hexane extract was not fractionated to isolate the compounds because of unfavourable TLC patterns. From the ethyl acetate extract two compounds with activity against miracidia were isolated. The anti-helminthic activity shown by this plant is in agreement with the literature according to Githiori (2002) who reported that the plant *R. melanophloeos* shows some in vivo activity against the helminth *Haemonchus contortus* in sheep. Other plants from the family Myrsinaceae such as *Embelia schimperi* have been reported to have in vitro anti-helminthic activity thus confirming the said activity (Bogh *et al.*, 1996).

4.7 Bio-assay of fractions from ethyl acetate extract

The *R. melanophloeos* ethyl acetate crude extract was subjected to column chromatography to yield four fractions which were then tested for schistosomicidal activity against *S. mansoni* miracidia. The percentage mortality, LC₅₀ and LC₉₀ (at 95% confidence level) observed are tabulated in tables 19-22.

Table 19: Miracidicidal activity of *R. melanophloeos* ethyl acetate F₁

Conc. (ppm)	Mean % mortality	LC ₅₀ (ppm)	LC ₉₀ (ppm)
4000	100 ±0		
2000	100 ±0		
1000	100 ±0		
500	100 ±0		
450	83.3±7.6		
400	66.7±5.8		
350	55±5		
300	48.3±2.9	283.17 (219.34-	618.42 (456.57-
250	28.3±7.6	362.26)	1219.19)
200	21.7±2.9		
150	13.3±2.9		
100	6.7±7.6		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

All the four fractions did show activity against miracidia as indicated in tables 19-22. The fractions F₁, F₄ and methanol showed 100% mortality at a concentration of 500 ppm while F₃ had 100% mortality at 400 ppm. The LC₅₀ was calculated to be 283.17, 182.28, 238.93 and 215.93 for fractions F₁, F₃, F₄ and methanol, respectively. Similarly, F₁, F₃, F₄ and methanol fractions had LC₉₀ of 618.42, 344.42, 472.12 and 451.28 ppm, respectively. This indicates that F₃ was the most active followed by methanol, then F₄ and lastly F₁.

Table 20: Miracidicidal activity of *R. melanophloeos* ethyl acetate F₃

Conc. (ppm)	Mean % mortality	LC₅₀ (ppm)	LC₉₀ (ppm)
4000	100 ±0		
2000	100 ±0		
1000	100 ±0		
500	100 ±0		
450	100 ±0		
400	100 ±0		
350	91.7±7.6		
300	78.3±2.9	182.28 (158.19-	344.42 (300.12-
250	61.7±2.9	205.00)	417.85)
200	50±5		
150	36.7±7.6		
100	16.7±7.6		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

Table 21: Miracidicidal activity of *R. melanophloeos* ethyl acetate F₄

Conc. (ppm)	Mean % mortality	LC₅₀ (ppm)	LC₉₀ (ppm)
4000	100 ±0		
2000	100 ±0		
1000	100 ±0		
500	100 ±0		
450	96.7 ±5.8		
400	86.7±7.6		
350	65±5	238.93 (210.48-	472.12 (405.09-
300	55±5	267.53)	592.19)
250	46.7±2.9		
200	33.3±7.6		
150	21.7±7.6		
100	8.3±2.9		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

Table 22: Miracidicidal activity of *R. melanophloeos* ethyl acetate methanol fraction

Conc. (ppm)	Mean % mortality	LC ₅₀ (ppm)	LC ₉₀ (ppm)
4000	100 ±0		
2000	100 ±0		
1000	100 ±0		
500	100 ±0		
450	93.3±7.6		
400	86.7±7.6		
350	73.3±2.9	215.95 (187.28-	451.28 (384.80-
300	61.7±5.8	243.88)	570.65)
250	55±5		
200	43.3±5.8		
150	28.3±7.6		
100	15±5		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

4.8 Bio-assay of purified compounds isolated from *R. melanophloeos*

Fraction F₃ was subjected to PTLC where it yielded two pure compounds. On subjecting these compounds to *S. mansoni* miracidia activity, all the compounds were found to be active. The percentage mortalities, LC₅₀ and LC₉₀ calculated are tabulated in Tables 23 and 24.

The compounds were found to cause 100% mortality for **23** and **24** at the concentrations of 350 ppm and 300ppm respectively. The LC₅₀ at 95% confidence level, were calculated to be 139.89 and 98.06 for **23** and **24**, respectively. The LC₉₀ at 95% confidence level was similarly found to be 314.23 and 236.52 for compounds **23** and **24**, respectively. From this, it is clear that compound **24** was the most active, followed by compound **23**. Compound **24** was contaminated with compound **23** and synergism of the two compounds may have been the cause of the observed high activity in compound **24**. However, the activity of all of them compared favourably well with those of commercial Praziquantel which was used as the reference standard.

Table 23: Miracidicidal activity of compound 23

Conc. (ppm)	Mean % mortality	LC₅₀ (ppm)	LC₉₀ (ppm)
1000	100±0		
500	100±0		
450	100±0		
400	100±0		
350	100±0		
300	95±5		
250	71.7±7.6	139.89 (111.35-	314.23 (250.00-
200	53.3±7.6	170.07)	445.02)
150	41.7±7.6		
100	28.3±7.6		
75	16.7±2.9		
50	10±5		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

Table 24: Miracidicidal activity of compound 24

Conc. (ppm)	Mean % mortality	LC ₅₀ (ppm)	LC ₉₀ (ppm)
1000	100±0		
500	100±0		
450	100±0		
400	100±0		
350	100±0		
300	100±0		
250	95±5	98.06 (81.92-	236.51 (197.25-
200	71.7±7.6	114.55)	301.31)
150	58.3±2.8		
100	48.3±2.8		
75	36.7±7.6		
50	16.7±7.6		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control, b – Positive control

The comparisons of the LC values of the two compounds **23** and **24** showed excellent toxicities against *S. mansoni* miracidia. The miracidia mortality of the compounds had dose dependent effect, that is, the activity was proportional to the concentration. There are few reported studies of miracidicidal activity of isolated pure compounds on *S. mansoni* miracida. Tannins extracted from *Punica granatum* showed high activity against this miracidia with 100% mortality being registered with concentrations as low as 0.39 and 50 ppm within 50-150 minutes and 5-15 minutes respectively (Abozeid *et al.*, 2012). The methanolic extract of *Pleclanthus tenuiflorus* has also been shown to have an LC₅₀ of 24.37 mg/100ml on *S. mansoni* miracidia (Abdel-Aziz *et al.*, 2011). Aqueous extracts from *Phytolacca dodecandra* at a concentration of 4 ppm was determined to prevent *S. mansoni* miracidia from infecting snails (Birrie *et al.*, 1998) and also shown to have an LC₅₀ of 8.2 ppm (Madhina and Shiff, 1996). The plant *Iris pseudacorus* leaves extract at a concentration as low as 0.9mg/l have been shown to have miracidicidal effects (Ahmed and El Hamshary, 2005). *Nigella sativa* crushed seeds at a

concentration of 4ppm were reported to have miracidicidal activity within one minutes against *S. mansoni* miracidia (Azza *et al.*, 2005). A concentration of 400ppm extracts from *Tetrapleura tetraptera* have been shown to have lethal effects against *S. haematobium*, *S. bovis* and *S. mansoni* miracidia (Aladesanmi, 2007). Saponins isolated from *Furcraea selloea* were shown to kill 100% of miracidia at 50 ppm with the saponin 3-O- β -D-glucopyranosyl-(1 \rightarrow 3)- β -D-glucopyranosyl-(1-3)- β -D-xylopyranoside gloriogenin causing a 100% mortality at a concentration of 6ppm (El-Nahas *et al.*, 2005). Isoflavonoids isolated from *Milletis thoniningii* have also been shown to have miracidicidal activity (Lyddiard *et al.*, 2002). Eissa *et al.* (2011), also reported that miltefosine, the oral drug licensed for the treatment of Leishmaniasis, has lethal effects against both *S. mansoni* and *S. haematobium* miracidia.

4.9 Anti-miracidia assay of *T. nobilis* essential oil

The percentage yield of the essential oil obtained from *T. nobilis* leaves was calculated to be 0.33% v/w with the density determined to be 0.99 g/ml. The oil anti-helminthic activity against *S. mansoni* miracidia was evaluated and the results observed tabulated in table 25.

At a concentration of 400 ppm, the oil was found to cause 100% miracidia mortality within 30 minutes. Decreasing the concentration to 350 ppm induced 93.3% mortality. At 50 ppm, the lowest concentration tested, an observed mortality of 5% was recorded. Through log probit regression analysis, at 95% confidence level, the LC₅₀ and LC₉₀ were determined to be 196.29 and 367.24 respectively after 30 minutes of exposure. Therefore, it is evident that the constituents of the oil have anti-helminthic compounds active against *S. mansoni* miracidia.

The activity of the essential oil was in strong agreement with other plants essential oils that have been shown to have schistosomicidal activity. Rai and Kon (2013), reported that essential oils from *Baccharis dracunculifolia* had *in vitro* activity against *S. mansoni* at a concentration of 10 μ g/ml while those of *Bidens pilosa* and *Tagetes erecta* at a concentration of 100 μ g/ml. *Piper cubeba* and *Ageratum conyzoides* have also been reported to be active against adult *S. mansoni* worms. Furthermore, essential oils from *Apium graveolens*, *Piper cubeba*, *Piper marginatum*, *Eucalyptus* spp and *Zingiber officinale* have been reported to be active against *S. mansoni* cercariae (Rai and Kon, 2013).

Table 25: Miracidicidal activity of *T. nobilis* essential oil

Conc. (ppm)	Mean % mortality	LC ₅₀ (ppm)	LC ₉₀ (ppm)
4000	100 ±0		
2000	100 ±0		
1000	100 ±0		
500	100 ±0		
450	100 ±0		
400	100 ±0		
350	93.3±2.9	196.29 (153.12-	367.24 (296.12-
300	75±5	237.58)	541.19)
250	56.7±7.6		
200	46.7±5		
150	21.7±7.6		
100	11.7±2.9		
1% DMSO ^a	0.0±0.0		
Praziquantel (1,000 ppm) ^b	100±0.0		

a- Negative control b- Positive control

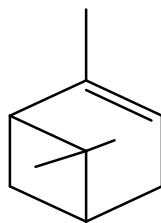
4.10 Determination of the chemical composition of the oil extract

The chemical composition of the essential oil extracted from *T. nobilis* was determined by the GC-MS. The identification of the compounds in the oil was determined by comparing the electron impact mass spectrum of the compounds in the oil and those in the Wiley7N.1, FLAVORS.L and HPCH1607.L computer library databases. From the data obtained, it was revealed that monoterpenes (42.21%) and sesquiterpenes (33.09%) are the main constituents of the oil. The major monoterpenes were found to be β - Ocimene, γ - Terpinene, α - Pinene, Limonene and Methyl Eugenol and with concentrations of 10.15, 6.11, 3.95, 3.34 and 2.68%, respectively. Sesquiterpenes β - Cadinene, 1,6-Germacradien-5-ol, α -Amorphine, tau-Cadinol and Germacrene D having concentrations of 4.98, 4.38, 3.96, 3.56 and 3.06%, respectively were determined to be the major ones.

Table 26: Major constituents of *T. nobilis* essential oil

Compound Number	R.T (min)	Compound Name	%Concentration	Detection method
25	6.81	α – Pinene	3.95	GC/MS
26	8.01	γ – Terpinene	6.11	GC/MS
27	8.45	β – Myrcene	2.17	GC/MS
28	9.50	Limonene	3.34	GC/MS
29	10.29	β – Ocimene	10.15	GC/MS
30	11.65	Linalool	1.13	GC/MS
31	12.55	Neo-allo-Ocimene	3.68	GC/MS
32	20.21	Methyl Eugenol	2.68	GC/MS
33	23.89	Elemicin	2.89	GC/MS
34	19.59	β – Gurjunene	1.26	GC/MS
35	20.44	Germacrene D	3.06	GC/MS
36	22.39	α – Amorphine	3.96	GC/MS
37	23.08	β – Cadinene	4.98	GC/MS
38	23.16	δ – Cadinene	1.28	GC/MS
39	24.84	Guaiol	1.01	GC/MS
40	25.88	Tau- Cadinol	3.56	GC/MS
41	26.25	α – Cadinol	2.57	GC/MS

The compound (**25**) was observed to occur at 6.81 minutes retention time and was identified as α -Pinene. The compound has a molecular weight of 136amu and constituted 3.95% of the total oil. Analysis of the MS spectrum showed the presence of a small peak at m/z 136 corresponding to $[C_{10}H_{16}]^+$. Another peak at m/z 121 was observed which corresponded with $[M-CH_3]^+$. The peaks at m/z 105, 93 and 77 were corresponding to $[M-C_2H_5]^+$, $[M-H-C_3 H_7]^+$ and $[M+H-C_4 H_{10}]^+$, respectively. The structure of the compound and its MS spectra are shown in figure 24.



25

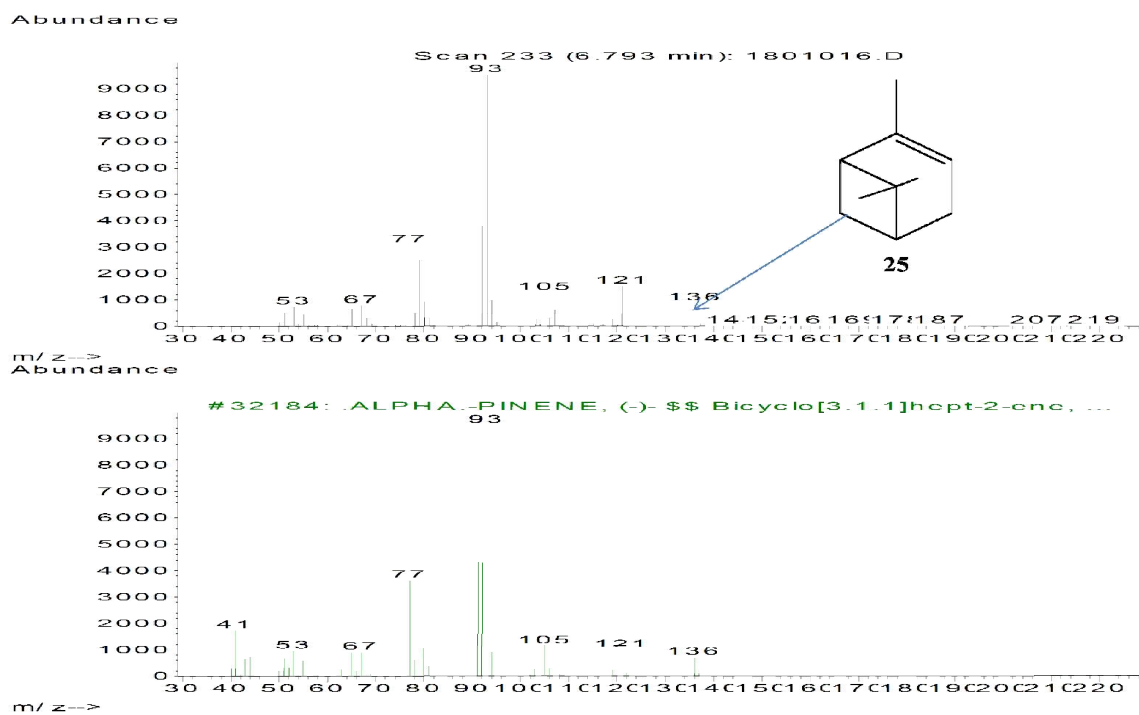


Figure 24: Mass spectra of compound 25

This compound has been reported to be the major constituent in the essential oil of *Plectranthus neochilus*. This oil was reported to have anthelmintic properties against *S. mansoni* adult worms causing 100% mortality at a concentration of 0.1mg/ml. It also reduces hatching of *S.mansoni* eggs (Caixeta *et al.*, 2011). Its isomer β -pinene is a major constituent of *Piper cubeba* essential oil which was reported to cause mortality of *S. mansoni* adult worms at a concentration of 0.13mg/ml within 120 h (De Oliveira *et al.*, 2012). Therefore, even though the compounds' individual activity has not been reported, its effect against *S. mansoni* miracidia may not be ruled out.

The compound (**28**) occurred at the retention time 9.50 and was identified as limonene with molecular weight of 136 and was 3.34% of the total oil. The GC-MS spectrum of the oil showed a peak at m/z 121 corresponding to the fragment $[C_9H_{13}]^+$ after the loss of a methyl group. The peaks at m/z 68 $[C_5H_8]^+$ and at m/z 67 $[C_5H_7]^+$ are due to a retro Diels Alder reaction the compound undergoes. The peaks at m/z 107, 93 and 79 corresponds to $[C_8H_{13}]^+$, $[C_7H_9]^+$, and $[C_6H_7]^+$ respectively. Limonene is a monoterpene as it has ten carbon atoms. The compound has not been reported to show antischistosomal activity or any anthelmintic activity. However, (+) Limonene epoxide isomers has been shown to have similar effects to those of praziquantel at 25 μ g/ml on *S. mansoni* adult worms with reduction in motility and death after 120 h (Moraes *et al.*, 2013). The compound has been reported to be a major constituent of essential oils from the plants *Eucalyptus citridora*, *Eucalyptus staigeriana* and *Eucalyptus globules*. These oils have been shown to have antihelmintic activity against *Haemonchus contortus* by reduction of egg hatching and impairing larval development (Macedo *et al.*, 2009; Macedo *et al.*, 2010; Macedo *et al.*, 2011). The structure and the mass spectra for Limonene are shown in figure 25 below.

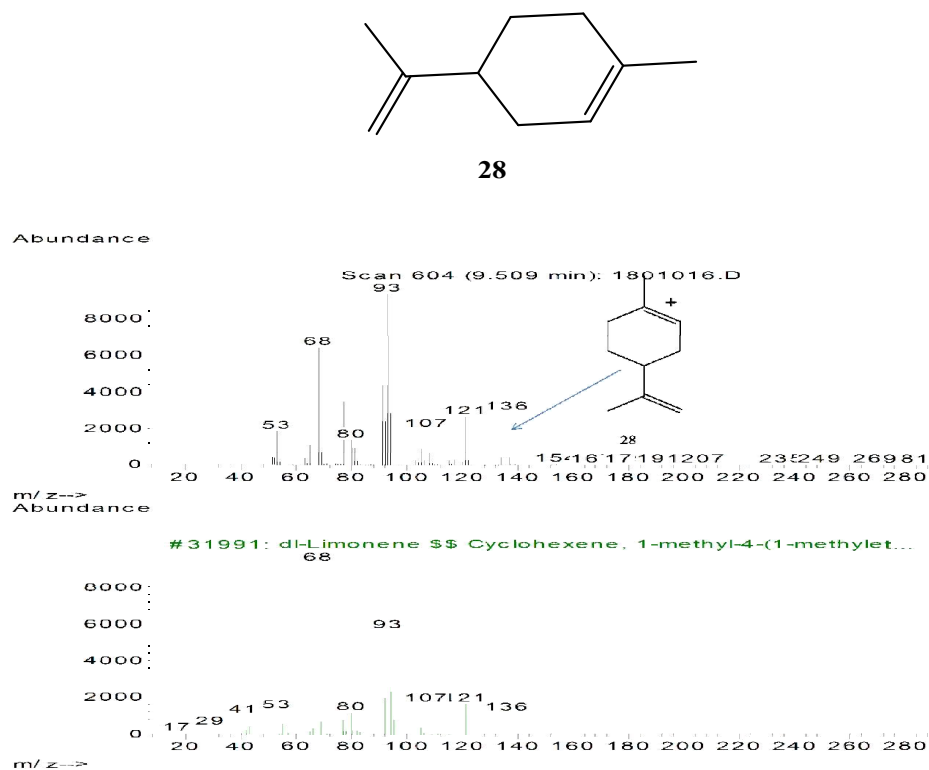


Figure 25: Mass spectra of compound 28

Ocimene isomers were found to be the major constituents of *T. nobilis* oil, with β -Ocimene constituting 10.15% of the total oil. β -Ocimene (29) was observed at retention time 10.29 minutes and identified to have a molecular weight of 136. Analysis of the MS spectrum showed the presence of a small peak at m/z 136 corresponding to $[C_{10}H_{16}]^+$. Another peak at m/z 121 was observed which corresponded with $[M-CH_3]^+$. The peaks at m/z 105, 93 and 79 were corresponding to $[M-C_2H_5]^+$, $[M-H-C_3 H_7]^+$ and $[M+H-C_4 H_8]^+$, respectively. The structure of the compound and its MS spectra are shown below. Compound (29) has not been reported to have anthelmintic activities. However, its activity may not be ruled out.

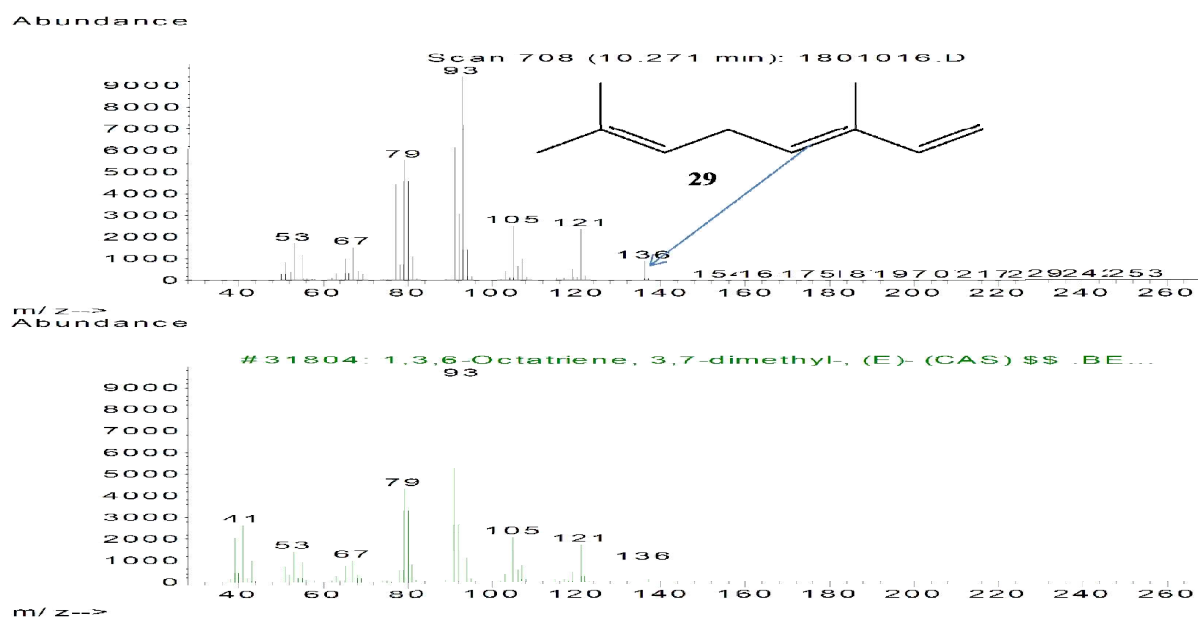
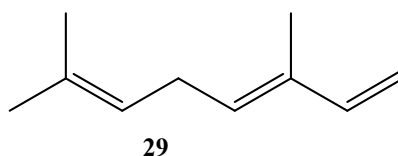
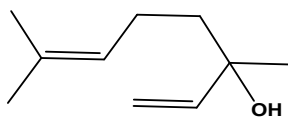


Figure 26: Mass spectra of compound 29

Compound (30) occurring at retention time 11.65 was identified as Linalool constituting 1.13% of the total oil. The compound was identified with molecular weight of 154. The molecular ion peak m/z 154 was observed in both spectra though the peaks were very small. The

peaks at m/z 136, 121, 107, 93, 71 and 41 were observed. These peaks are shown in figure 27 below and correspond to fragments shown as per the scheme in figure 28 below.



Dearth of literature is available on the activity of Linalool against helminthes. However, it has been reported to reduce the infection of *S. mansoni* miracidia (Ahmed, 2006). The activity of the oil observed may not be solely attributed to Linalool.

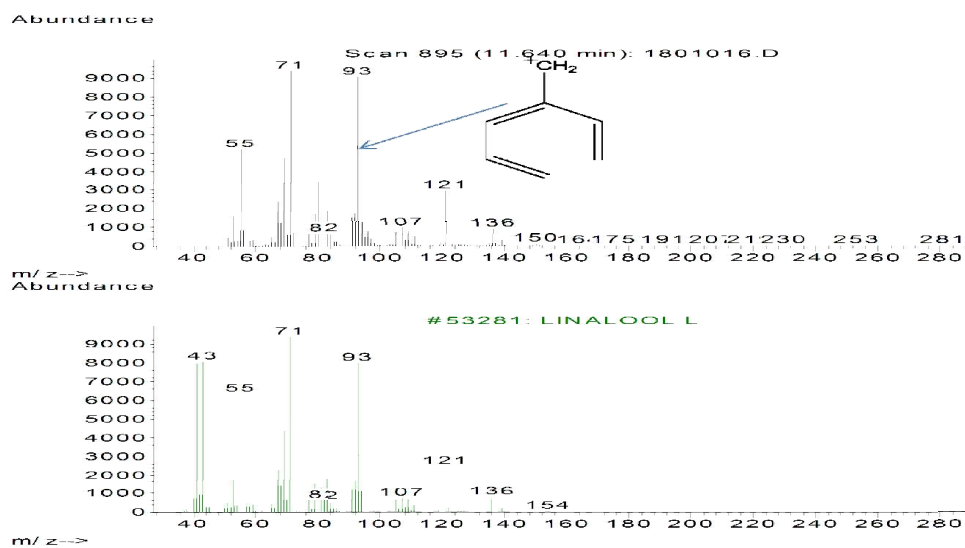


Figure 27: Mass spectra of compound 30

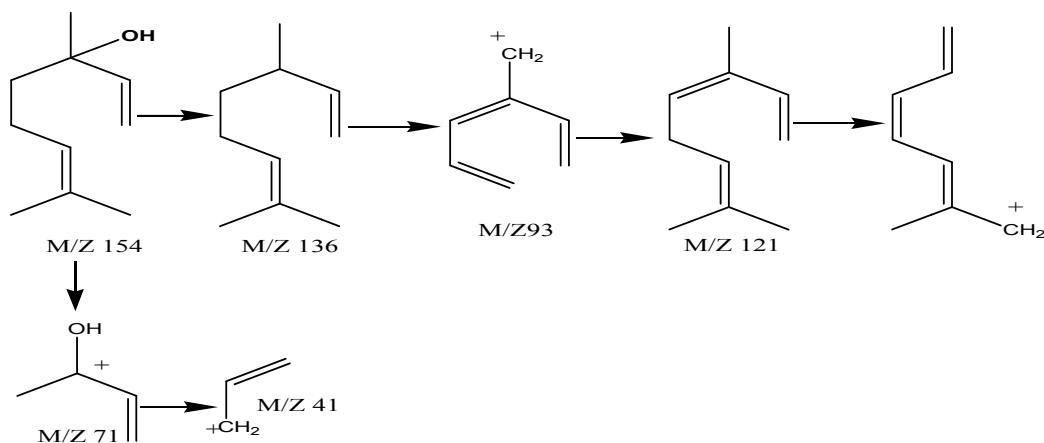


Figure 28: Scheme showing fragmentation pattern of compound 30

The compound (**32**) was identified as methyl eugenol. It had a concentration of 2.68% and appeared at 20.21 retention time. It has a molecular weight of 78 amu. The analysis of the ms spectra showed the presence of a strong peak at m/z 178 corresponding to $[C_{11}H_{14}O_2]^+$. There was another peak at m/z 163 corresponding to the fragment $[C_{10}H_{11}O_2]^+$ after loss of a methyl group. The fragment $[C_{10}H_{11}O]^+$ which occurred after loss of a methoxy group showed a peak at m/z 147. The spectra (figure 29) and structure of the molecule are shown below.

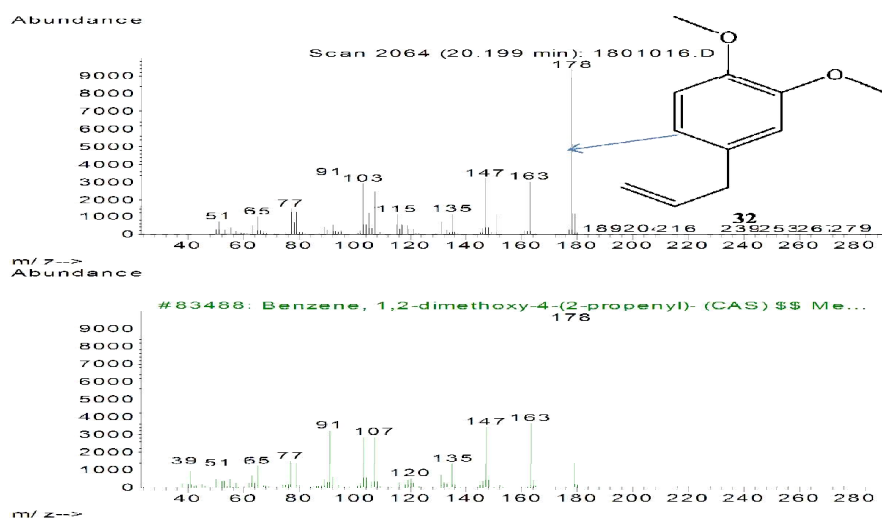
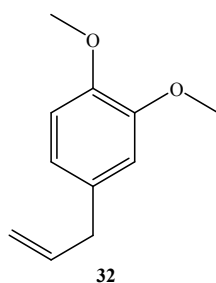


Figure 29: Mass spectra of compound 32

The compound Eugenol which belongs to the same class of compounds as compound **32** has been reported to have lethal effects against cercariae and miracidia larval stages of *S. mansoni* (Mansour *et al.*, 2003). It has also been shown to reduce the infection of snails by *S. mansoni* miracidia (Ahmed, 2006). Eugenol has also been reported to be the major constituent of

Piper longum essential oil which has been shown to have antihelmintic activity against *Fasciola gigantica* by causing paralysis of the worms (Singh *et al.*, 2009). Therefore, compound **32** may have contributed to the activity shown by the oil.

The compounds **26**, **27**, **31**, **33**, **34**, **35**, **36**, **37**, **38**, **39**, **40** and **41** shown in figure 30 were also found in appreciable amount each constituting more than one percent of the total oil. None of these compounds however have been reported to show anti-helmintic activity. Therefore, their individual anthelmintic activity needs evaluation.

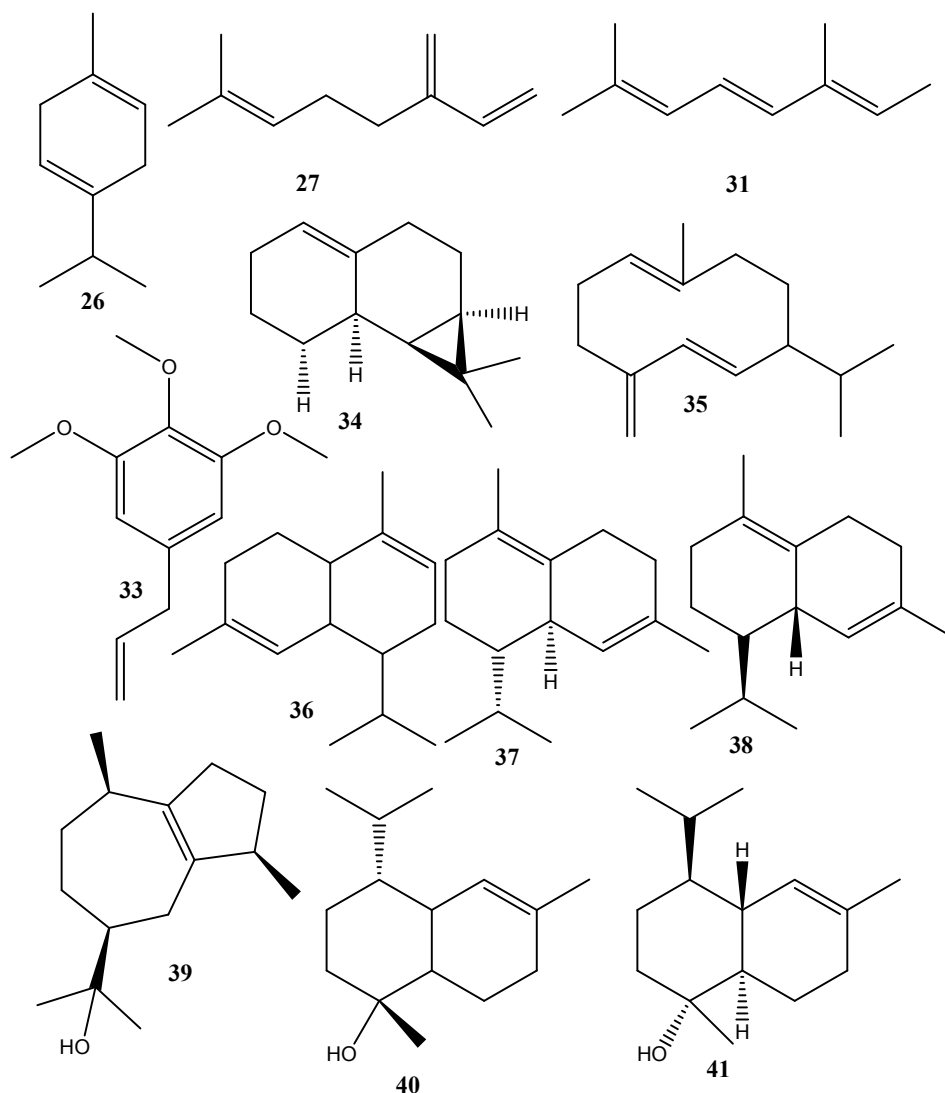


Figure 30: Some major compounds of *T. nobilis* essential oil

CHAPTER FIVE

CONCLUSIONS AND RECOMMENDATIONS

5.1 Conclusion

The essential oil from *T. nobilis* was determined to be majorly constituted of monoterpenes (42.21%) and sesquiterpenes (33.09%). The major compounds were β -Ocimene, γ -Terpinene, β -Cadinene and 1,6-Germacradien-5-ol which had concentrations of 10.15%, 6.11%, 4.98% and 4.38% respectively. The essential oil showed lethal anti-helminthic effects against *S. mansoni* miracidia with recorded LC₅₀ and LC₉₀ values of 196.29 ppm and 367.24 ppm respectively.

The non-volatiles of both *T. nobilis* and *R. melanophloeos* were active against *S. mansoni* miracidia. The *R. melanophloeos* methanol extract was the most active with LC₅₀ and LC₉₀ values of 96.57 and 257.98 ppm, respectively. The *T. nobilis* methanol extract recorded LC₅₀ and LC₉₀ values of 261.69 and 575.74 ppm, respectively. This indicated that the two plants do contain secondary metabolites that contributed to this activity.

Fractionation of *T. nobilis* and *R. melanophloeos* leaf extracts over silica gel column chromatography yielded six compounds which were successfully identified through analysis of their NMR and MS data as well as comparison with literature data. Of the six compounds, *T. nobilis* had four furoquinoline alkaloids; Tecleoxine **10**, Methylnkolbisine **13**, Kokusagine **21** and Nkolbisine **22** while the *R. melanophloeos* had two benzoic acid derivatives Myrsinoic acid B **23** and Myrsinoic acid C **24**. All these compounds were active against *S. mansoni* miracidia. Compound **23** which recorded mortality LC₅₀ and LC₉₀ values of 98.06 and 236.51 ppm, respectively was the most potent followed by compound **24** which registered LC₅₀ and LC₉₀ values of 139.89 and 314.23 ppm, respectively. The compounds **10**, **13** and **21** mixture registered LC₅₀ and LC₉₀ values of 270.18 and 690.93 ppm, respectively and compound **22** which recorded LC₅₀ and LC₉₀ values of 287.97 and 631.733 ppm, respectively was the least potent.

Results from this study shows that the four furoquinoline alkaloids and the essential oil from *T. nobilis* leaves as well as the two Myrsinoic acid compounds from *R. melanophloeos* have potential applicability in the control of *S. mansoni* miracidia. This study underscores the fact that bioactive plant constituents can be useful agents in the control of Schistosomiasis. These findings

also show that these compounds can be used as lead compounds in the development of new, biodegradable, environmentally benign and more potent anti-helminthics.

5.2 Recommendations

In relation to this study, the following recommendations were made;

- i. More advanced methods of compound purification such as use of HPLC, be used to purify compounds **10**, **13** and **21** and separate the three alkaloids and their individual miracidicidal activity tested.
- ii. Toxicity tests to be carried out for the alkaloids, myrsinoic acids and the essential oil.
- iii. The mode of action of each of the pure compounds and that of the essential oil to be investigated.
- iv. The components of the essential oil to be isolated and each tested for its miracidicidal activity.

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APPENDICES

Appendix 1: Generated LC values for *Teclea nobilis* methanol crude extract

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	62.544	37.114	86.206	1.796	1.570	1.936
.020	73.965	46.311	98.815	1.869	1.666	1.995
.030	82.269	53.277	107.787	1.915	1.727	2.033
.040	89.126	59.189	115.093	1.950	1.772	2.061
.050	95.122	64.469	121.417	1.978	1.809	2.084
.060	100.543	69.324	127.088	2.002	1.841	2.104
.070	105.549	73.873	132.292	2.023	1.868	2.122
.080	110.243	78.191	137.146	2.042	1.893	2.137
.090	114.693	82.330	141.726	2.060	1.916	2.151
.100	118.948	86.326	146.090	2.075	1.936	2.165
.150	138.310	104.937	165.800	2.141	2.021	2.220
.200	155.922	122.359	183.633	2.193	2.088	2.264
.250	172.809	139.370	200.772	2.238	2.144	2.303
.300	189.527	156.383	217.899	2.278	2.194	2.338
.350	206.458	173.657	235.531	2.315	2.240	2.372
.400	223.922	191.380	254.148	2.350	2.282	2.405
.450	242.222	209.708	274.263	2.384	2.322	2.438
<u>.500</u>	<u>261.693</u>	<u>228.791</u>	<u>296.475</u>	<u>2.418</u>	<u>2.359</u>	<u>2.472</u>
.550	282.728	248.813	321.513	2.451	2.396	2.507
.600	305.835	270.037	350.306	2.485	2.431	2.544
.650	331.704	292.873	384.091	2.521	2.467	2.584
.700	361.337	317.969	424.642	2.558	2.502	2.628
.750	396.294	346.373	474.719	2.598	2.540	2.676
.800	439.213	379.861	539.070	2.643	2.580	2.732
.850	495.142	421.781	626.979	2.695	2.625	2.797
<u>.900</u>	<u>575.739</u>	<u>479.707</u>	<u>760.528</u>	<u>2.760</u>	<u>2.681</u>	<u>2.881</u>
.910	597.097	494.670	797.132	2.776	2.694	2.902
.920	621.199	511.388	839.005	2.793	2.709	2.924
.930	648.824	530.349	887.714	2.812	2.725	2.948
.940	681.132	552.273	945.610	2.833	2.742	2.976
.950	719.948	578.287	1016.439	2.857	2.762	3.007
.960	768.386	610.294	1106.686	2.886	2.786	3.044
.970	832.424	651.914	1229.002	2.920	2.814	3.090
.980	925.889	711.412	1413.288	2.967	2.852	3.150
.990	1094.960	815.879	1762.568	3.039	2.912	3.246

a Logarithm base = 10.

Appendix 2: Generated LC values for *Teclea nobilis* hexane extract

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	81.887	50.321	109.783	1.913	1.702	2.041
.020	95.635	61.935	124.456	1.981	1.792	2.095
.030	105.532	70.635	134.809	2.023	1.849	2.130
.040	113.646	77.959	143.192	2.056	1.892	2.156
.050	120.704	84.460	150.419	2.082	1.927	2.177
.060	127.056	90.406	156.877	2.104	1.956	2.196
.070	132.900	95.952	162.786	2.124	1.982	2.212
.080	138.360	101.195	168.284	2.141	2.005	2.226
.090	143.521	106.202	173.461	2.157	2.026	2.239
.100	148.441	111.019	178.384	2.172	2.045	2.251
.150	170.672	133.248	200.529	2.232	2.125	2.302
.200	190.692	153.768	220.474	2.280	2.187	2.343
.250	209.729	173.545	239.611	2.322	2.239	2.380
.300	228.438	193.064	258.744	2.359	2.286	2.413
.350	247.263	212.605	278.488	2.393	2.328	2.445
.400	266.557	232.354	299.411	2.426	2.366	2.476
.450	286.657	252.454	322.105	2.457	2.402	2.508
<u>.500</u>	<u>307.917</u>	<u>273.054</u>	<u>347.233</u>	<u>2.488</u>	<u>2.436</u>	<u>2.541</u>
.550	330.754	294.355	375.566	2.520	2.469	2.575
.600	355.695	316.666	408.054	2.551	2.501	2.611
.650	383.451	340.456	445.959	2.584	2.532	2.649
.700	415.049	366.435	491.099	2.618	2.564	2.691
.750	452.075	395.696	546.332	2.655	2.597	2.737
.800	497.206	430.041	616.601	2.697	2.634	2.790
.850	555.528	472.813	711.558	2.745	2.675	2.852
<u>.900</u>	<u>638.726</u>	<u>531.509</u>	<u>854.029</u>	<u>2.805</u>	<u>2.726</u>	<u>2.931</u>
.910	660.623	546.594	892.767	2.820	2.738	2.951
.920	685.262	563.411	936.937	2.836	2.751	2.972
.930	713.416	582.438	988.135	2.853	2.765	2.995
.940	746.229	604.377	1048.754	2.873	2.781	3.021
.950	785.499	630.326	1122.590	2.895	2.800	3.050
.960	834.285	662.136	1216.199	2.921	2.821	3.085
.970	898.432	703.310	1342.310	2.953	2.847	3.128
.980	991.404	761.820	1530.854	2.996	2.882	3.185
.990	1157.855	863.639	1884.175	3.064	2.936	3.275

a Logarithm base = 10.

Appendix 3: Generated LC values for *Teclea nobilis* ethyl acetate extract

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
LOGIT .010	49.512	24.674	74.215	1.695	1.392	1.870
.020	64.869	35.681	92.017	1.812	1.552	1.964
.030	76.102	44.353	104.537	1.881	1.647	2.019
.040	85.333	51.824	114.583	1.931	1.715	2.059
.050	93.345	58.536	123.155	1.970	1.767	2.090
.060	100.525	64.717	130.739	2.002	1.811	2.116
.070	107.096	70.503	137.610	2.030	1.848	2.139
.080	113.202	75.982	143.942	2.054	1.881	2.158
.090	118.940	81.216	149.852	2.075	1.910	2.176
.100	124.379	86.249	155.423	2.095	1.936	2.192
.150	148.569	109.382	179.936	2.172	2.039	2.255
.200	169.839	130.541	201.324	2.230	2.116	2.304
.250	189.684	150.762	221.346	2.278	2.178	2.345
.300	208.909	170.602	240.984	2.320	2.232	2.382
.350	228.054	190.412	260.960	2.358	2.280	2.417
.400	247.553	210.444	281.924	2.394	2.323	2.450
.450	267.815	230.906	304.559	2.428	2.363	2.484
<u>.500</u>	<u>289.276</u>	<u>252.000</u>	<u>329.651</u>	<u>2.461</u>	<u>2.401</u>	<u>2.518</u>
.550	312.457	273.983	358.163	2.495	2.438	2.554
.600	338.031	297.224	391.342	2.529	2.473	2.593
.650	366.934	322.301	430.904	2.565	2.508	2.634
.700	400.560	350.129	479.413	2.603	2.544	2.681
.750	441.159	382.198	541.075	2.645	2.582	2.733
.800	492.707	421.111	623.547	2.693	2.624	2.795
.850	563.245	471.994	742.861	2.751	2.674	2.871
<u>.900</u>	<u>672.786</u>	<u>547.229</u>	<u>940.737</u>	<u>2.828</u>	<u>2.738</u>	<u>2.973</u>
.910	703.554	567.724	998.780	2.847	2.754	2.999
.920	739.215	591.186	1067.306	2.869	2.772	3.028
.930	781.360	618.544	1149.962	2.893	2.791	3.061
.940	832.440	651.212	1252.461	2.920	2.814	3.098
.950	896.468	691.483	1384.377	2.953	2.840	3.141
.960	980.639	743.400	1563.293	2.992	2.871	3.194
.970	1099.593	815.047	1826.150	3.041	2.911	3.262
.980	1289.992	926.191	2269.353	3.111	2.967	3.356
.990	1690.125	1148.707	3280.647	3.228	3.060	3.516

a Logarithm base = 10.

Appendix 4: Generated LC values for *Teclea nobilis* fraction F₁

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	152.436	110.961	188.359	2.183	2.045	2.275
.020	172.567	129.645	208.990	2.237	2.113	2.320
.030	186.697	143.073	223.275	2.271	2.156	2.349
.040	198.084	154.065	234.689	2.297	2.188	2.370
.050	207.857	163.613	244.422	2.318	2.214	2.388
.060	216.554	172.193	253.042	2.336	2.236	2.403
.070	224.479	180.074	260.866	2.351	2.255	2.416
.080	231.820	187.426	268.091	2.365	2.273	2.428
.090	238.705	194.363	274.848	2.378	2.289	2.439
.100	245.223	200.966	281.231	2.390	2.303	2.449
.150	274.159	230.649	309.451	2.438	2.363	2.491
.200	299.572	257.115	334.173	2.477	2.410	2.524
.250	323.243	281.986	357.258	2.510	2.450	2.553
.300	346.091	306.080	379.697	2.539	2.486	2.579
.350	368.702	329.899	402.162	2.567	2.518	2.604
.400	391.522	353.791	425.213	2.593	2.549	2.629
.450	414.945	378.033	449.389	2.618	2.578	2.653
<u>.500</u>	<u>439.363</u>	<u>402.871</u>	<u>475.278</u>	<u>2.643</u>	<u>2.605</u>	<u>2.677</u>
.550	465.218	428.569	503.563	2.668	2.632	2.702
.600	493.049	455.457	535.089	2.693	2.658	2.728
.650	523.566	484.004	570.948	2.719	2.685	2.757
.700	557.772	514.919	612.656	2.746	2.712	2.787
.750	597.196	549.334	662.494	2.776	2.740	2.821
.800	644.386	589.149	724.274	2.809	2.770	2.860
.850	704.117	637.906	805.243	2.848	2.805	2.906
<u>.900</u>	<u>787.201</u>	<u>703.499</u>	<u>922.100</u>	<u>2.896</u>	<u>2.847</u>	<u>2.965</u>
.910	808.697	720.137	953.031	2.908	2.857	2.979
.920	832.714	738.589	987.903	2.920	2.868	2.995
.930	859.947	759.349	1027.824	2.934	2.880	3.012
.940	891.416	783.138	1074.438	2.950	2.894	3.031
.950	928.714	811.081	1130.326	2.968	2.909	3.053
.960	974.534	845.064	1199.889	2.989	2.927	3.079
.970	1033.973	888.636	1291.536	3.015	2.949	3.111
.980	1118.636	949.813	1424.656	3.049	2.978	3.154
.990	1266.369	1054.419	1663.626	3.103	3.023	3.221

a Logarithm base = 10.

Appendix 5: Generated LC values for *Teclea nobilis* fraction F₃

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	50.632	28.765	71.740	1.704	1.459	1.856
.020	60.372	36.270	82.789	1.781	1.560	1.918
.030	67.503	42.005	90.691	1.829	1.623	1.958
.040	73.417	46.901	97.146	1.866	1.671	1.987
.050	78.607	51.294	102.748	1.895	1.710	2.012
.060	83.313	55.350	107.782	1.921	1.743	2.033
.070	87.671	59.164	112.409	1.943	1.772	2.051
.080	91.766	62.796	116.731	1.963	1.798	2.067
.090	95.656	66.287	120.814	1.981	1.821	2.082
.100	99.382	69.667	124.708	1.997	1.843	2.096
.150	116.418	85.520	142.337	2.066	1.932	2.153
.200	132.017	100.519	158.323	2.121	2.002	2.200
.250	147.056	115.310	173.696	2.167	2.062	2.240
.300	162.016	130.248	189.047	2.210	2.115	2.277
.350	177.233	145.572	204.816	2.249	2.163	2.311
.400	192.993	161.470	221.415	2.286	2.208	2.345
.450	209.573	178.105	239.287	2.321	2.251	2.379
<u>.500</u>	<u>227.280</u>	<u>195.637</u>	<u>258.959</u>	<u>2.357</u>	<u>2.291</u>	<u>2.413</u>
.550	246.483	214.245	281.096	2.392	2.331	2.449
.600	267.659	234.170	306.571	2.428	2.370	2.487
.650	291.459	255.769	336.567	2.465	2.408	2.527
.700	318.834	279.617	372.786	2.504	2.447	2.571
.750	351.269	306.676	417.855	2.546	2.487	2.621
.800	391.283	338.622	476.271	2.592	2.530	2.678
.850	443.713	378.671	556.818	2.647	2.578	2.746
<u>.900</u>	<u>519.775</u>	<u>434.154</u>	<u>680.449</u>	<u>2.716</u>	<u>2.638</u>	<u>2.833</u>
.910	540.023	448.518	714.554	2.732	2.652	2.854
.920	562.913	464.584	753.670	2.750	2.667	2.877
.930	589.205	482.829	799.298	2.770	2.684	2.903
.940	620.022	503.956	853.697	2.792	2.702	2.931
.950	657.142	529.066	920.474	2.818	2.724	2.964
.960	703.602	560.026	1005.890	2.847	2.748	3.003
.970	765.244	600.388	1122.197	2.884	2.778	3.050
.980	855.626	658.288	1298.464	2.932	2.818	3.113
.990	1020.237	760.487	1635.456	3.009	2.881	3.214

a Logarithm base = 10.

Appendix 6: Generated LC values for *Teclea nobilis* fraction F₄

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	96.357	63.593	124.159	1.984	1.803	2.094
.020	110.129	75.932	138.395	2.042	1.880	2.141
.030	119.870	84.951	148.306	2.079	1.929	2.171
.040	127.762	92.417	156.255	2.106	1.966	2.194
.050	134.563	98.957	163.058	2.129	1.995	2.212
.060	140.637	104.875	169.101	2.148	2.021	2.228
.070	146.187	110.343	174.601	2.165	2.043	2.242
.080	151.342	115.470	179.694	2.180	2.062	2.255
.090	156.188	120.329	184.470	2.194	2.080	2.266
.100	160.786	124.973	188.994	2.206	2.097	2.276
.150	181.308	146.041	209.138	2.258	2.164	2.320
.200	199.470	165.039	227.015	2.300	2.218	2.356
.250	216.496	183.012	243.941	2.335	2.262	2.387
.300	233.020	200.480	260.649	2.367	2.302	2.416
.350	249.455	217.747	277.665	2.397	2.338	2.444
.400	266.122	235.017	295.450	2.425	2.371	2.470
.450	283.306	252.448	314.463	2.452	2.402	2.498
<u>.500</u>	<u>301.301</u>	<u>270.188</u>	<u>335.205</u>	<u>2.479</u>	<u>2.432</u>	<u>2.525</u>
.550	320.439	288.420	358.251	2.506	2.460	2.554
.600	341.130	307.395	384.303	2.533	2.488	2.585
.650	363.922	327.481	414.280	2.561	2.515	2.617
.700	389.590	349.227	449.489	2.591	2.543	2.653
.750	419.326	373.478	491.953	2.623	2.572	2.692
.800	455.116	401.623	545.123	2.658	2.604	2.736
.850	500.708	436.221	615.656	2.700	2.640	2.789
<u>.900</u>	<u>564.615</u>	<u>482.970</u>	<u>719.066</u>	<u>2.752</u>	<u>2.684</u>	<u>2.857</u>
.910	581.236	494.861	746.744	2.764	2.694	2.873
.920	599.848	508.062	778.095	2.778	2.706	2.891
.930	621.001	522.931	814.172	2.793	2.718	2.911
.940	645.509	539.990	856.542	2.810	2.732	2.933
.950	674.643	560.055	907.681	2.829	2.748	2.958
.960	710.556	584.494	971.825	2.852	2.767	2.988
.970	757.337	615.885	1057.131	2.879	2.789	3.024
.980	824.326	660.058	1182.558	2.916	2.820	3.073
.990	942.142	735.838	1411.836	2.974	2.867	3.150

a Logarithm base = 10.

Appendix 7: Generated LC values for *Teclea nobilis* methanol fraction

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	263.318	210.733	303.551	2.420	2.324	2.482
.020	285.177	233.564	324.219	2.455	2.368	2.511
.030	299.977	249.275	338.111	2.477	2.397	2.529
.040	311.615	261.759	348.988	2.494	2.418	2.543
.050	321.413	272.352	358.122	2.507	2.435	2.554
.060	329.996	281.687	366.109	2.519	2.450	2.564
.070	337.710	290.118	373.279	2.529	2.463	2.572
.080	344.769	297.866	379.837	2.538	2.474	2.580
.090	351.317	305.078	385.919	2.546	2.484	2.586
.100	357.455	311.858	391.621	2.553	2.494	2.593
.150	384.029	341.385	416.361	2.584	2.533	2.619
.200	406.551	366.530	437.495	2.609	2.564	2.641
.250	426.922	389.251	456.857	2.630	2.590	2.660
.300	446.085	410.491	475.394	2.649	2.613	2.677
.350	464.609	430.788	493.720	2.667	2.634	2.693
.400	482.897	450.490	512.313	2.684	2.654	2.710
.450	501.277	469.851	531.599	2.700	2.672	2.726
<u>.500</u>	<u>520.047</u>	<u>489.088</u>	<u>551.998</u>	<u>2.716</u>	<u>2.689</u>	<u>2.742</u>
.550	539.521	508.422	573.958	2.732	2.706	2.759
.600	560.055	528.121	597.992	2.748	2.723	2.777
.650	582.100	548.538	624.739	2.765	2.739	2.796
.700	606.272	570.165	655.075	2.783	2.756	2.816
.750	633.485	593.728	690.320	2.802	2.774	2.839
.800	665.228	620.383	732.665	2.823	2.793	2.865
.850	704.241	652.209	786.226	2.848	2.814	2.896
<u>.900</u>	<u>756.596</u>	<u>693.719</u>	<u>860.285</u>	<u>2.879</u>	<u>2.841</u>	<u>2.935</u>
.910	769.814	704.026	879.330	2.886	2.848	2.944
.920	784.435	715.358	900.548	2.895	2.855	2.955
.930	800.833	727.984	924.525	2.904	2.862	2.966
.940	819.552	742.301	952.122	2.914	2.871	2.979
.950	841.437	758.916	984.677	2.925	2.880	2.993
.960	867.895	778.843	1024.442	2.938	2.891	3.010
.970	901.566	803.967	1075.656	2.955	2.905	3.032
.980	948.356	838.492	1147.906	2.977	2.923	3.060
.990	1027.082	895.681	1272.128	3.012	2.952	3.105

a Logarithm base = 10.

Appendix 8: Generated LC values for *Teclea nobilis* compounds 10, 13 and 21

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	49.139	26.395	71.667	1.691	1.422	1.855
.020	60.002	34.425	84.250	1.778	1.537	1.926
.030	68.108	40.730	93.390	1.833	1.610	1.970
.040	74.920	46.212	100.936	1.875	1.665	2.004
.050	80.961	51.202	107.542	1.908	1.709	2.032
.060	86.484	55.863	113.520	1.937	1.747	2.055
.070	91.637	60.290	119.051	1.962	1.780	2.076
.080	96.510	64.543	124.245	1.985	1.810	2.094
.090	101.166	68.664	129.179	2.005	1.837	2.111
.100	105.651	72.682	133.908	2.024	1.861	2.127
.150	126.436	91.861	155.597	2.102	1.963	2.192
.200	145.833	110.433	175.664	2.164	2.043	2.245
.250	164.830	129.062	195.335	2.217	2.111	2.291
.300	183.990	148.113	215.367	2.265	2.171	2.333
.350	203.727	167.824	236.384	2.309	2.225	2.374
.400	224.410	188.368	259.015	2.351	2.275	2.413
.450	246.417	209.892	283.973	2.392	2.322	2.453
<u>.500</u>	<u>270.181</u>	<u>232.551</u>	<u>312.116</u>	<u>2.432</u>	<u>2.367</u>	<u>2.494</u>
.550	296.237	256.561	344.512	2.472	2.409	2.537
.600	325.288	282.273	382.531	2.512	2.451	2.583
.650	358.312	310.255	428.033	2.554	2.492	2.631
.700	396.748	341.418	483.735	2.599	2.533	2.685
.750	442.866	377.234	553.960	2.646	2.577	2.743
.800	500.556	420.199	646.297	2.699	2.623	2.810
.850	577.351	475.052	775.877	2.761	2.677	2.890
<u>.900</u>	<u>690.932</u>	<u>552.631</u>	<u>979.485</u>	<u>2.839</u>	<u>2.742</u>	<u>2.991</u>
.910	721.561	572.980	1036.585	2.858	2.758	3.016
.920	756.377	595.858	1102.540	2.879	2.775	3.042
.930	796.600	621.981	1180.083	2.901	2.794	3.072
.940	844.058	652.411	1273.347	2.926	2.815	3.105
.950	901.644	688.818	1388.988	2.955	2.838	3.143
.960	974.339	734.043	1538.658	2.989	2.866	3.187
.970	1071.789	793.524	1745.389	3.030	2.900	3.242
.980	1216.585	879.802	2064.580	3.085	2.944	3.315
.990	1485.530	1034.548	2692.012	3.172	3.015	3.430

a Logarithm base = 10.

Appendix 9: Generated LC values for *Teclea nobilis* compound 22

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	69.184	40.909	95.055	1.840	1.612	1.978
.020	81.767	51.125	108.817	1.913	1.709	2.037
.030	90.913	58.874	118.603	1.959	1.770	2.074
.040	98.461	65.454	126.567	1.993	1.816	2.102
.050	105.061	71.334	133.459	2.021	1.853	2.125
.060	111.025	76.742	139.639	2.045	1.885	2.145
.070	116.533	81.811	145.310	2.066	1.913	2.162
.080	121.696	86.624	150.598	2.085	1.938	2.178
.090	126.590	91.238	155.589	2.102	1.960	2.192
.100	131.269	95.694	160.344	2.118	1.981	2.205
.150	152.552	116.445	181.836	2.183	2.066	2.260
.200	171.903	135.858	201.312	2.235	2.133	2.304
.250	190.449	154.786	220.083	2.280	2.190	2.343
.300	208.803	173.667	238.913	2.320	2.240	2.378
.350	227.386	192.765	258.396	2.357	2.285	2.412
.400	246.547	212.261	279.091	2.392	2.327	2.446
.450	266.620	232.296	301.599	2.426	2.366	2.479
<u>.500</u>	<u>287.970</u>	<u>253.014</u>	<u>326.608</u>	<u>2.459</u>	<u>2.403</u>	<u>2.514</u>
.550	311.030	274.609	354.943	2.493	2.439	2.550
.600	336.354	297.378	387.633	2.527	2.473	2.588
.650	364.696	321.791	426.046	2.562	2.508	2.629
.700	397.153	348.581	472.155	2.599	2.542	2.674
.750	435.428	378.898	529.060	2.639	2.579	2.724
.800	482.405	414.666	602.132	2.683	2.618	2.780
.850	543.596	459.477	701.913	2.735	2.662	2.846
<u>.900</u>	<u>631.733</u>	<u>521.437</u>	<u>853.505</u>	<u>2.801</u>	<u>2.717</u>	<u>2.931</u>
.910	655.082	537.443	895.068	2.816	2.730	2.952
.920	681.426	555.327	942.622	2.833	2.745	2.974
.930	711.617	575.610	997.953	2.852	2.760	2.999
.940	746.919	599.060	1063.737	2.873	2.777	3.027
.950	789.324	626.879	1144.246	2.897	2.797	3.059
.960	842.231	661.101	1246.872	2.925	2.820	3.096
.970	912.157	705.585	1386.047	2.960	2.849	3.142
.980	1014.180	769.145	1595.897	3.006	2.886	3.203
.990	1198.640	880.647	1994.123	3.079	2.945	3.300

a Logarithm base = 10.

Appendix 10: Generated LC values for *Teclea nobilis* Essential oil

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT(b)						
.010	62.956	26.247	93.926	1.799	1.419	1.973
.020	71.929	32.578	103.734	1.857	1.513	2.016
.030	78.275	37.349	110.529	1.894	1.572	2.043
.040	83.414	41.382	115.963	1.921	1.617	2.064
.050	87.843	44.973	120.605	1.944	1.653	2.081
.060	91.797	48.265	124.722	1.963	1.684	2.096
.070	95.411	51.343	128.465	1.980	1.710	2.109
.080	98.767	54.258	131.928	1.995	1.734	2.120
.090	101.921	57.047	135.173	2.008	1.756	2.131
.100	104.914	59.734	138.245	2.021	1.776	2.141
.150	118.268	72.185	151.912	2.073	1.858	2.182
.200	130.084	83.755	164.030	2.114	1.923	2.215
.250	141.157	94.980	175.503	2.150	1.978	2.244
.300	151.903	106.143	186.829	2.182	2.026	2.271
.350	162.589	117.422	198.368	2.211	2.070	2.297
.400	173.423	128.946	210.441	2.239	2.110	2.323
.450	184.592	140.817	223.380	2.266	2.149	2.349
<u>.500</u>	<u>196.286</u>	<u>153.124</u>	<u>237.575</u>	<u>2.293</u>	<u>2.185</u>	<u>2.376</u>
.550	208.720	165.955	253.511	2.320	2.220	2.404
.600	222.162	179.416	271.824	2.347	2.254	2.434
.650	236.966	193.655	293.384	2.375	2.287	2.467
.700	253.637	208.910	319.443	2.404	2.320	2.504
.750	272.944	225.589	351.927	2.436	2.353	2.546
.800	296.179	244.430	394.094	2.472	2.388	2.596
.850	325.769	266.859	452.234	2.513	2.426	2.655
<u>.900</u>	<u>367.236</u>	<u>296.119</u>	<u>541.191</u>	<u>2.565</u>	<u>2.471</u>	<u>2.733</u>
.910	378.019	303.407	565.641	2.578	2.482	2.753
.920	390.092	311.437	593.629	2.591	2.493	2.774
.930	403.812	320.410	626.204	2.606	2.506	2.797
.940	419.707	330.622	664.933	2.623	2.519	2.823
.950	438.601	342.531	712.319	2.642	2.535	2.853
.960	461.887	356.905	772.685	2.665	2.553	2.888
.970	492.217	375.184	854.454	2.692	2.574	2.932
.980	535.639	400.610	977.509	2.729	2.603	2.990
.990	611.986	443.575	1210.195	2.787	2.647	3.083

a Logarithm base = 10.

b A heterogeneity factor is used.

Appendix 11: Generated LC values for *Rapanea melanophloeos* methanol crude extract

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	16.222	7.989	24.646	1.210	.902	1.392
.020	19.994	10.610	29.161	1.301	1.026	1.465
.030	22.829	12.695	32.463	1.358	1.104	1.511
.040	25.224	14.525	35.203	1.402	1.162	1.547
.050	27.357	16.202	37.612	1.437	1.210	1.575
.060	29.313	17.777	39.801	1.467	1.250	1.600
.070	31.143	19.280	41.832	1.493	1.285	1.622
.080	32.879	20.730	43.746	1.517	1.317	1.641
.090	34.541	22.139	45.570	1.538	1.345	1.659
.100	36.145	23.517	47.324	1.558	1.371	1.675
.150	43.619	30.144	55.430	1.640	1.479	1.744
.200	50.646	36.616	63.029	1.705	1.564	1.800
.250	57.571	43.143	70.572	1.760	1.635	1.849
.300	64.593	49.841	78.346	1.810	1.698	1.894
.350	71.863	56.787	86.594	1.857	1.754	1.937
.400	79.516	64.043	95.562	1.900	1.806	1.980
.450	87.695	71.667	105.530	1.943	1.855	2.023
<u>.500</u>	<u>96.565</u>	<u>79.726</u>	<u>116.833</u>	<u>1.985</u>	<u>1.902</u>	<u>2.068</u>
.550	106.333	88.317	129.896	2.027	1.946	2.114
.600	117.270	97.584	145.276	2.069	1.989	2.162
.650	129.759	107.745	163.755	2.113	2.032	2.214
.700	144.362	119.141	186.494	2.159	2.076	2.271
.750	161.970	132.316	215.365	2.209	2.122	2.333
.800	184.116	148.199	253.673	2.265	2.171	2.404
.850	213.779	168.563	308.050	2.330	2.227	2.489
<u>.900</u>	<u>257.984</u>	<u>197.486</u>	<u>394.756</u>	<u>2.412</u>	<u>2.296</u>	<u>2.596</u>
.910	269.966	205.092	419.315	2.431	2.312	2.623
.920	283.613	213.653	447.804	2.453	2.330	2.651
.930	299.418	223.439	481.453	2.476	2.349	2.683
.940	318.112	234.854	522.133	2.503	2.371	2.718
.950	340.861	248.531	572.869	2.533	2.395	2.758
.960	369.676	265.550	638.981	2.568	2.424	2.805
.970	408.459	287.980	731.050	2.611	2.459	2.864
.980	466.385	320.603	874.716	2.669	2.506	2.942
.990	574.809	379.353	1161.603	2.760	2.579	3.065

a Logarithm base = 10.

Appendix 12: Generated LC values for *Rapanea melanophloeos* Ethyl acetate extract

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	53.072	30.733	74.517	1.725	1.488	1.872
.020	63.400	38.803	86.166	1.802	1.589	1.935
.030	70.972	44.975	94.513	1.851	1.653	1.975
.040	77.258	50.248	101.342	1.888	1.701	2.006
.050	82.780	54.980	107.275	1.918	1.740	2.030
.060	87.790	59.351	112.611	1.943	1.773	2.052
.070	92.432	63.461	117.521	1.966	1.803	2.070
.080	96.796	67.376	122.110	1.986	1.829	2.087
.090	100.943	71.140	126.450	2.004	1.852	2.102
.100	104.918	74.784	130.591	2.021	1.874	2.116
.150	123.109	91.874	149.383	2.090	1.963	2.174
.200	139.791	108.033	166.485	2.145	2.034	2.221
.250	155.893	123.950	182.996	2.193	2.093	2.262
.300	171.928	139.997	199.551	2.235	2.146	2.300
.350	188.256	156.417	216.636	2.275	2.194	2.336
.400	205.180	173.399	234.708	2.312	2.239	2.371
.450	223.002	191.102	254.265	2.348	2.281	2.405
<u>.500</u>	<u>242.052</u>	<u>209.685</u>	<u>275.891</u>	<u>2.384</u>	<u>2.322</u>	<u>2.441</u>
.550	262.729	229.341	300.316	2.420	2.360	2.478
.600	285.549	250.336	328.482	2.456	2.399	2.517
.650	311.220	273.081	361.663	2.493	2.436	2.558
.700	340.776	298.230	401.688	2.532	2.475	2.604
.750	375.827	326.848	451.415	2.575	2.514	2.655
.800	419.118	360.769	515.760	2.622	2.557	2.712
.850	475.911	403.475	604.370	2.678	2.606	2.781
<u>.900</u>	<u>558.427</u>	<u>462.893</u>	<u>740.301</u>	<u>2.747</u>	<u>2.665</u>	<u>2.869</u>
.910	580.415	478.313	777.799	2.764	2.680	2.891
.920	605.284	495.575	820.811	2.782	2.695	2.914
.930	633.860	515.196	870.991	2.802	2.712	2.940
.940	667.374	537.936	930.828	2.824	2.731	2.969
.950	707.764	564.991	1004.302	2.850	2.752	3.002
.960	758.351	598.383	1098.317	2.880	2.777	3.041
.970	825.523	641.967	1226.392	2.917	2.808	3.089
.980	924.116	704.576	1420.627	2.966	2.848	3.152
.990	1103.959	815.299	1792.369	3.043	2.911	3.253

a Logarithm base = 10.

Appendix 13: Generated LC values for *Rapanea melanophloeos* hexane extract

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	36.806	21.354	51.957	1.566	1.329	1.716
.020	43.418	26.453	59.523	1.638	1.422	1.775
.030	48.216	30.294	64.902	1.683	1.481	1.812
.040	52.172	33.542	69.278	1.717	1.526	1.841
.050	55.627	36.435	73.063	1.745	1.562	1.864
.060	58.749	39.088	76.454	1.769	1.592	1.883
.070	61.629	41.570	79.564	1.790	1.619	1.901
.080	64.328	43.923	82.461	1.808	1.643	1.916
.090	66.885	46.175	85.193	1.825	1.664	1.930
.100	69.328	48.347	87.794	1.841	1.684	1.943
.150	80.431	58.439	99.502	1.905	1.767	1.998
.200	90.510	67.872	110.027	1.957	1.832	2.041
.250	100.158	77.092	120.059	2.001	1.887	2.079
.300	109.696	86.346	129.978	2.040	1.936	2.114
.350	119.342	95.805	140.053	2.077	1.981	2.146
.400	129.279	105.607	150.519	2.112	2.024	2.178
.450	139.681	115.883	161.613	2.145	2.064	2.208
<u>.500</u>	<u>150.734</u>	<u>126.766</u>	<u>173.611</u>	<u>2.178</u>	<u>2.103</u>	<u>2.240</u>
.550	162.661	138.409	186.854	2.211	2.141	2.272
.600	175.748	150.996	201.796	2.245	2.179	2.305
.650	190.382	164.778	219.071	2.280	2.217	2.341
.700	207.124	180.117	239.605	2.316	2.256	2.379
.750	226.847	197.589	264.848	2.356	2.296	2.423
.800	251.028	218.191	297.259	2.400	2.339	2.473
.850	282.486	243.855	341.579	2.451	2.387	2.533
<u>.900</u>	<u>327.726</u>	<u>279.031</u>	<u>408.958</u>	<u>2.516</u>	<u>2.446</u>	<u>2.612</u>
.910	339.698	288.069	427.419	2.531	2.459	2.631
.920	353.200	298.147	448.528	2.548	2.474	2.652
.930	368.667	309.556	473.068	2.567	2.491	2.675
.940	386.743	322.721	502.211	2.587	2.509	2.701
.950	408.443	338.309	537.823	2.611	2.529	2.731
.960	435.498	357.449	583.127	2.639	2.553	2.766
.970	471.227	382.284	644.399	2.673	2.582	2.809
.980	523.301	417.707	736.429	2.719	2.621	2.867
.990	617.304	479.741	909.978	2.790	2.681	2.959

a Logarithm base = 10.

Appendix 14: Generated LC values for *Rapanea melanophloeos* fraction F₁

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT(b)						
.010	68.590	20.918	110.904	1.836	1.321	2.045
.020	80.988	28.075	125.019	1.908	1.448	2.097
.030	89.992	33.817	134.978	1.954	1.529	2.130
.040	97.419	38.881	143.049	1.989	1.590	2.155
.050	103.910	43.541	150.017	2.017	1.639	2.176
.060	109.774	47.932	156.256	2.041	1.681	2.194
.070	115.188	52.133	161.977	2.061	1.717	2.209
.080	120.262	56.195	167.312	2.080	1.750	2.224
.090	125.070	60.151	172.349	2.097	1.779	2.236
.100	129.665	64.027	177.151	2.113	1.806	2.248
.150	150.559	82.730	198.953	2.178	1.918	2.299
.200	169.540	101.056	218.961	2.229	2.005	2.340
.250	187.721	119.537	238.608	2.274	2.078	2.378
.300	205.704	138.425	258.813	2.313	2.141	2.413
.350	223.901	157.840	280.371	2.350	2.198	2.448
.400	242.656	177.815	304.119	2.385	2.250	2.483
.450	262.294	198.328	331.026	2.419	2.297	2.520
<u>.500</u>	<u>283.173</u>	<u>219.340</u>	<u>362.264</u>	<u>2.452</u>	<u>2.341</u>	<u>2.559</u>
.550	305.714	240.861	399.278	2.485	2.382	2.601
.600	330.456	263.012	443.916	2.519	2.420	2.647
.650	358.135	286.097	498.685	2.554	2.457	2.698
.700	389.818	310.668	567.274	2.591	2.492	2.754
.750	427.160	337.635	655.624	2.631	2.528	2.817
.800	472.967	368.517	774.285	2.675	2.566	2.889
.850	532.596	406.102	944.584	2.726	2.609	2.975
<u>.900</u>	<u>618.416</u>	<u>456.571</u>	<u>1219.189</u>	<u>2.791</u>	<u>2.660</u>	<u>3.086</u>
.910	641.139	469.384	1297.498	2.807	2.672	3.113
.920	666.771	483.608	1388.580	2.824	2.684	3.143
.930	696.140	499.630	1496.475	2.843	2.699	3.175
.940	730.472	518.020	1627.333	2.864	2.714	3.211
.950	771.699	539.664	1791.118	2.887	2.732	3.253
.960	823.117	566.059	2005.412	2.915	2.753	3.302
.970	891.049	600.025	2305.301	2.950	2.778	3.363
.980	990.109	647.960	2776.176	2.996	2.812	3.443
.990	1169.070	730.606	3725.156	3.068	2.864	3.571

a Logarithm base = 10.

b A heterogeneity factor is used.

Appendix 15: Generated LC values for *Rapanea melanophloeos* fraction F₃

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	57.424	36.504	76.257	1.759	1.562	1.882
.020	65.746	43.561	85.207	1.818	1.639	1.930
.030	71.642	48.718	91.444	1.855	1.688	1.961
.040	76.423	52.989	96.449	1.883	1.724	1.984
.050	80.546	56.731	100.732	1.906	1.754	2.003
.060	84.231	60.119	104.537	1.925	1.779	2.019
.070	87.599	63.250	107.999	1.943	1.801	2.033
.080	90.730	66.188	111.203	1.958	1.821	2.046
.090	93.674	68.974	114.206	1.972	1.839	2.058
.100	96.469	71.639	117.048	1.984	1.855	2.068
.150	108.954	83.756	129.673	2.037	1.923	2.113
.200	120.019	94.741	140.805	2.079	1.977	2.149
.250	130.404	105.212	151.251	2.115	2.022	2.180
.300	140.493	115.489	161.441	2.148	2.063	2.208
.350	150.538	125.780	171.668	2.178	2.100	2.235
.400	160.733	136.240	182.175	2.206	2.134	2.260
.450	171.253	147.000	193.198	2.234	2.167	2.286
.500	182.279	158.186	204.997	2.261	2.199	2.312
.550	194.014	169.933	217.888	2.288	2.230	2.338
.600	206.713	182.403	232.277	2.315	2.261	2.366
.650	220.712	195.808	248.713	2.344	2.292	2.396
.700	236.492	210.459	267.985	2.374	2.323	2.428
.750	254.789	226.848	291.297	2.406	2.356	2.464
.800	276.835	245.826	320.663	2.442	2.391	2.506
.850	304.950	269.019	359.904	2.484	2.430	2.556
.900	344.418	300.120	417.855	2.537	2.477	2.621
.910	354.693	307.996	433.422	2.550	2.489	2.637
.920	366.202	316.727	451.077	2.564	2.501	2.654
.930	379.290	326.547	471.416	2.579	2.514	2.673
.940	394.459	337.800	495.328	2.596	2.529	2.695
.950	412.503	351.018	524.217	2.615	2.545	2.720
.960	434.759	367.100	560.487	2.638	2.565	2.749
.970	463.772	387.737	608.763	2.666	2.589	2.784
.980	505.359	416.756	679.794	2.704	2.620	2.832
.990	578.605	466.528	809.708	2.762	2.669	2.908

a Logarithm base = 10.

Appendix 16: Generated LC values for *Rapanea melanophloeos* fraction F₄

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	69.395	44.016	92.132	1.841	1.644	1.964
.020	80.213	53.216	103.722	1.904	1.726	2.016
.030	87.936	60.010	111.849	1.944	1.778	2.049
.040	94.231	65.675	118.401	1.974	1.817	2.073
.050	99.683	70.667	124.027	1.999	1.849	2.094
.060	104.571	75.206	129.040	2.019	1.876	2.111
.070	109.054	79.418	133.613	2.038	1.900	2.126
.080	113.231	83.382	137.856	2.054	1.921	2.139
.090	117.168	87.153	141.842	2.069	1.940	2.152
.100	120.913	90.769	145.622	2.082	1.958	2.163
.150	137.736	107.318	162.510	2.139	2.031	2.211
.200	152.761	122.446	177.537	2.184	2.088	2.249
.250	166.950	136.945	191.766	2.223	2.137	2.283
.300	180.813	151.227	205.781	2.257	2.180	2.313
.350	194.684	165.549	219.997	2.289	2.219	2.342
.400	208.828	180.098	234.775	2.320	2.256	2.371
.450	223.491	195.028	250.482	2.349	2.290	2.399
<u>.500</u>	<u>238.926</u>	<u>210.482</u>	<u>267.531</u>	<u>2.378</u>	<u>2.323</u>	<u>2.427</u>
.550	255.427	226.619	286.424	2.407	2.355	2.457
.600	273.362	243.646	307.794	2.437	2.387	2.488
.650	293.223	261.857	332.485	2.467	2.418	2.522
.700	315.717	281.713	361.691	2.499	2.450	2.558
.750	341.932	303.953	397.240	2.534	2.483	2.599
.800	373.693	329.842	442.220	2.573	2.518	2.646
.850	414.456	361.765	502.571	2.617	2.558	2.701
<u>.900</u>	<u>472.121</u>	<u>405.091</u>	<u>592.188</u>	<u>2.674</u>	<u>2.608</u>	<u>2.772</u>
.910	487.212	416.150	616.367	2.688	2.619	2.790
.920	504.153	428.449	643.842	2.703	2.632	2.809
.930	523.462	442.326	675.568	2.719	2.646	2.830
.940	545.902	458.282	712.968	2.737	2.661	2.853
.950	572.671	477.096	758.296	2.758	2.679	2.880
.960	605.804	500.082	815.423	2.782	2.699	2.911
.970	649.175	529.718	891.827	2.812	2.724	2.950
.980	711.674	571.632	1004.972	2.852	2.757	3.002
.990	822.618	644.088	1213.975	2.915	2.809	3.084

a Logarithm base = 10.

Appendix 17: Generated LC values for *Rapanea melanophloeos* methanol fraction

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	56.665	33.964	77.759	1.753	1.531	1.891
.020	66.283	41.742	88.361	1.821	1.621	1.946
.030	73.215	47.563	95.849	1.865	1.677	1.982
.040	78.903	52.463	101.915	1.897	1.720	2.008
.050	83.855	56.812	107.144	1.924	1.754	2.030
.060	88.314	60.791	111.816	1.946	1.784	2.049
.070	92.418	64.502	116.090	1.966	1.810	2.065
.080	96.254	68.012	120.064	1.983	1.833	2.079
.090	99.881	71.365	123.805	1.999	1.853	2.093
.100	103.340	74.593	127.360	2.014	1.873	2.105
.150	118.985	89.515	143.306	2.075	1.952	2.156
.200	133.092	103.356	157.576	2.124	2.014	2.197
.250	146.519	116.789	171.142	2.166	2.067	2.233
.300	159.728	130.174	184.544	2.203	2.115	2.266
.350	173.029	143.747	198.168	2.238	2.158	2.297
.400	186.673	157.688	212.354	2.271	2.198	2.327
.450	200.897	172.149	227.456	2.303	2.236	2.357
<u>.500</u>	<u>215.954</u>	<u>187.278</u>	<u>243.883</u>	<u>2.334</u>	<u>2.272</u>	<u>2.387</u>
.550	232.138	203.235	262.141	2.366	2.308	2.419
.600	249.826	220.223	282.889	2.398	2.343	2.452
.650	269.526	238.529	307.018	2.431	2.378	2.487
.700	291.970	258.601	335.800	2.465	2.413	2.526
.750	318.292	281.182	371.184	2.503	2.449	2.570
.800	350.405	307.565	416.461	2.545	2.488	2.620
.850	391.947	340.229	477.973	2.593	2.532	2.679
<u>.900</u>	<u>451.285</u>	<u>384.803</u>	<u>570.647</u>	<u>2.654</u>	<u>2.585</u>	<u>2.756</u>
.910	466.915	396.228	595.888	2.669	2.598	2.775
.920	484.509	408.956	624.680	2.685	2.612	2.796
.930	504.621	423.347	658.065	2.703	2.627	2.818
.940	528.071	439.931	697.603	2.723	2.643	2.844
.950	556.149	459.538	745.770	2.745	2.662	2.873
.960	591.051	483.567	806.837	2.772	2.684	2.907
.970	636.974	514.667	889.096	2.804	2.712	2.949
.980	703.591	558.875	1012.031	2.847	2.747	3.005
.990	823.013	635.872	1242.205	2.915	2.803	3.094

a Logarithm base = 10.

Appendix 18: Generated LC values for *Rapanea melanophloeos* compound 23

Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT(b)						
.010	32.197	15.675	48.552	1.508	1.195	1.686
.020	38.245	19.915	55.694	1.583	1.299	1.746
.030	42.658	23.171	60.789	1.630	1.365	1.784
.040	46.311	25.959	64.946	1.666	1.414	1.813
.050	49.512	28.466	68.551	1.695	1.454	1.836
.060	52.410	30.784	71.790	1.719	1.488	1.856
.070	55.090	32.967	74.766	1.741	1.518	1.874
.080	57.606	35.048	77.546	1.760	1.545	1.890
.090	59.994	37.050	80.174	1.778	1.569	1.904
.100	62.279	38.989	82.680	1.794	1.591	1.917
.150	72.705	48.096	94.040	1.862	1.682	1.973
.200	82.222	56.720	104.372	1.915	1.754	2.019
.250	91.374	65.222	114.343	1.961	1.814	2.058
.300	100.458	73.803	124.334	2.002	1.868	2.095
.350	109.679	82.600	134.631	2.040	1.917	2.129
.400	119.211	91.722	145.493	2.076	1.962	2.163
.450	129.221	101.271	157.197	2.111	2.005	2.196
<u>.500</u>	<u>139.893</u>	<u>111.354</u>	<u>170.070</u>	<u>2.146</u>	<u>2.047</u>	<u>2.231</u>
.550	151.446	122.094	184.519	2.180	2.087	2.266
.600	164.163	133.650	201.088	2.215	2.126	2.303
.650	178.430	146.242	220.533	2.251	2.165	2.343
.700	194.809	160.204	243.965	2.290	2.205	2.387
.750	214.175	176.070	273.133	2.331	2.246	2.436
.800	238.015	194.764	311.056	2.377	2.290	2.493
.850	269.172	218.057	363.642	2.430	2.339	2.561
<u>.900</u>	<u>314.232</u>	<u>250.005</u>	<u>445.021</u>	<u>2.497</u>	<u>2.398</u>	<u>2.648</u>
.910	326.202	258.215	467.597	2.513	2.412	2.670
.920	339.723	267.372	493.550	2.531	2.427	2.693
.930	355.238	277.736	523.901	2.551	2.444	2.719
.940	373.405	289.696	560.187	2.572	2.462	2.748
.950	395.261	303.856	604.868	2.597	2.483	2.782
.960	422.578	321.240	662.221	2.626	2.507	2.821
.970	458.763	343.791	740.643	2.662	2.536	2.870
.980	511.705	375.945	860.125	2.709	2.575	2.935
.990	607.824	432.224	1090.309	2.784	2.636	3.038

a Logarithm base = 10.

b A heterogeneity factor is used.

Appendix 19: Generated LC values for *Rapanea melanophloeos* compound 24

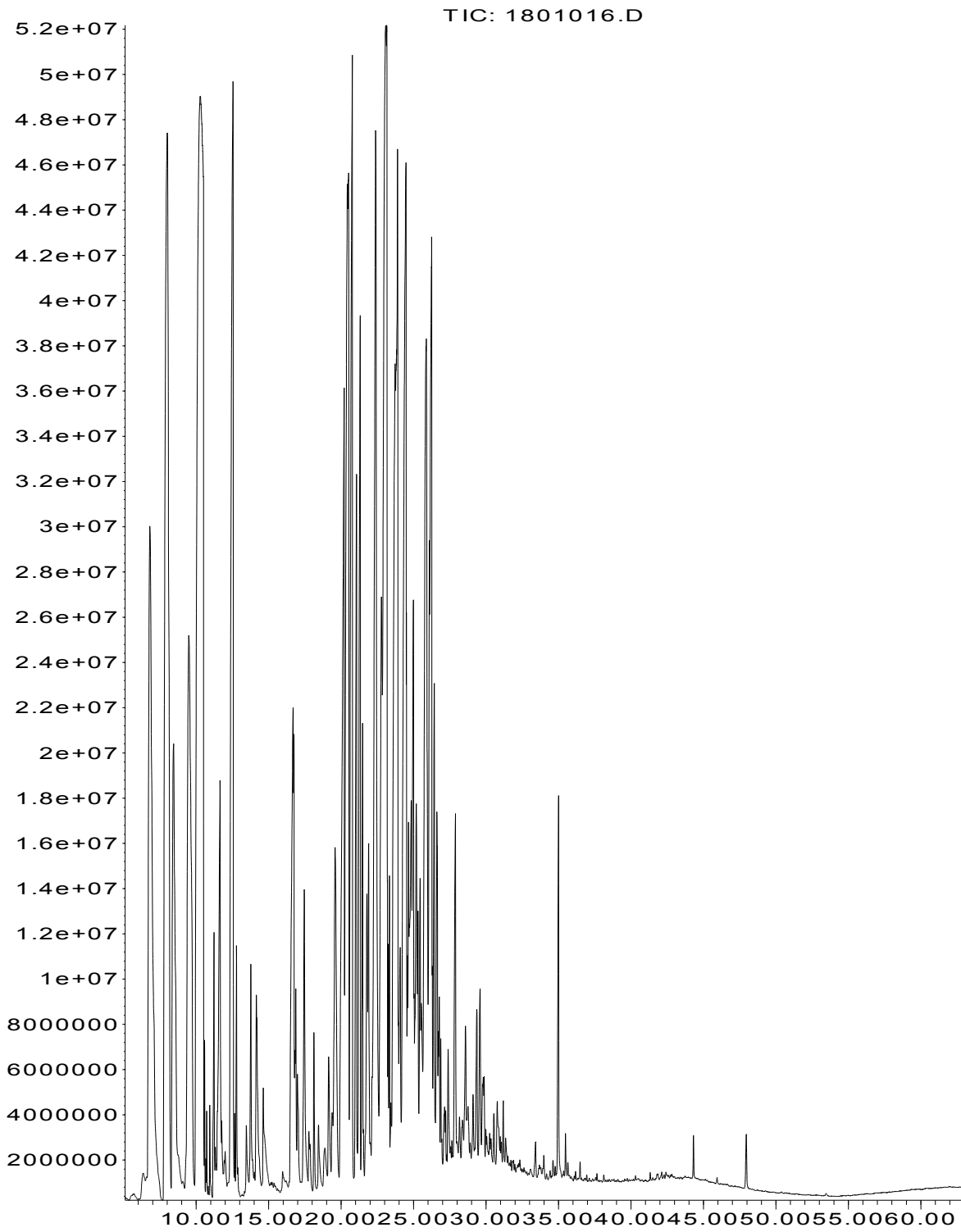
Confidence Limits

Probability	95% Confidence Limits for conc			95% Confidence Limits for log(conc)(a)		
	Estimate	Lower Bound	Upper Bound	Estimate	Lower Bound	Upper Bound
PROBIT .010	19.832	11.622	28.307	1.297	1.065	1.452
.020	23.917	14.707	33.126	1.379	1.168	1.520
.030	26.935	17.071	36.612	1.430	1.232	1.564
.040	29.453	19.093	39.483	1.469	1.281	1.596
.050	31.674	20.909	41.990	1.501	1.320	1.623
.060	33.697	22.588	44.254	1.528	1.354	1.646
.070	35.576	24.168	46.344	1.551	1.383	1.666
.080	37.347	25.674	48.304	1.572	1.409	1.684
.090	39.034	27.122	50.162	1.591	1.433	1.700
.100	40.654	28.526	51.940	1.609	1.455	1.716
.150	48.111	35.121	60.049	1.682	1.546	1.779
.200	55.000	41.380	67.475	1.740	1.617	1.829
.250	61.692	47.573	74.664	1.790	1.677	1.873
.300	68.393	53.854	81.872	1.835	1.731	1.913
.350	75.251	60.335	89.287	1.877	1.781	1.951
.400	82.392	67.109	97.080	1.916	1.827	1.987
.450	89.946	74.269	105.432	1.954	1.871	2.023
<u>.500</u>	<u>98.057</u>	<u>81.918</u>	<u>114.553</u>	<u>1.991</u>	<u>1.913</u>	<u>2.059</u>
.550	106.898	90.175	124.711	2.029	1.955	2.096
.600	116.699	99.194	136.263	2.067	1.996	2.134
.650	127.774	109.187	149.707	2.106	2.038	2.175
.700	140.585	120.468	165.784	2.148	2.081	2.220
.750	155.855	133.532	185.659	2.193	2.126	2.269
.800	174.818	149.230	211.344	2.243	2.174	2.325
.850	199.854	169.204	246.772	2.301	2.228	2.392
<u>.900</u>	<u>236.509</u>	<u>197.252</u>	<u>301.310</u>	<u>2.374</u>	<u>2.295</u>	<u>2.479</u>
.910	246.327	204.570	316.394	2.392	2.311	2.500
.920	257.455	212.779	333.713	2.411	2.328	2.523
.930	270.272	222.131	353.940	2.432	2.347	2.549
.940	285.343	232.997	378.089	2.455	2.367	2.578
.950	303.559	245.963	407.777	2.482	2.391	2.610
.960	326.453	262.021	445.818	2.514	2.418	2.649
.970	356.977	283.070	497.722	2.553	2.452	2.697
.980	402.017	313.477	576.585	2.604	2.496	2.761
.990	484.817	367.716	727.907	2.686	2.566	2.862

a Logarithm base = 10.

Appendix 20: GC/MS spectrum of *Teclea nobilis* essential oil

Abundance

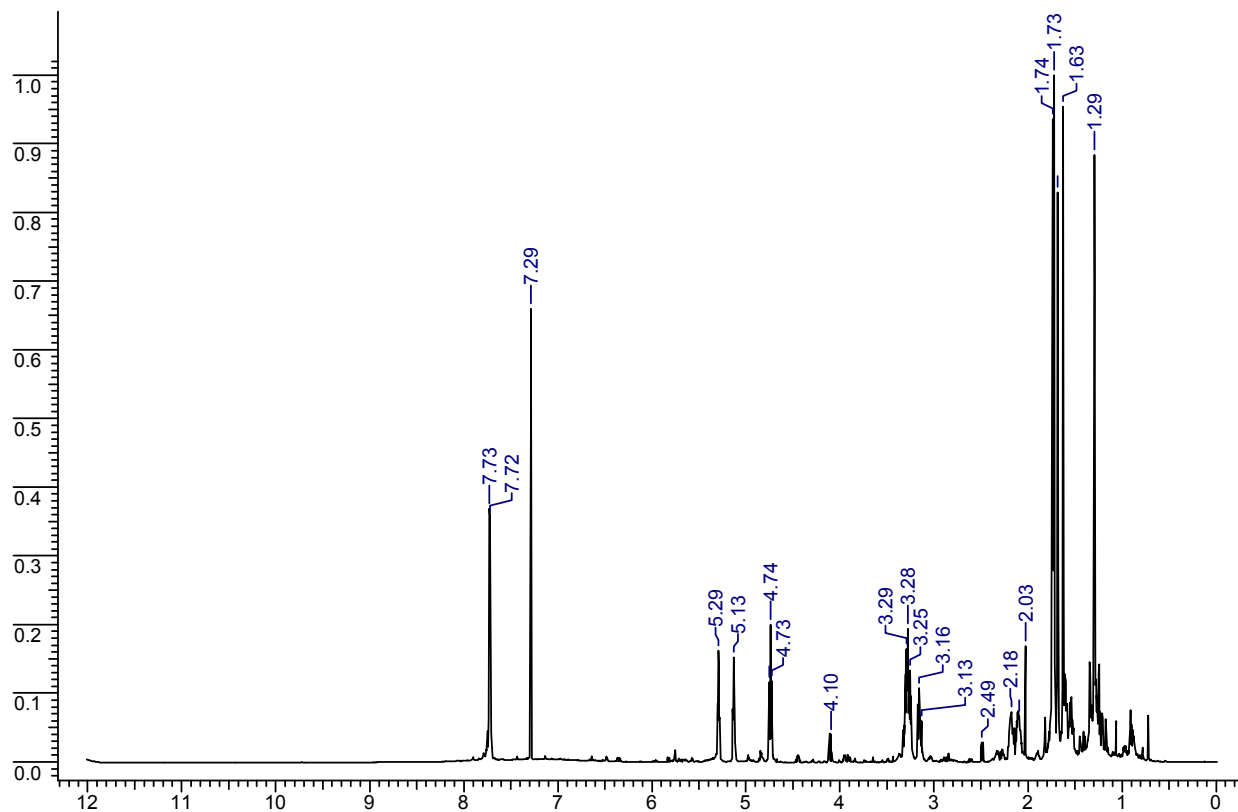


Appendix 21: Constituents of *Teclea nobilis* Essential Oil

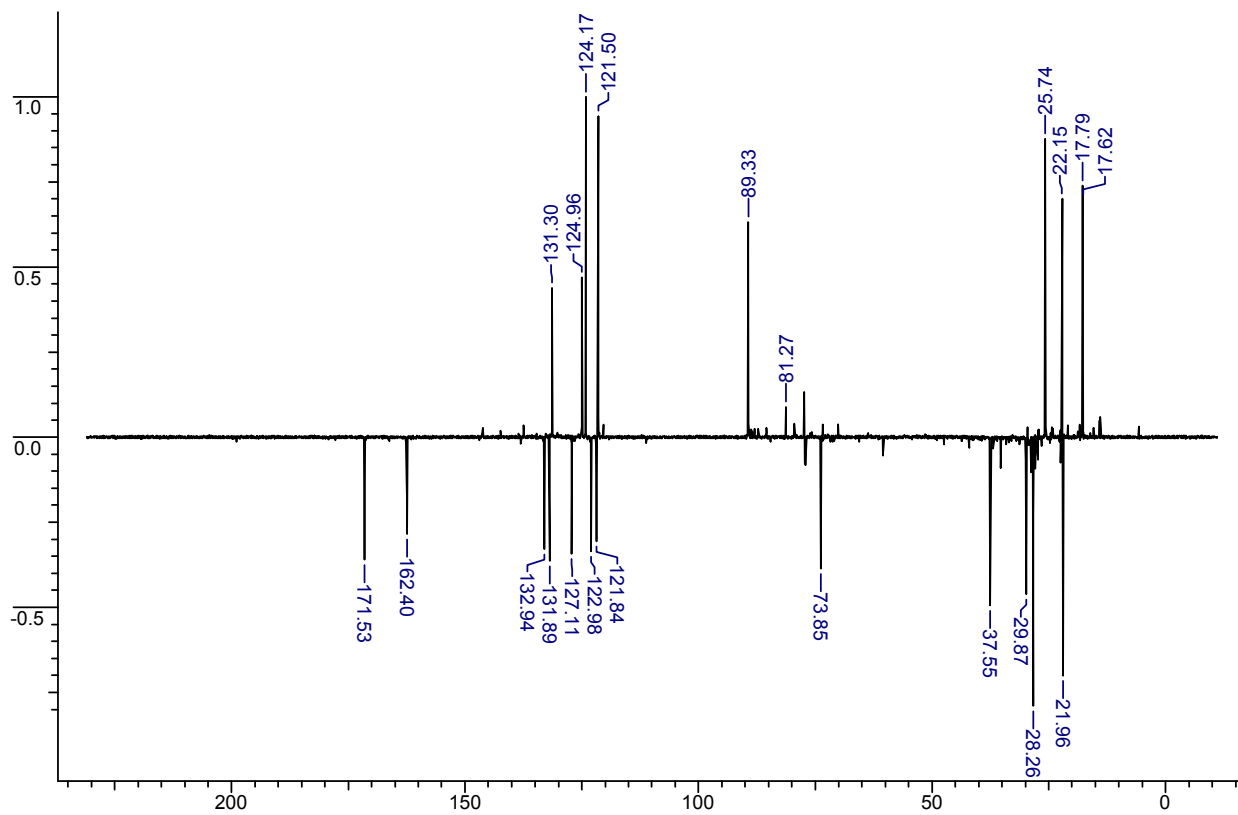
Compound	S/NO	Retention Time (Mins)	Conc. %	Detection Method
Monoterpenes				
α – Pinene	1	6.81	3.95	GC/MS
γ – Terpinene	2	8.01	6.11	GC/MS
β – Myrcene	3	8.45	2.17	GC/MS
Limonene	4	9.50	3.34	GC/MS
β – Ocimene	5	10.29	10.15	GC/MS
α – Terpinolene	6	11.23	0.33	GC/MS
Linalool	7	11.65	1.13	GC/MS
Neo-allo-Ocimene	8	12.55	3.68	GC/MS
Cis-epoxy-Ocimene	9	12.78	0.27	GC/MS
4-methyl-1-(1-methylethyl)-3-cyclohexen-1-ol	10	13.77	0.37	GC/MS
β – Fenchyl alcohol	11	14.15	0.24	GC/MS
α – Terpineol	12	14.19	0.37	GC/MS
3-Ethenyl-Cyclohexane	13	16.81	0.13	GC/MS
Anise Camphor	14	16.88	0.25	GC/MS
Methyl decadienoate	15	17.46	0.71	GC/MS
1,5,5- Trimethyl-6-methylene-cyclohexene	16	18.13	0.22	GC/MS
Methyl Eugenol	17	20.21	2.68	GC/MS
1,2,3,4,4a,5,6,8a-Octahydro-7-methyl-4-methylene-1-(1-naphthalene	18	21.78	0.60	GC/MS
4-ethenyl- α,α ,4-trimethyl-3-(methylethenyl)-[1R-(1 α ,3 α ,4 β)]-Cyclohexanemethanol	19	23.73	2.62	GC/MS
Elemicin	20	23.89	2.89	GC/MS
Total	20		42.21	
Sesquiterpenes				
α – Cubebene	1	18.44	0.9	GC/MS
α – Copaene	2	19.15	0.32	GC/MS
β – Gurjunene	3	19.59	1.26	GC/MS
Germacrene D	4	20.44	3.06	GC/MS
α – Amorphine	5	22.39	3.96	GC/MS
β – Cadinene	6	23.08	4.98	GC/MS
δ – Cadinene	7	23.16	1.28	GC/MS
α – Muurolene	8	23.33	0.29	GC/MS
Nerolidol	9	24.08	0.55	GC/MS
1,6-Germacradien-5-ol	10	24.48	4.38	GC/MS

Guaiol	11	24.84	1.01	GC/MS
β - Elemenone	12	24.99	0.96	GC/MS
1-Methylidene-2b-hydroxymethyl-3,3-dimethyl-4a(3-methylbut-2-enyl)- cyclohexane	13	25.30	0.38	GC/MS
1,2,3,4,6,8a-Hexahydro-1-isopropyl-4,7-dimethyl-Naphthalene	14	25.45	0.52	GC/MS
Tau- Cadinol	15	25.88	3.56	GC/MS
Trans-Isoelemicin	16	26.09	0.83	GC/MS
α – Cadinol	17	26.25	2.57	GC/MS
Bulnesol	18	26.43	0.83	GC/MS
Isoaromadendrene epoxide	19	26.95	0.07	GC/MS
1-Formyl-2,2,6-trimethyl-3-(3-methyl-but-2-enyl)-6-cyclohexene	20	27.98	0.09	GC/MS
Farnesol 2	21	27.39	0.28	GC/MS
(4S,5R)-5-Hydrocaryophyll-8(13)-ene-4,12-epoxide	22	28.17	0.17	GC/MS
Caryophylla-3,8-(13)-dien-5-alpha –ol	23	28.91	0.08	GC/MS
6-Isopropenyl-4,8a-dimethyl-1,2,3,5,6,7,8,8a-octahydronaphthalen-2-ol	24	29.59	0.44	GC/MS
E and Z isomers of 1-(2,6,6-trimethyl-1-cyclohexen-1-yl)-3-methyl-2-heptene	25	29.87	0.13	GC/MS
Dihydrocuminy l aldehyde	26	29.99	0.07	GC/MS
4-methyl-1-(2,3,3-trimethylbicyclo[3.1.0]he-2-yl)-2,4-pentadien-1-one	27	31.36	0.12	GC/MS
Ledene	28	41.51	0.003	GC/MS
Total	28		33.09	
Diterpenes				
(1R,3S)-Cembra-4,7,11,15-tetraen-3-ol	1	33.64	0.03	GC/MS
Phytol	2	35.00	0.60	GC/MS
Docosane	3	41.51	0.01	GC/MS
Tricosane	4	44.31	0.04	GC/MS
Total	4		0.68	
Others				
Octasane	1	45.94	0.01	GC/MS
Nonacosane	2	47.96	0.09	GC/MS
Total	2		0.1	

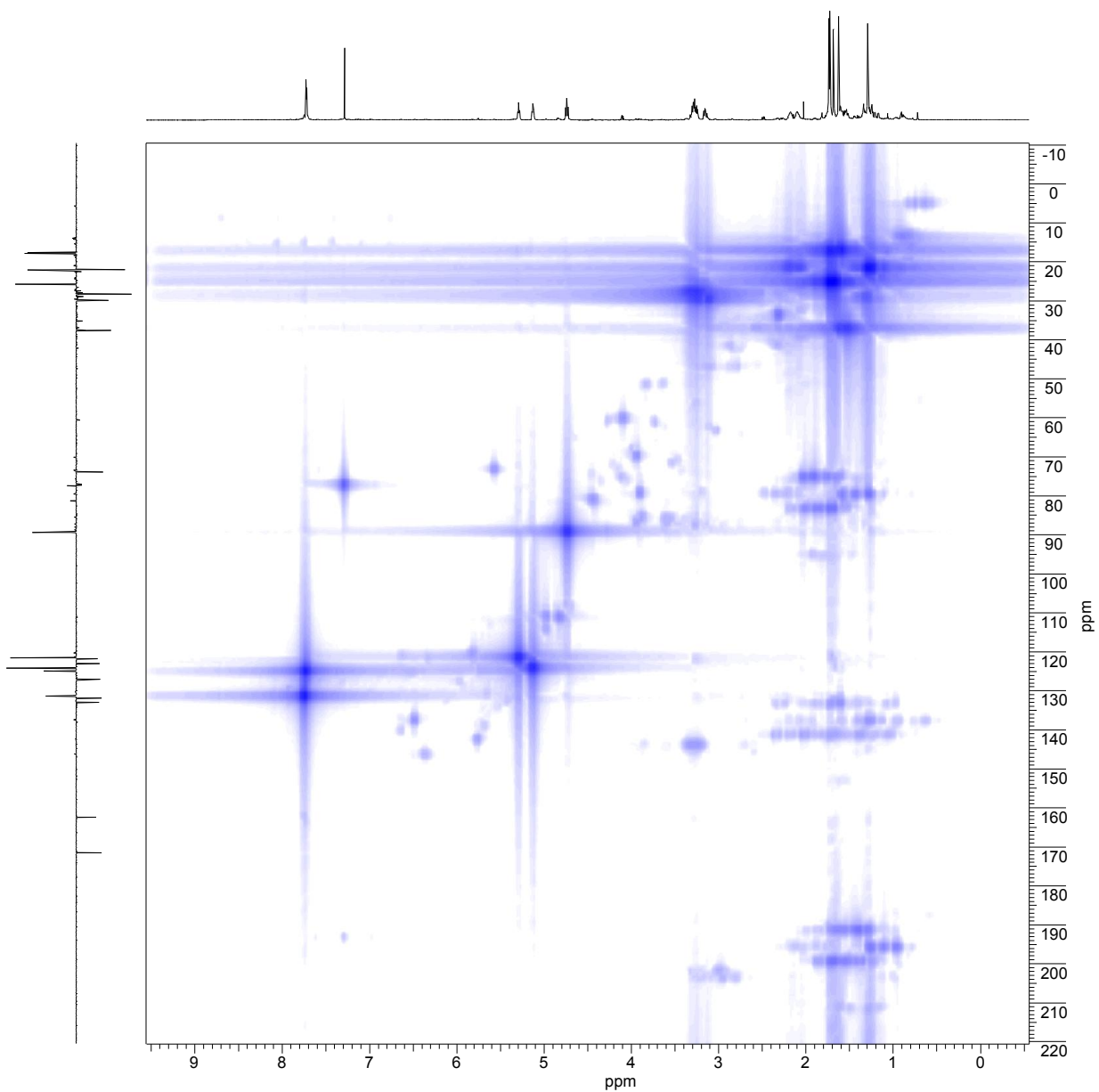
Appendix 22: ^1H NMR spectrum of compound 23



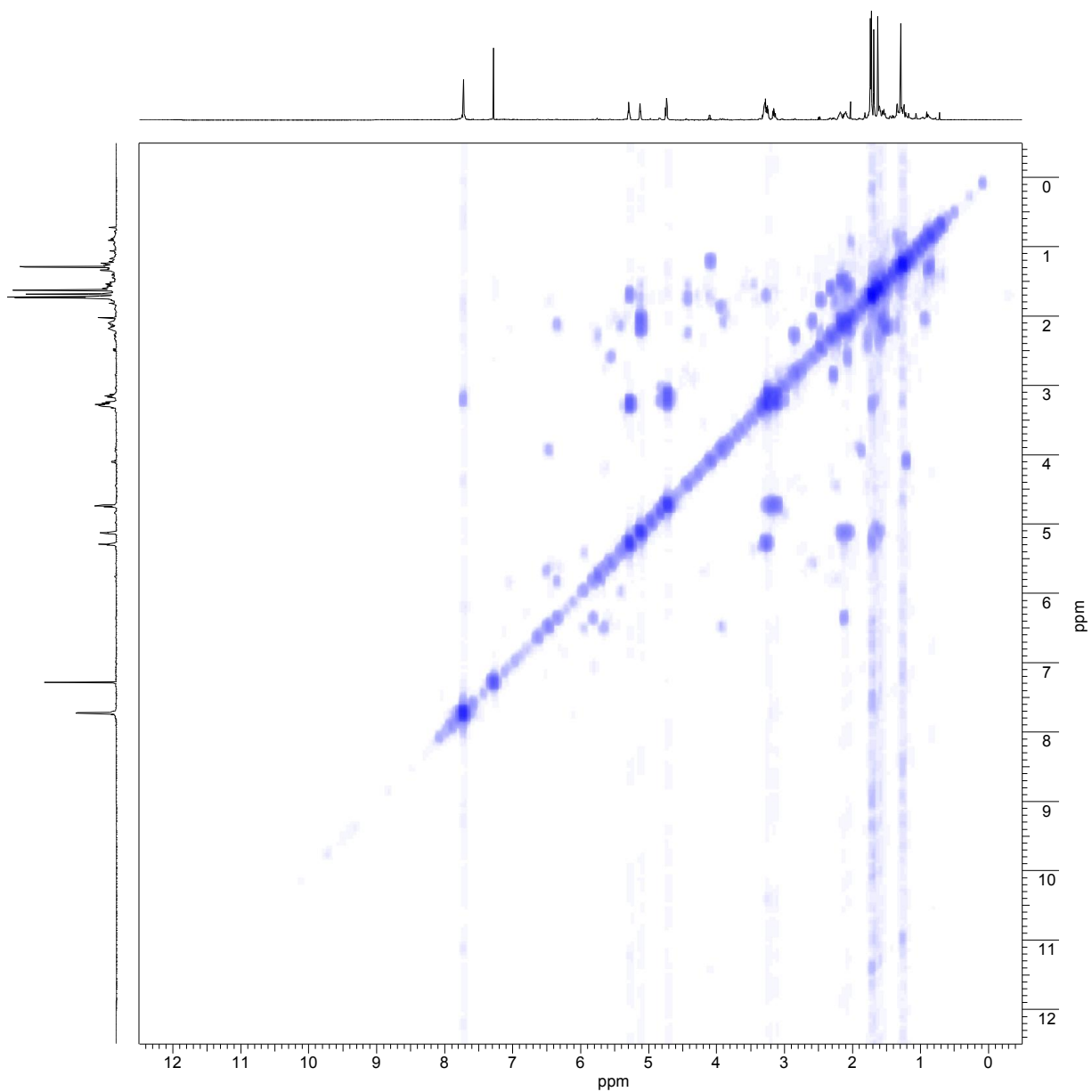
Appendix 23: APT spectrum of compound 23



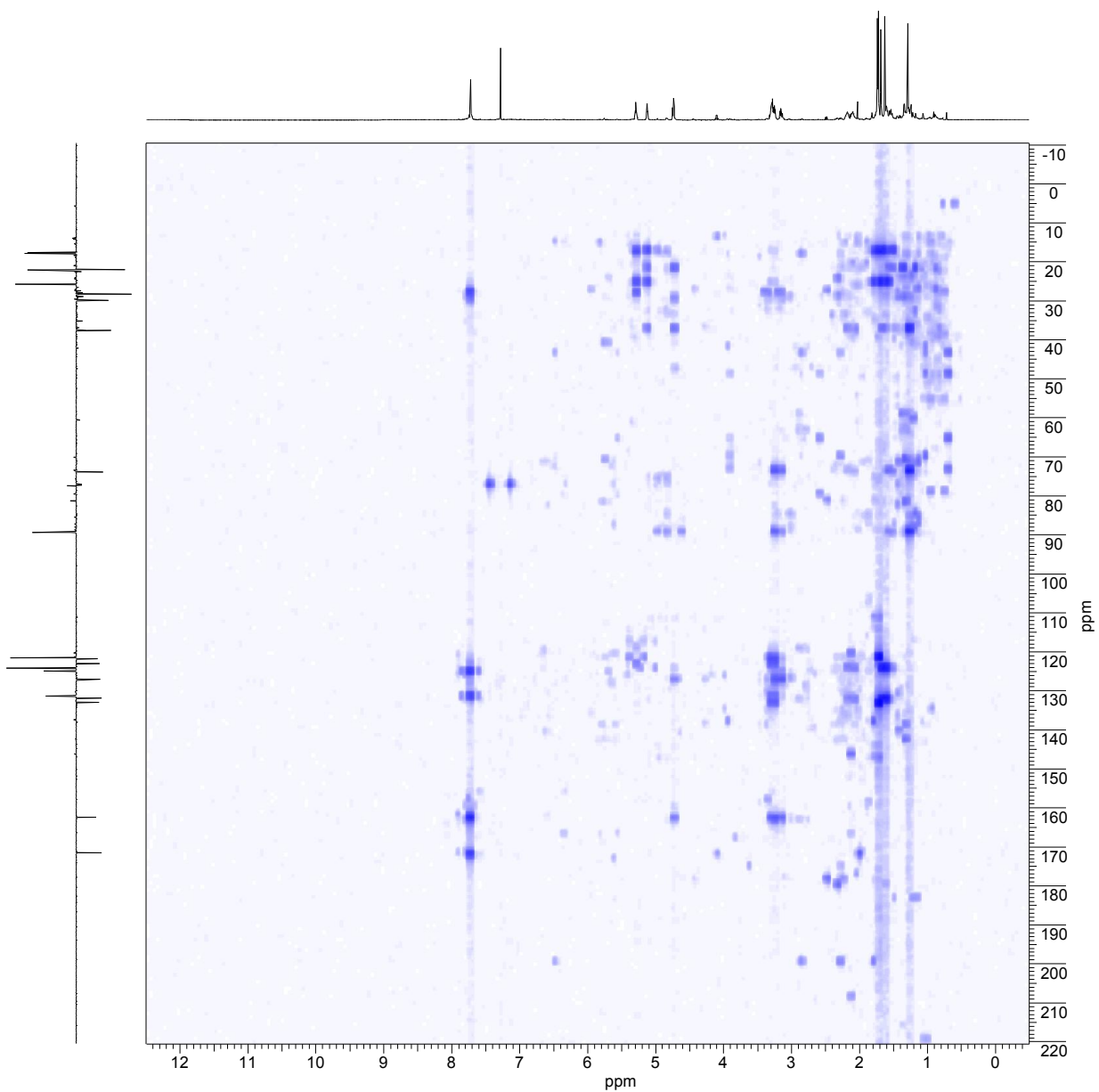
Appendix 24: HSQC spectrum for the compound **23**



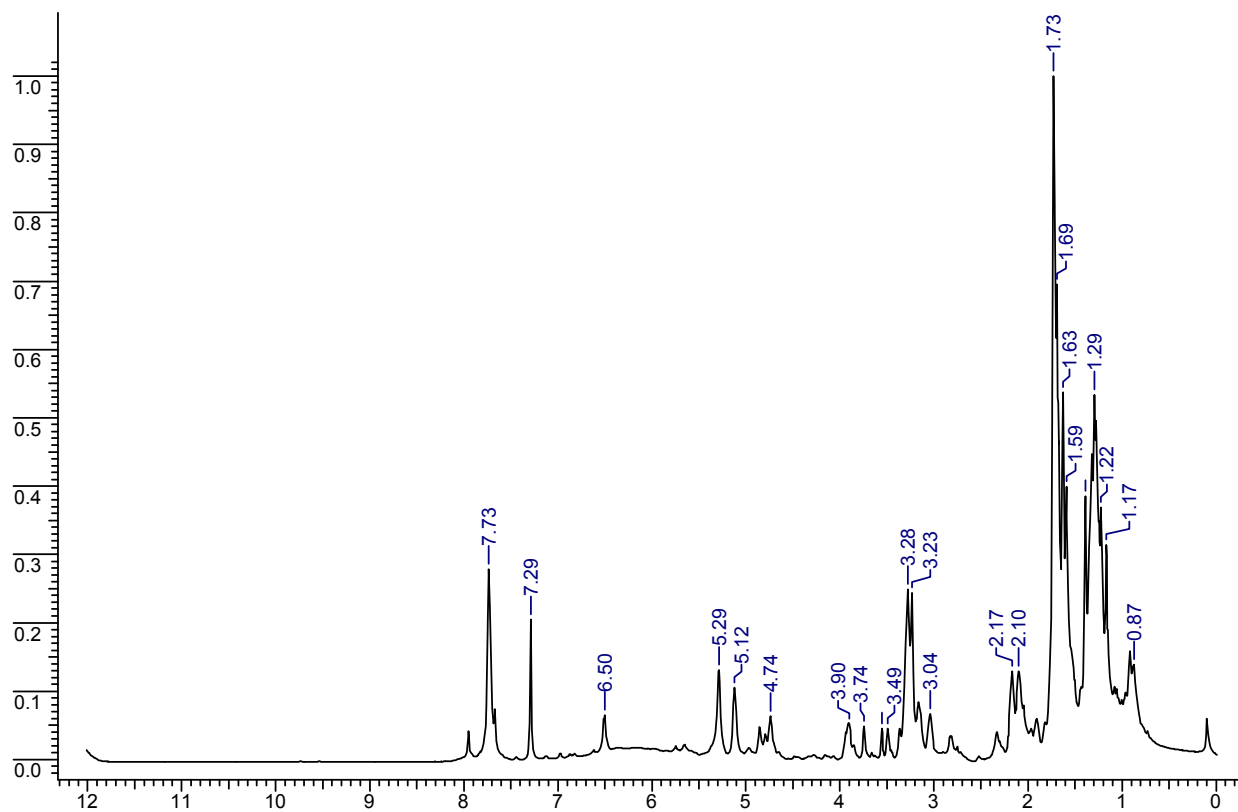
Appendix 25: ^1H - ^1H COSY spectrum for the compound 23



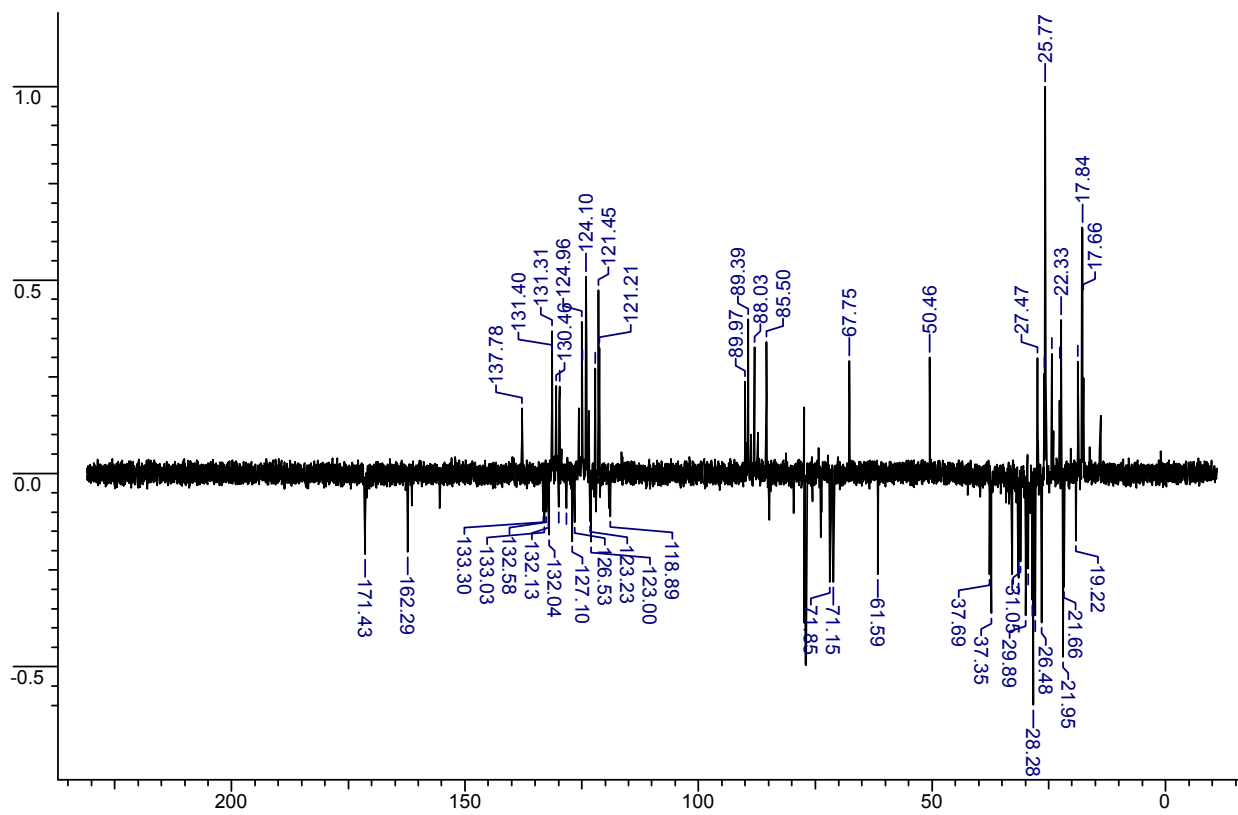
Appendix 26: HMBC spectrum for the compound 23



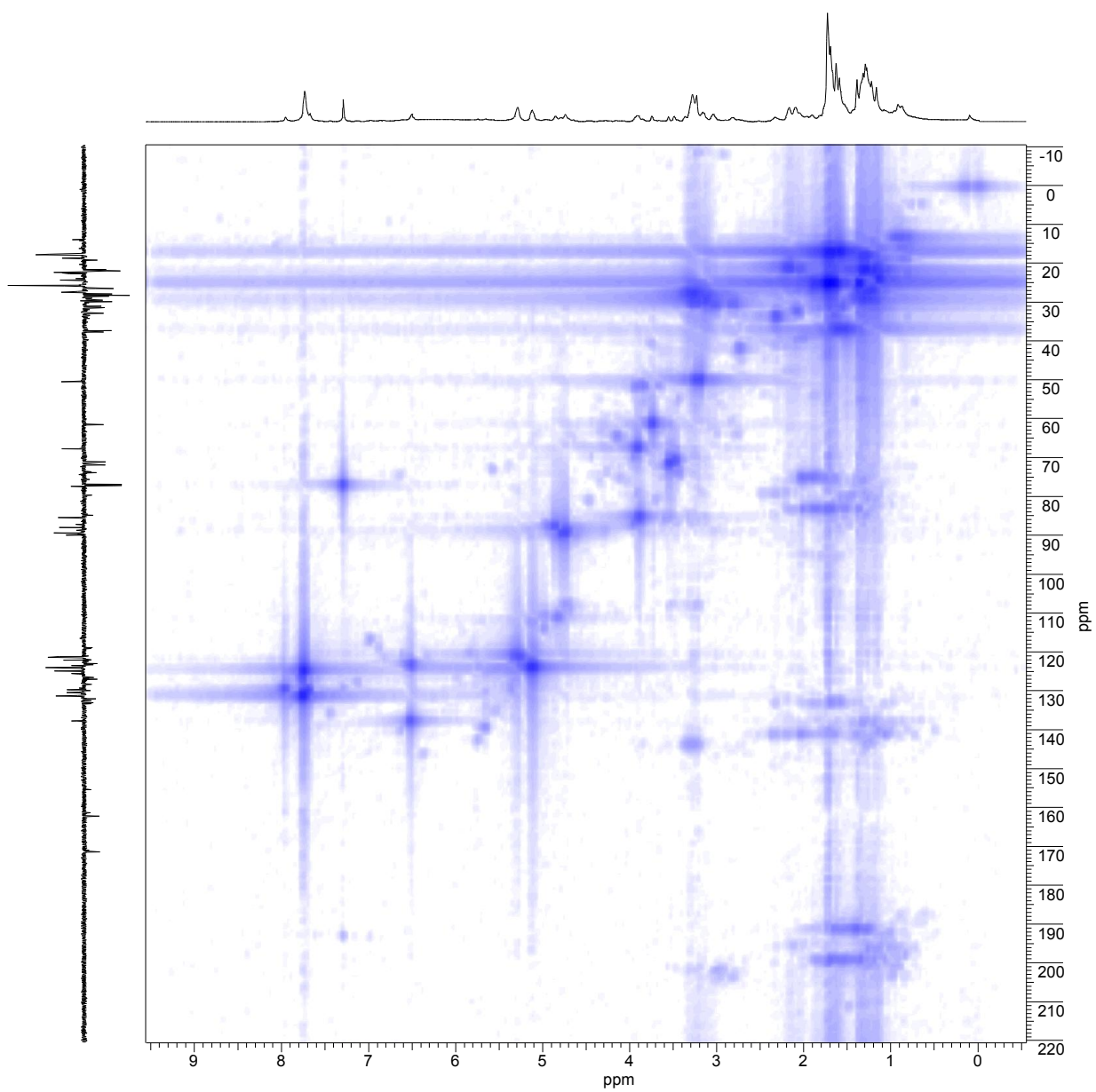
Appendix 27: ^1H NMR spectrum of compound 24



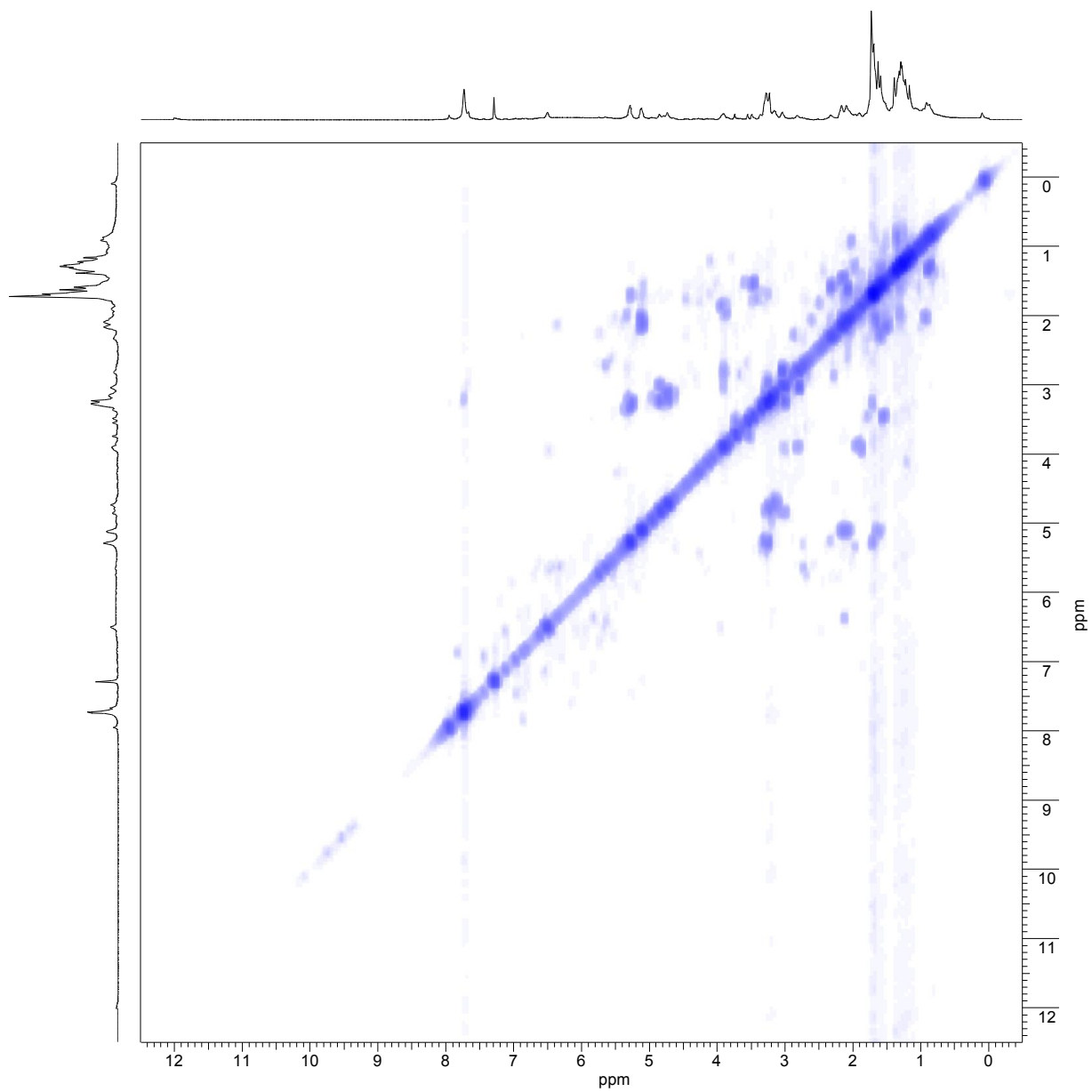
Appendix 28: APT NMR spectrum of compound 24



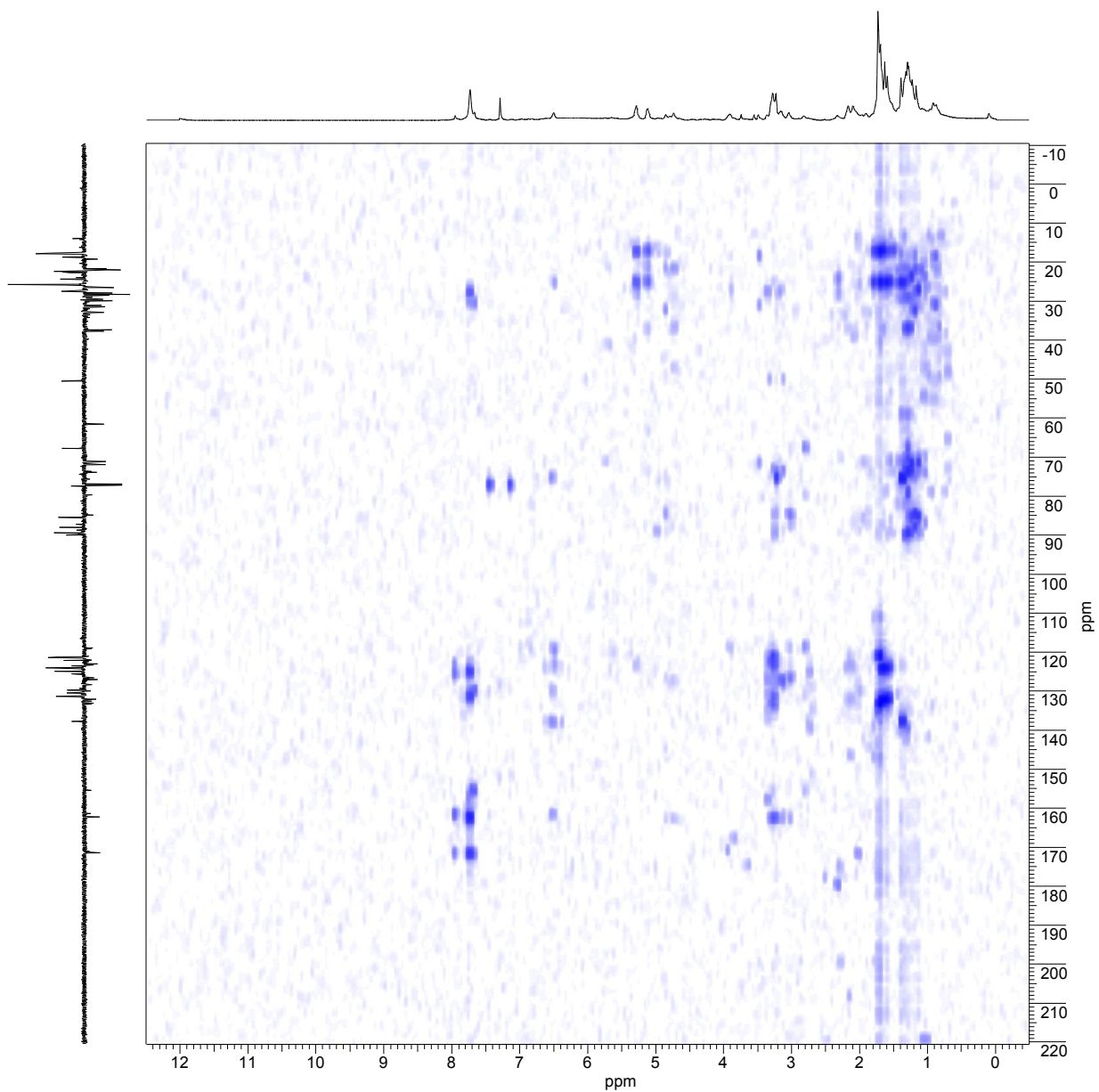
Appendix 29: HSQC spectrum of compound 24



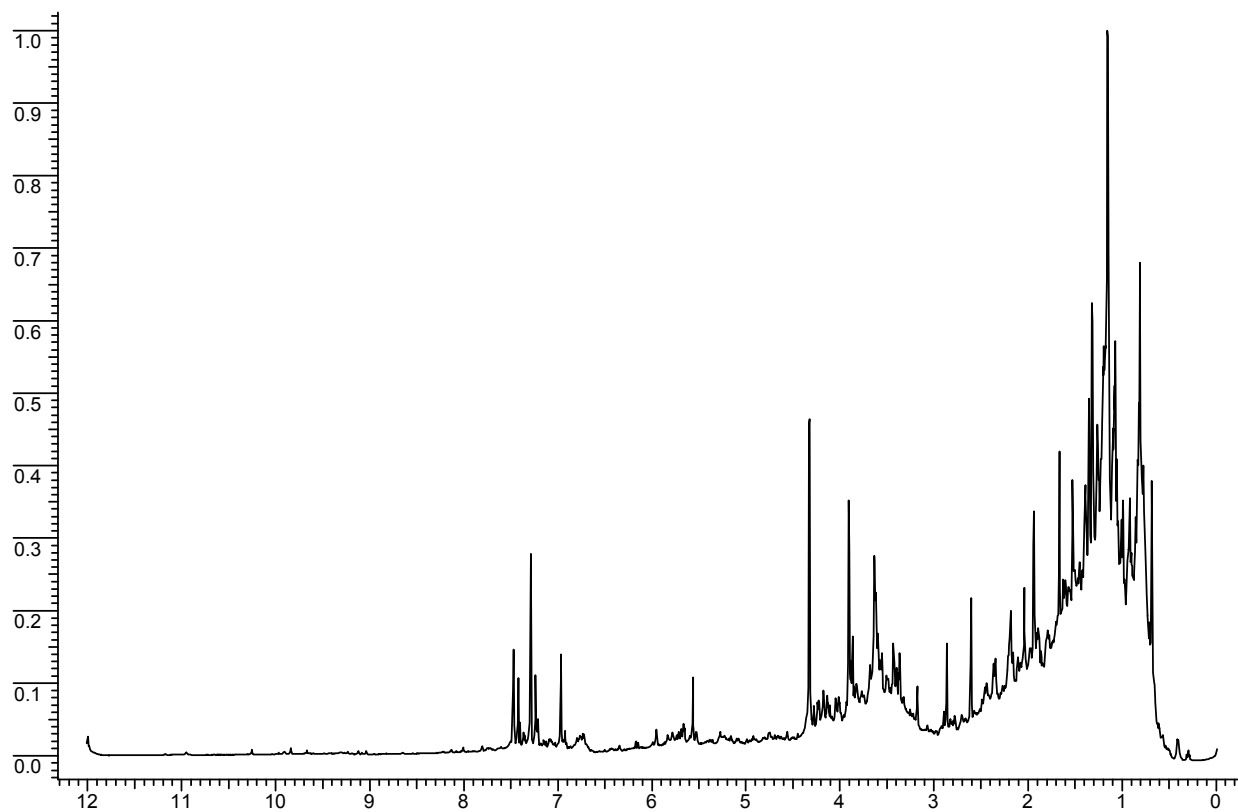
Appendix 30: ^1H - ^1H COSY NMR spectrum of compound **24**



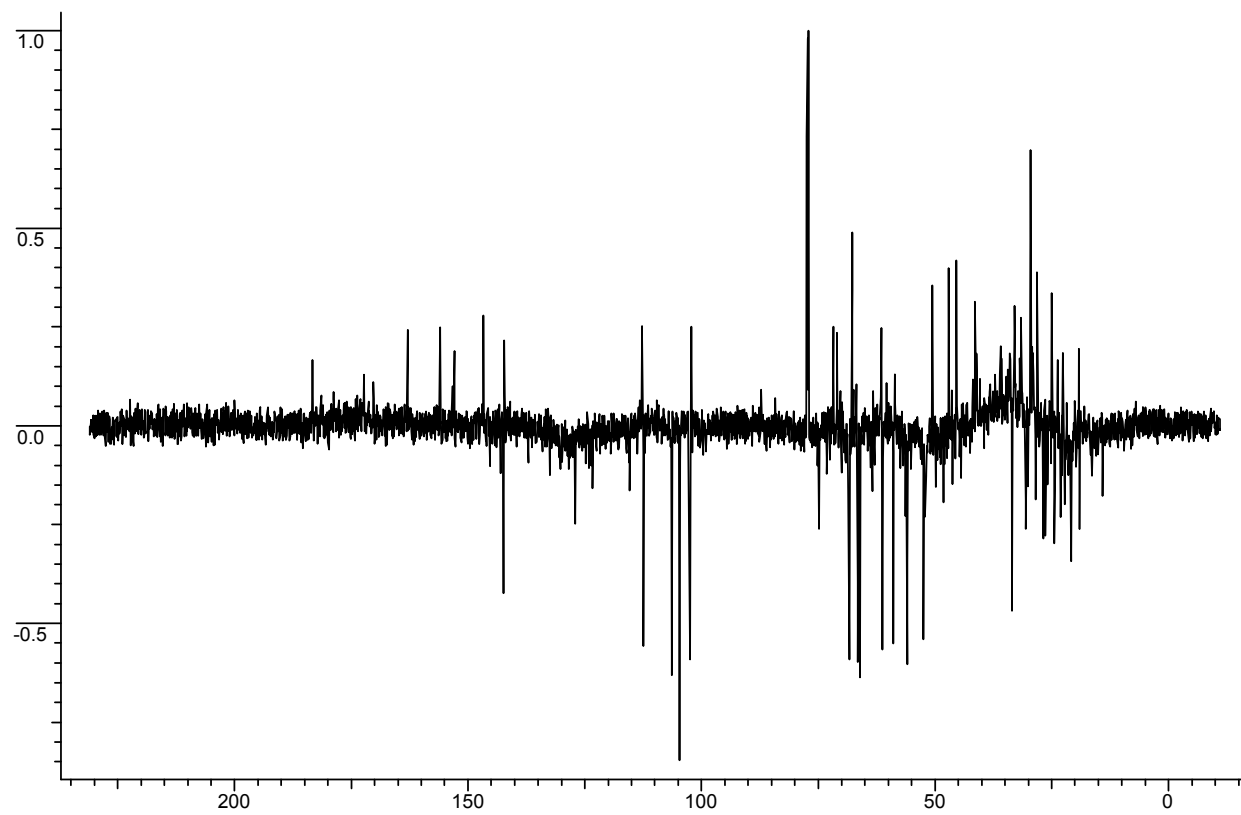
Appendix 31: HMBC NMR spectrum of compound **24**



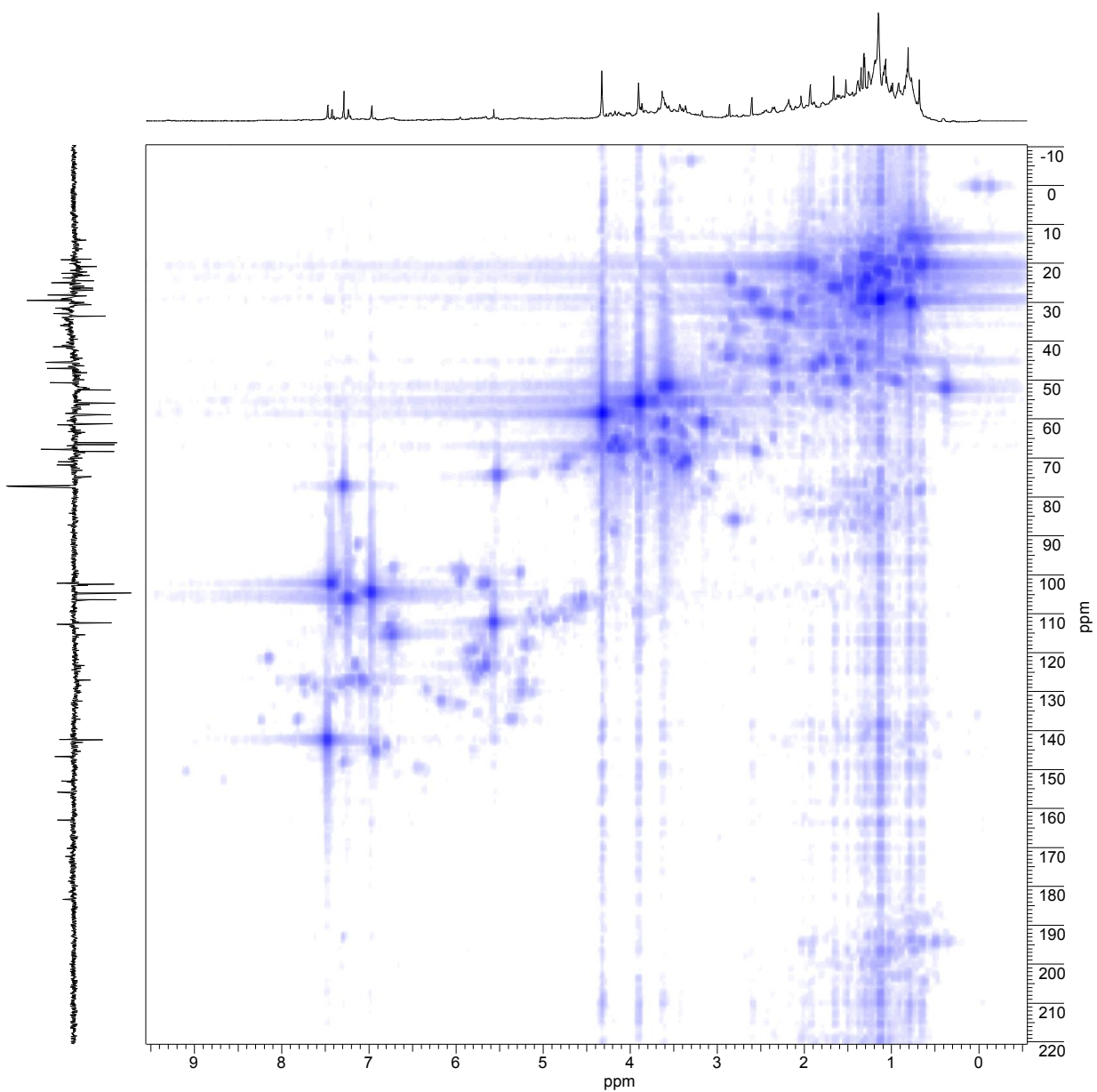
Appendix 32: ^1H NMR spectrum of compounds **10**, **13** and **21**



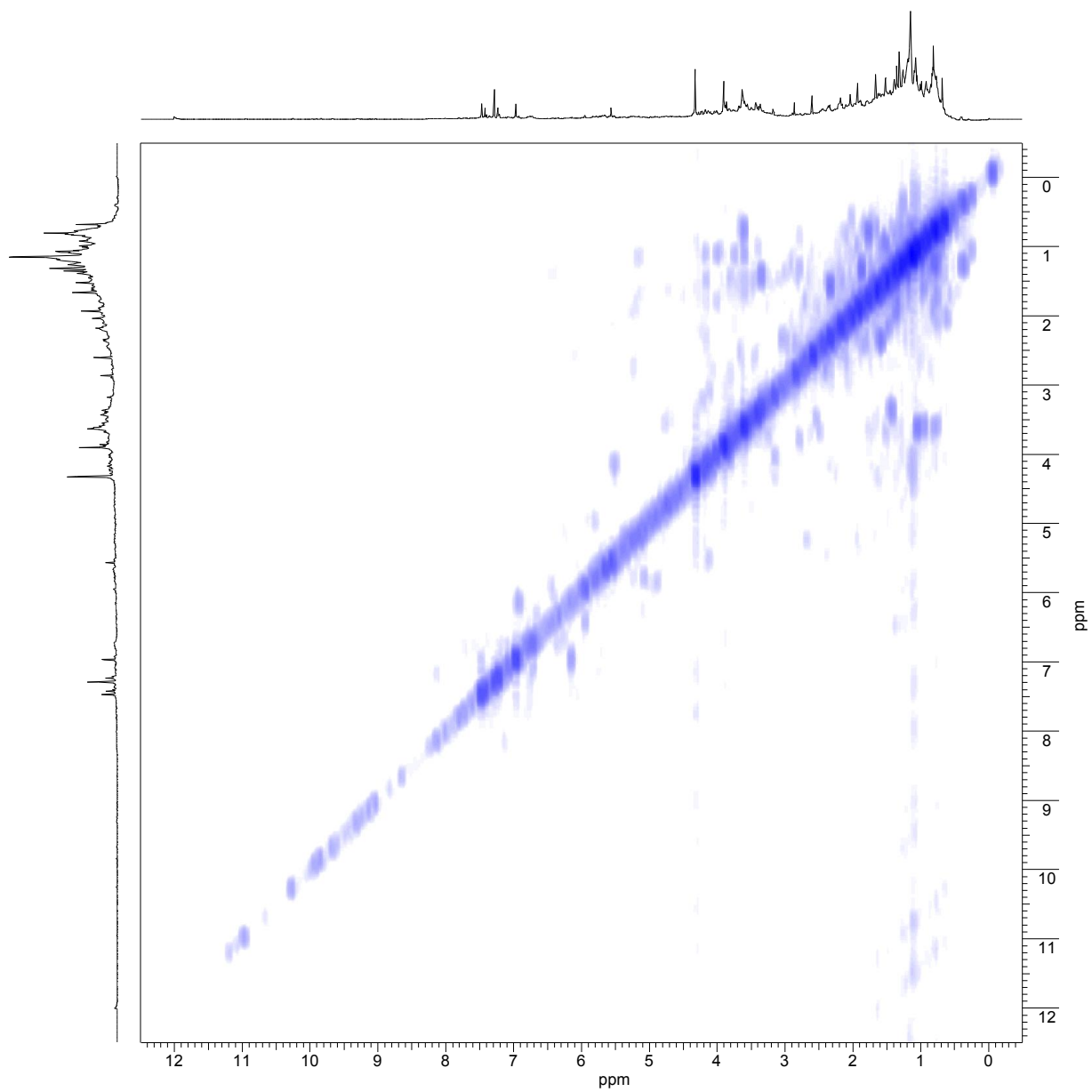
Appendix 33: APT NMR spectrum of compounds **10**, **13** and **21**



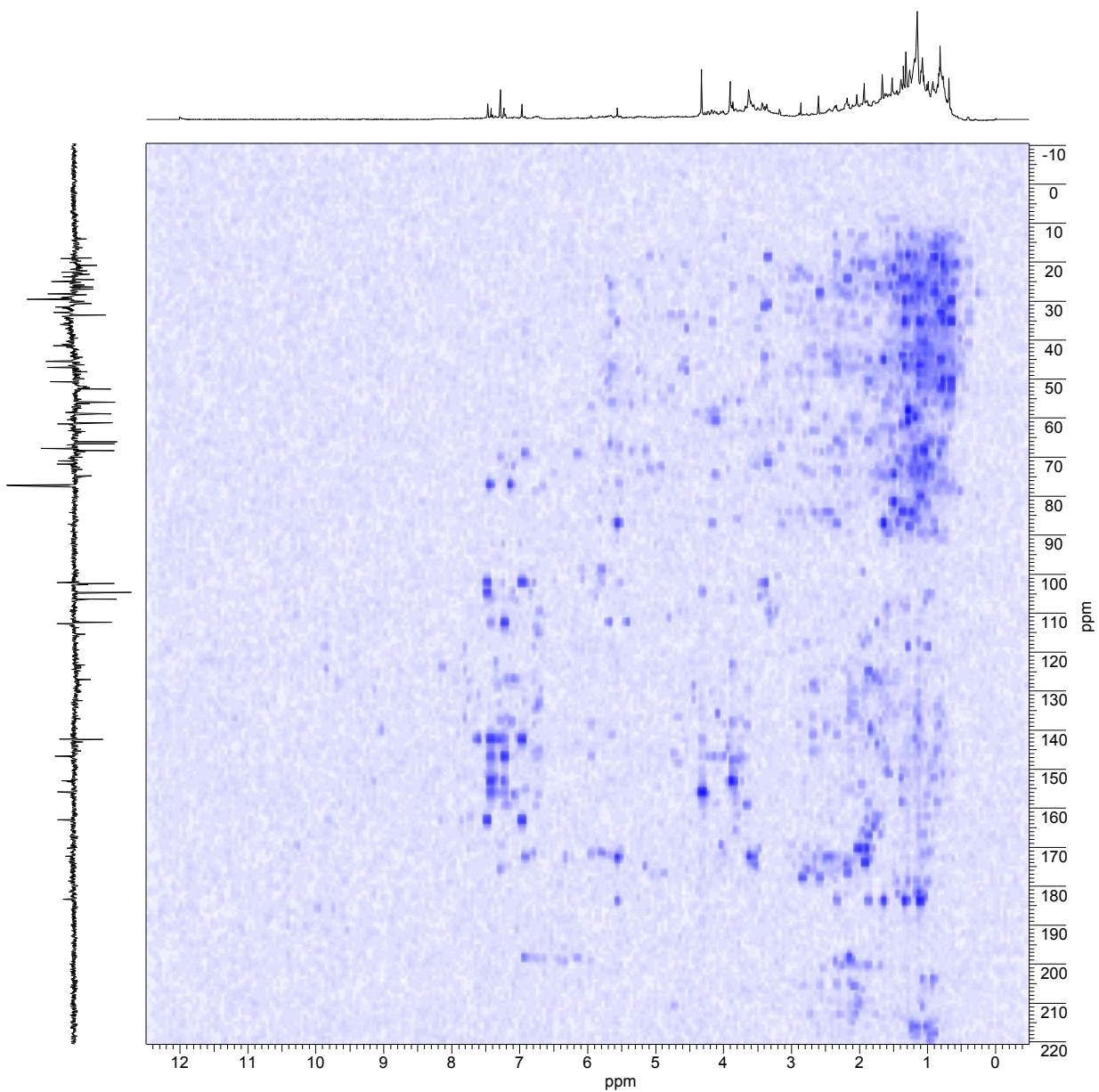
Appendix 34: HSQC NMR spectrum of compounds **10**, **13** and **21**



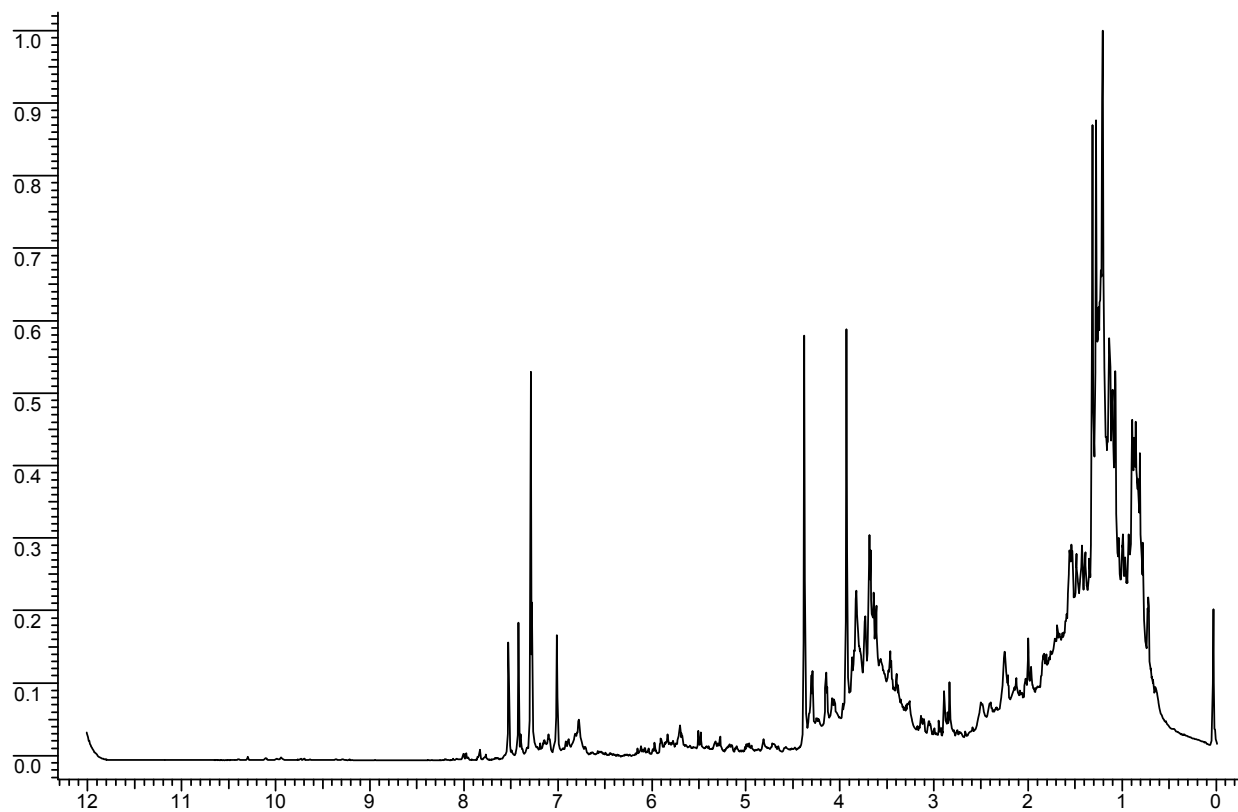
Appendix 35: ^1H - ^1H COSY NMR spectrum of compounds **10**, **13** and **21**



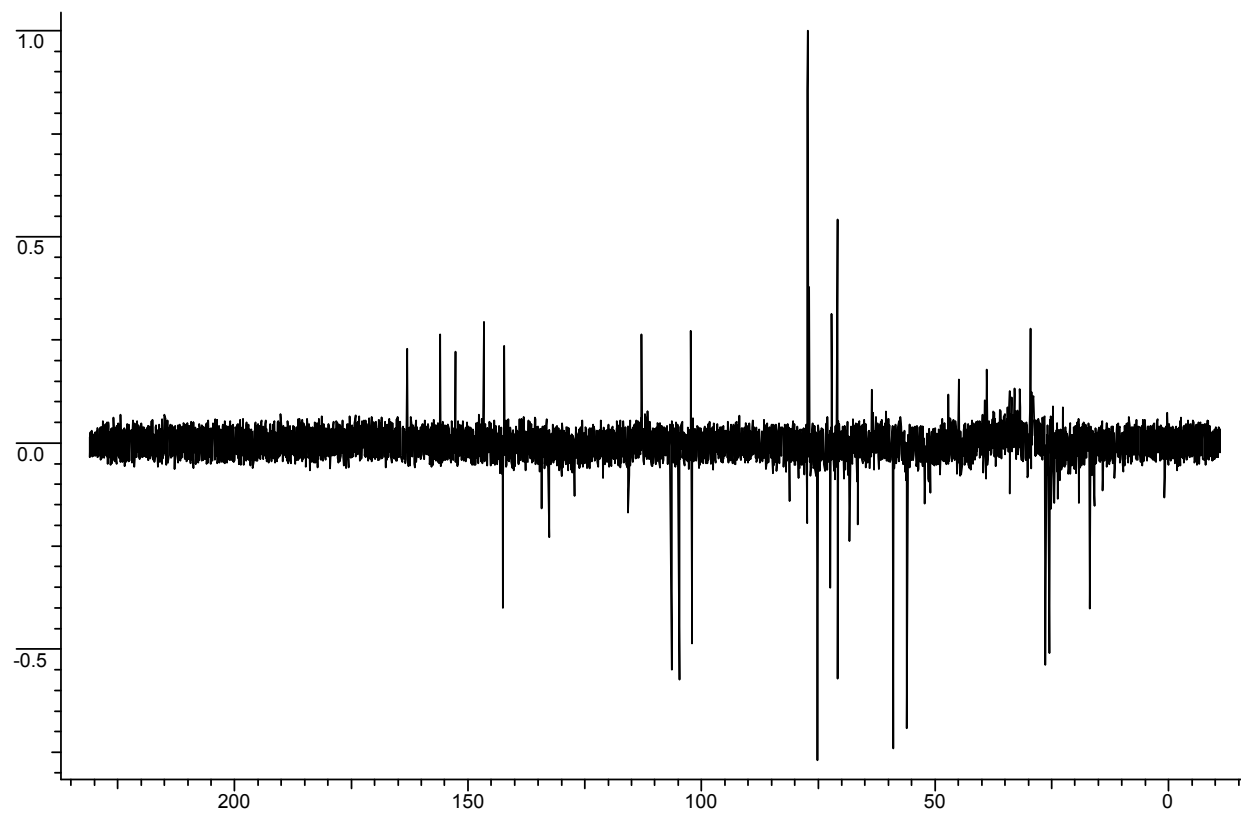
Appendix 36: HMBC NMR spectrum of compounds **10**, **13** and **21**



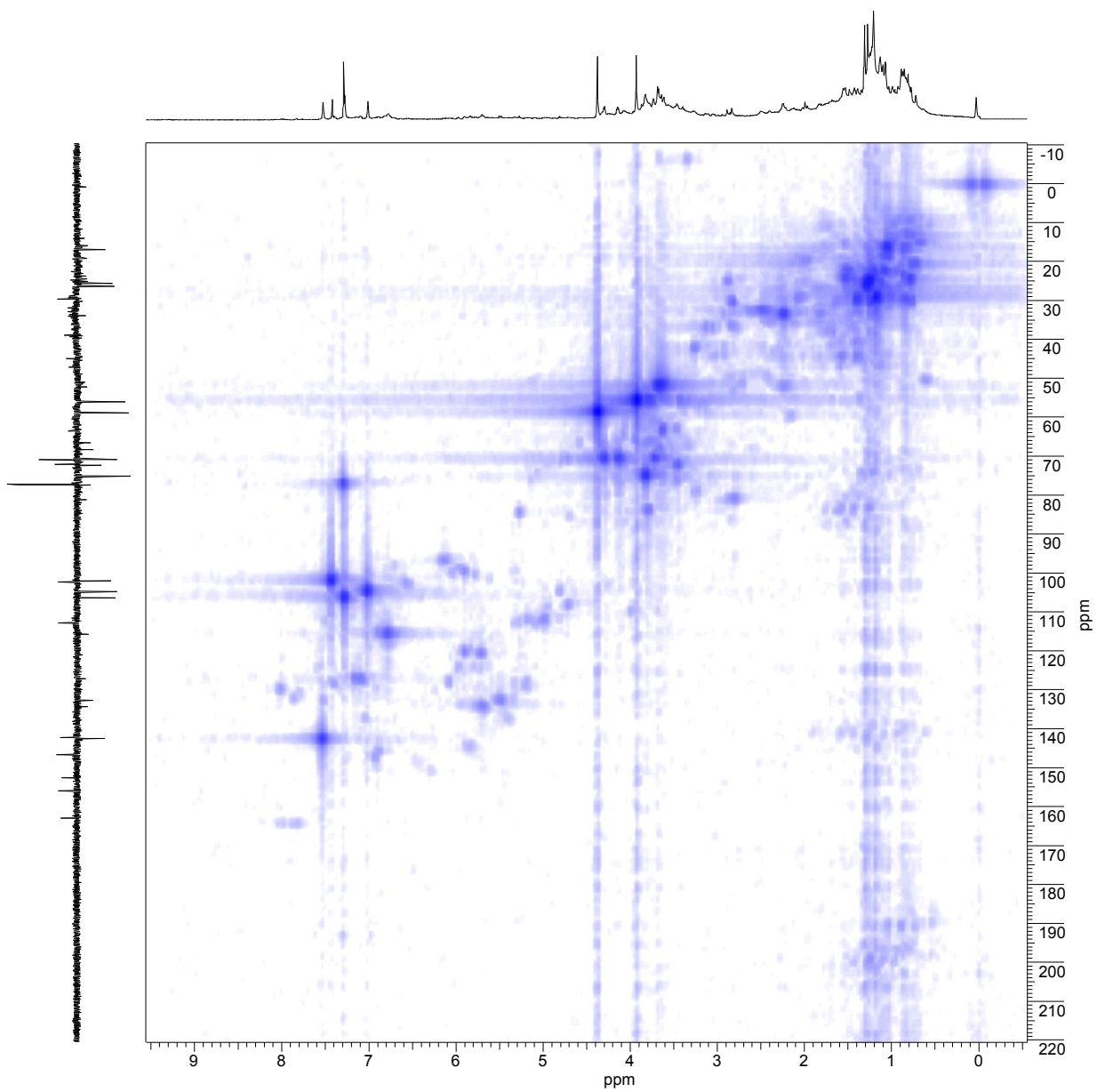
Appendix 37: ^1H NMR spectrum of compound **22**



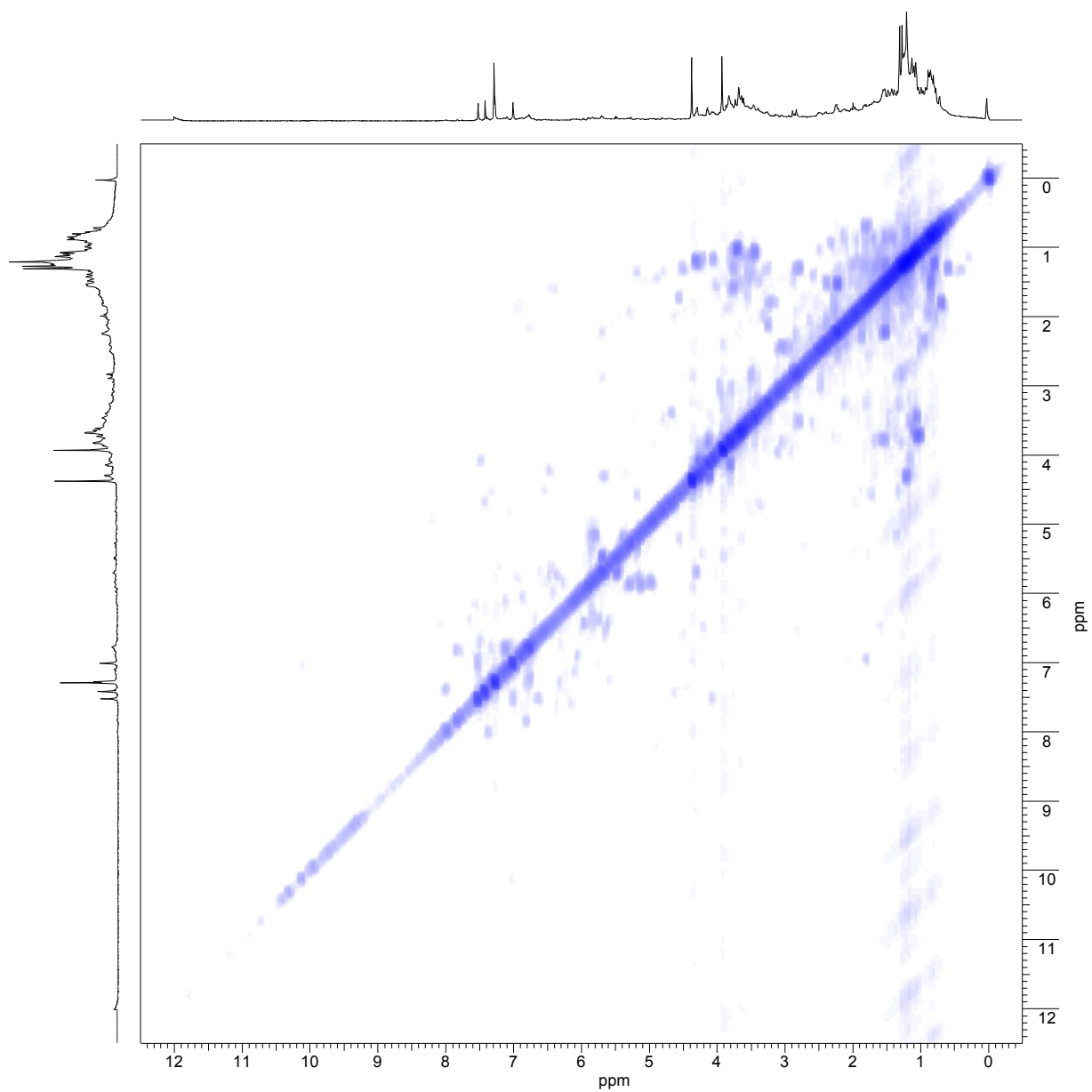
Appendix 38: APT NMR spectrum of compound **22**



Appendix 39: HSQC NMR spectrum of compound 22



Appendix 40: ^1H - ^1H COSY NMR spectrum of compound **22**



Appendix 41: HMBC NMR spectrum of compound 22

